

Module description

for the degree programme

Master of Science

Molecular Science

(Version of examination regulation: 20202)

for the summer term 2026

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1	Module name 1999	Masterarbeit (M.Sc. Molecular Science 20202) Master's thesis	30 ECTS
2	Courses / lectures	No courses / lectures available for this module for this semester! Please note: <ul style="list-style-type: none"> The master's thesis will be written at the student's choice normally in one of the research groups of the Department of Chemistry and Pharmacy or the Department of Biology! Students must independently apply for a master's thesis in one of the department's research groups! 	
3	Lecturers	No lecturers available since there are no courses / lectures for this module for this semester!	

4	Module coordinator	Prof. Dr. Petra Imhof
5	Contents	Written elaboration in form of a scientific manuscript with a length of approx. 20,000 words. It describes the scientific findings as well as the way leading to these findings. It contains justifications for decisions regarding chosen methods for the thesis and discarded alternatives. The student's own substantial contribution to the achieved results has to be evident. In addition, the student presents his work in a seminar, in which the scientific quality and the scientific independence of his achievements are evaluated. Workload: 900h
6	Learning objectives and skills	Students <ul style="list-style-type: none"> demonstrate their ability to perform independent scientific work focusing on an adequately challenging research topic. rank their own research results in the context of current literature and research papers in the field and record their results in appropriate scientific writing and documentation style. give oral and written presentations of the results and acquired knowledge in an appropriate scientific style in English language.
7	Prerequisites	Before commencing work on the Master's thesis, students must have successfully completed the two subject-related compulsory elective modules, the supplementary compulsory elective module and the research module, coming to a total of 75 ECTS credits.
8	Integration in curriculum	semester: 4
9	Module compatibility	Pflichtmodul Master of Science Molecular Science 20202
10	Method of examination	Written (6 Monate) <ul style="list-style-type: none"> Thesis (2 hard copies in bound form + electronic version) with a length of approx. 20,000 words; Referee report, 2 experts

11	Grading procedure	Written (100%) Averaged grade of the two reports
12	Module frequency	Every semester
13	Resit examinations	The exams of this moduls can only be resit once.
14	Workload in clock hours	Contact hours: 675 h Independent study: 225 h
15	Module duration	1 semester
16	Teaching and examination language	english
17	Bibliography	

Studienrichtung Drug Discovery/ Molecular Life Science

1	Module name 63065	Medicinal Chemistry Medicinal chemistry	10 ECTS
2	Courses / lectures	Vorlesung mit Übung: Medicinal Chemistry (3 SWS) (WiSe 2025)	5 ECTS
		Vorlesung mit Übung: Medicinal Chemistry (3 SWS) (SoSe 2026)	5 ECTS
3	Lecturers	Dr. Dorothee Weikert Dr. Dorothee Weikert Prof. Dr. Jürgen Schatz Prof. Dr. Peter Gmeiner Prof. Dr. Andriy Mokhir Prof. Dr. Nicolai Burzlaff Prof. Dr. Jutta Eichler Prof. Dr. Monika Pischetsrieder Prof. Dr. Markus Heinrich Dr. Jürgen Einsiedel Dr. Stefan Löber Prof. Dr. Susanne Mühlich	

4	Module coordinator	Prof. Dr. Peter Gmeiner
5	Contents	Genomics, transcriptomics, proteomics; in-vitro assay systems, assay technology; target screening and drug production in plants, drug screening and production in yeast; experimental structural biology; chemoinformatics; molecular modeling: molecular dynamics simulation, force-fields, modeling of proteins, proteinligand docking; drug synthesis and combinatorial chemistry; redox-active metal complexes, metalloenzyme inhibitors; stereochemistry in drug design; organic reactions in medicinal chemistry; drug metabolism; peptidomimetics; bioanalysis and instrumental analysis in drug discovery
6	Learning objectives and skills	The students are able <ul style="list-style-type: none"> to understand the basic and advanced principles of medicinal chemical, molecular biological and Computer chemistry based applications in the field of modern drug design research
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202 Studienrichtung Molecular Life Science Master of Science Molecular Science 20202
10	Method of examination	Written examination (90 minutes)
11	Grading procedure	Written examination (100%)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 90 h

		Independent study: 210 h
14	Module duration	2 semester
15	Teaching and examination language	english
16	Bibliography	G. Klebe, Drug Design: Methodology, Concepts, and Mode-of-Action, Springer 2013

1	Module name 63066	Biology-informed Drug Discovery Concepts in biology	10 ECTS
2	Courses / lectures	Vorlesung: Biology-informed drug discovery (2 SWS) (SoSe 2026)	5 ECTS
		Vorlesung: Biology-informed drug discovery (2 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Anja Lux Prof. Dr. Anja Lux	

4	Module coordinator	Prof. Dr. Yves Muller
5	Contents	<p>Basic biology concepts of importance for the drug discovery process. The lectures cover three main topics, namely (i) selected experimental methods, (ii) the use of model organisms in drug discovery and (iii) the discussion of the biology of specific classes of drugs, namely antibodies and natural products.</p> <p>Part (i) focusses on omics methods (genomics, transcriptomics, proteomics, metabolomics) in drug target identification and drug validation as well as on the biological readout of high-throughput screening techniques. In addition, the question how to gain atomic insight into macromolecular drug targets is addressed by covering the topics of molecular modelling, nuclear magnetic resonance spectroscopy, X-ray crystallography and cryo-electron microscopy.</p> <p>In part (ii) basic methods in molecular biology are reviewed and the following model organisms are being addressed: Yeast as a not yet fully explored model organism for the identification of novel drug targets. The production of recombinant proteins and the synthesis of small molecule drugs by yeasts and aspergillus species are discussed. Plants as even less well characterized organisms for drug discovery, target screening and drug production as well as the third model organism mice and how the latter are exploited as disease model systems. Topics are: xenograft tumour models in mice and genetic mouse models for inherited diseases and tumour diseases.</p> <p>In part (iii) the isolation, production and use of monoclonal antibodies are being discussed together with natural products. A particular focus of the latter are novel antibiotic drugs and their role in the fight against increasingly prevalent bacterial infections. In addition, several concepts to enhance the targeted delivery of natural products are evaluated, using model dispositions, such as coeliac disease.</p>
6	Learning objectives and skills	<p>Learning objectives are the acquaintance of novel insights, concepts, and methods in biology-informed drug discovery. To be able to understand state-of-the-art methods/concepts in drug discovery and to critically assess the potential and limitations of the individual methods/concepts. To be able to independently develop working hypotheses and to design experimental approaches to further the drug discovery process.</p> <p>As an integral part of the module, each student will have to present the content of a primary research article in a seminar presentation of approximately 20 minutes. Each student acquires the skill to</p>

		<p>understand, present, and critically discuss primary research reports in the field of drug discovery and to defend the findings before an audience.</p> <p>Attendance of the student presentations is mandatory. Only regular attendance of these student presentations will lead to a successful completion of the module. Justification: The learning objectives of this part can only be reached if the students actively discuss their literature findings with fellow students.</p>
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	<p>Studienrichtung Drug Discovery Master of Science Molecular Science 20202</p> <p>Studienrichtung Molecular Life Science Master of Science Molecular Science 20202</p>
10	Method of examination	<p>Written examination (90 minutes)</p> <p>PL: Written examination, 90 minutes</p>
11	Grading procedure	<p>Written examination (100%)</p> <p>(Please note: The grade of module "Biology-informed Drug Discovery", <i>former: Concepts in Biology</i>, is only a part of the total grade of module package "Drug Discovery"!)</p> <p>Grading procedure of "Drug Discovery": Medicinal Chemistry 25% + Biology-informed Drug Discovery (<i>former: Concepts in Biology</i>) 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%</p>
12	Module frequency	Every semester
13	Workload in clock hours	<p>Contact hours: 60 h</p> <p>Independent study: 240 h</p>
14	Module duration	2 semester
15	Teaching and examination language	english
16	Bibliography	Course material is provided via the StudOn platform

1	Module name 63067	Molecular Modelling Molecular modelling	10 ECTS
2	Courses / lectures	Vorlesung mit Übung: Modelling of Complex Systems (2V/1UE) (3 SWS) (WiSe 2025) Vorlesung mit Übung: Advanced Biomolecular Simulations (2V/1UE) (3 SWS) (SoSe 2026)	5 ECTS 5 ECTS
3	Lecturers	Prof. Dr. Dirk Zahn Prof. Dr. Petra Imhof Prof. Dr. Petra Imhof	

4	Module coordinator	Prof. Dr. Petra Imhof	
5	Contents	<p>WS:</p> <ul style="list-style-type: none"> Rationalizing Complex Systems from Statistics: Probability distributions, Concepts of Information and Entropy; Thermodynamics of Monte-Carlo simulation and state-of-the-art analyses of transition pathways; Molecular dynamics simulations of complex systems: data evaluation and reduction to key information. <p>SS:</p> <ul style="list-style-type: none"> Model building and setup of first-guess configurations; Enhanced sampling techniques to tackle manifolds of configurations and to find reaction paths; Machine-learning and prediction of real-world properties from molecular data. 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> master the basics of molecular modelling, are able to select and apply from a wide range of different simulation techniques the one that is relevant and suitable for different problems, can analyse, evaluate and present data and results, monitor and control their own progress. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202 Studienrichtung Molecular Life Science Master of Science Molecular Science 20202	
10	Method of examination	Written examination (90 minutes)	
11	Grading procedure	Written examination (100%) (Please note: The grade of module "Molecular Modelling" is only a part of the total grade of module package "Drug Discovery"!) Grading procedure of " Drug Discovery ": Medicinal Chemistry 25% + Biology-informed Drug Discovery 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%	
12	Module frequency	Every semester	
13	Workload in clock hours	Contact hours: 90 h	

		Independent study: 210 h
14	Module duration	2 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • B. Smit, D. Frenkel: Understanding Molecular Simulation: From Algorithms to Applications • A. Leach: Molecular Modelling: Principles and Applications

1	Module name 63068	Drug Discovery - Lab Drug discovery - Lab	10 ECTS
2	Courses / lectures	Praktikum: Drug Discovery - LAB (15 SWS) (SoSe 2026) or Praktikum: Drug Discovery - LAB (15 SWS) (WiSe 2026) Please note: Attendance in ab courses and safety instructions is compulsory!	10 ECTS 10 ECTS
3	Lecturers	Dr. Dorothee Weikert Dr. Hannah Zenker Prof. Dr. Nicolai Burzlaff Prof. Dr. Jutta Eichler Prof. Dr. Monika Pischetsrieder Prof. Dr. Olaf Prante Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Monika Pischetsrieder	
5	Contents	3 independent practical courses focusing on biomolecular simulations (block, 5 SWS), Pharmacopoeia-based drug analysis (block, 5 SWS) and instrumental and bioanalytical techniques (block, 5 SWS)	
6	Learning objectives and skills	<p>The students are able</p> <ul style="list-style-type: none"> to understand the basic and advanced principles of Pharmacopoeia-based drug analysis, instrumental and bioanalytics as well as computer chemistry-based applications in the field of modern drug research to utilize modern instrumental analytical techniques for characterisation of samples and quantification of target analytes to apply modern simulation techniques for the modeling of biomolecules to interpret and to critically summarize experimental results in written form (lab report) 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202 Studienrichtung Molecular Life Science Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol of 30 - 50 pages (plus raw data information)	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	Every semester	
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h	
14	Module duration	2 semester	

15	Teaching and examination language	english
16	Bibliography	Manuscripts are available on StudON

Studienrichtung Molecular Nanoscience

1	Module name 63071	Molecular Nanoscience I Molecular nanoscience I	15 ECTS
2	Courses / lectures	Seminar: Nanoprobes (2 SWS) (WiSe 2026) Seminar: Molecular Nano Science (winter semester) (2 SWS) (WiSe 2026) Vorlesung mit Übung: Self-Assembly: Molecular, Particulate and Hybrid Nanostructures (3 SWS) (WiSe 2026) Seminar: From 2D assemblies to bulk (2 SWS) (WiSe 2026) Seminar: Supramolecular Chemistry and Molecular Materials (2 SWS) (WiSe 2026)	
3	Lecturers	Prof. Dr. Rainer Fink Prof. Dr. Andreas Hirsch	

4	Module coordinator	Prof. Dr. Rainer Fink Prof. Dr. Andreas Hirsch
5	Contents	<ul style="list-style-type: none"> • Concepts in supramolecular chemistry; host-guest chemistry; energetics of supramolecular complexes: experimental methods; templates and self-assembly. Molecular devices. Supramolecular catalysis: principles of supramolecular catalysis, supramolecular metal catalysis, self-assembled catalysts, metal-free catalysis, enzyme mimics, antibodies, imprinted polymers. • Nanoscaled systems, general issues of microscopic techniques; experimental techniques with nanometer resolution: resolution determination; image processing techniques, light microscopic techniques with light in the visible and IR range, confocal techniques, scanning probes (STM, AFM and related scanning probes). • Specific topics in synthesis and analysis of specific molecule-based nanoscale objects • 2D film fabrication techniques (PVD, ALD, etc.)
6	Learning objectives and skills	<p>The students are capable ...</p> <ul style="list-style-type: none"> • to explain the fundamental chemical and physical properties of nano-scale materials • to distinguish and to compare some properties, structure and applications of different nanomaterials • to describe and to evaluate the major concepts in supramolecular chemistry, molecular self-assembly and nano-scaled 2D materials and transfer knowledge to related topics • to judge on potential nano-analytical tools with optimized image contrast
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2

9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202
10	Method of examination	Oral O30 (PL): Oral Examination, 30 Minutes
11	Grading procedure	Oral (100%) (Please note: The grade of module "Molecular Nanoscience I" is only a part of the total grade of module package "Molecular Nanoscience"!)
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 150 h Independent study: 300 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 63072	Molecular Nanoscience II Molecular nanoscience II	15 ECTS
2	Courses / lectures	<p>Vorlesung: Carbon allotropes: from characterization to applications (2 SWS)</p> <p>Seminar: Nanoprobes II (2 SWS)</p> <p>Vorlesung: Modelling & Simulations --- Start: 20.4. (2 SWS)</p> <p>Seminar: Molecular Nano Science (2 SWS)</p> <p>Vorlesung: Inorganic Nanoparticles & Supraparticles (2 SWS)</p> <p>Vorlesung mit Übung: Molecular Switches and Molecular Machines (3 SWS)</p>	
3	Lecturers	<p>Prof. Dr. Dirk Michael Guldi</p> <p>Prof. Dr. Rainer Fink</p> <p>Prof. Dr. Siow Woon Ng</p> <p>Prof. Dr. Dirk Zahn</p> <p>Prof. Dr. Bernd Meyer</p> <p>Prof. Dr. Andreas Hirsch</p> <p>Prof. Dr. Franziska Gröhn</p> <p>Prof. Dr. Karl Mandel</p> <p>Prof. Dr. Henry Dube</p>	

4	Module coordinator	Prof. Dr. Andreas Hirsch	
5	Contents	<ul style="list-style-type: none"> • Concepts in supramolecular chemistry; molecular switches, molecular machines; • Nanoanalytical tools with charged particle: ion microscopy (FIM, HeIM, Nano-SIMS), electron microscopy (PEEM, LEEM, SEM, (S)TEM), aberration correction, x-ray tools, hybrid type microscopies; coherent imaging; 3D imaging (tomography, laminography) • Carbon allotropes: from molecules to carbon nanodots, fullerenes, carbon nanotubes and nanohorns; synthesis, properties and applications. • Inorganic nanoparticles and industrial applications, synthesis methods, sol-gel science and technology, concept of inorganic supraparticles & practical assembly techniques; characterization methods for supraparticulate systems; applications of supraparticles. • Modelling of molecular materials, molecular assembly, molecular dynamics • Specific topics in molecular nanoscience 	
6	Learning objectives and skills	<p>The students are capable ...</p> <ul style="list-style-type: none"> • to explain the fundamental chemical and physical properties of nano-scale materials • to distinguish and to compare some properties, structure and applications of different nanomaterials 	

		<ul style="list-style-type: none"> to describe and to evaluate the major concepts in supramolecular chemistry, molecular machines, self-assembly and supraparticules, or carbon allotropes and transfer this knowledge to related topics to judge on potential nano-analytical tools with optimized image contrast
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202
10	Method of examination	Oral O30 (PL): Oral examination, 30 Minutes
11	Grading procedure	Oral (100%) (Please note: The grade of module "Molecular Nanoscience II" is only a part of the total grade of module package "Molecular Nanoscience"!)
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 180 h Independent study: 270 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> E. Katz: Molecular and Supramolecular Information Processing", Wiley-VCH D. Attwood: Soft x-rays and extreme ultraviolet radiation" (Cambridge); C. Jacobsen: X-ray microscopy" (Cambridge) L. Reimer: Scanning electron Microscopy (Springer, 2020) <p>and others</p>

1	Module name 63073	Molecular Nanoscience - Lab Molecular nanoscience - Lab	10 ECTS
2	Courses / lectures	Praktikum: Molecular Nanoscience - Lab (7 SWS) <ul style="list-style-type: none"> • Attendance in lab course is compulsory! • Attendance at safety instruction is compulsory! • Attendance in winter or summer term possible! • A valid laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	10 ECTS
3	Lecturers	Prof. Dr. Andreas Hirsch Prof. Dr. Julien Bachmann Prof. Dr. Rainer Fink	

4	Module coordinator	Prof. Dr. Rainer Fink
5	Contents	Research practicals in selected modern fields of molecular nanoscience: Topics according to an annually updated list of experiments, which includes molecule or nanoparticle synthesis, spectroscopic and microscopic analysis, thin-film and device preparation, characterization
6	Learning objectives and skills	The students are capable ... <ul style="list-style-type: none"> • to use their theoretical and practical background to plan and perform advanced research experiments under supervision of experienced scientists • to synthesize nanoscaled materials • to interpret spectroscopic or microscopic data taking advantage of modern research tools • to provide a state-of-the-art documentation and discussion of their experimental results • to present, communicate and discuss scientific results in scientific english.
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202
10	Method of examination	Practical achievement
11	Grading procedure	Practical achievement (100%)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	Lecture notes from the various course lectures in Molecular Nanoscience" (see module descriptions "Molecular Nanoscience-I" and "Molecular Nanoscience-II")

Advances in bio-organic and bio-inorganic chemistry

Mandatory elective module (20 ECTS in total):

Please choose **3 lecture modules out of 4:**

- *Advanced Bio-Organic and Bio-Inorganic Chemistry*
- *Special Aspects in Bio-Organic Chemistry*
- *Sensory Sciences Lab*
- *Medical Life Sciences*

Please note:

- The lab module "*Bio-Organic and Bio-Inorganic Chemistry - Lab*" is **mandatory!**

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Advances in interface research and catalysis B*
- *Advances in Organic Chemistry*

1	Module name 46511	Advanced Bio-Organic and Bio-Inorganic Chemistry Advanced bio-organic and bio-inorganic chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (2 SWS) (WiSe 2026) Seminar: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine - Seminar (1 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Nicolai Burzlaff	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	<p>The students</p> <ul style="list-style-type: none"> • are introduced into recent activities and achievements in the fields • of bioorganic and bioinorganic chemistry and metals in medicine: • porphyrins and heme enzymes • PDT • electron transfer (cofactors, ferredoxins, inner sphere mechanism, outer sphere mechanism, Marcus theory) • photosynthesis • copper containing proteins and enzymes • nitrogenases and other Mo containing enzymes • drugs based on Pt, Ru, Au and As • biominerals • implant materials and technology 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	Contact hours: 45 h	

		Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> W. Kaim, B. Schwederski, A. Klein, Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life, 2. Edition, John Wiley & Sons, Ltd, 2013

1	Module name 46512	Special Aspects in Bio-Organic Chemistry Special aspects in bio-organic chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Special Aspects in Bio-Organic Chemistry (2 SWS) (WiSe 2026) Übung: Special Aspects in Bio-Organic Chemistry: Seminar (1 SWS) (WiSe 2026) Please note: <ul style="list-style-type: none"> • Seminar: time and place by agreement! 	5 ECTS
3	Lecturers	Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff
5	Contents	The students <ul style="list-style-type: none"> • learn about: nucleic acid structure, synthesis, and reactivity; nucleic acid modifications; nucleic acid-protein interactions, nucleic acid-based drugs, nucleic acid biochemistry, modelling of nucleic acids
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • can explain, apply and reflect upon the theories, terminology, specialities, boundaries and different school of bioorganic and bioinorganic chemistry critically and in depth
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • A. Bloomfield, D.M. Crothers, I. Tinoco, Jr. „Nucleic Acids: structures, properties, and functions“, University Science Books, Sausalito, CA, USA 1999

- D. L. van Vranken, G. A. Weiss, Introduction to Bioorganic Chemistry and Chemical Biology, 1. Aufl., Garland Publishers 2012, ISBN: 978-0815342144
- W. Saenger, „Principles of Nucleic Acid Structure“, Springer-Verlag New York Inc. 1984
- T. Schlick: "Molecular modelling and simulation: An interdisciplinary guide", Springer New York Dordrecht Heidelberg London 2nd ed. 2010
- A. Vologodskii: "Biophysics of DNA", Cambridge University press, 2015, DOI 10.1017/CBO9781139542371

1	Module name 46514	Bio-Organic and Bio-Inorganic Chemistry - Lab Bio-organic and bio-inorganic chemistry - Lab	5 ECTS
2	Courses / lectures	Praktikum: Bio-Organic & Bio-Inorganic LAB (7 SWS) Please note: Attendance at lab course is compulsory!	5 ECTS
3	Lecturers	Prof. Dr. Andriy Mokhir Prof. Dr. Nicolai Burzlaff apl. Prof. Dr. Norbert Jux Prof. Dr. Petra Imhof Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	<p>The students</p> <ul style="list-style-type: none"> • deepen their knowledge in special topics of bioorganic and bioinorganic chemistry and nanomedicine that are in the research focus of the involved research groups of the department depending on their own choice • perform practical studies and small research projects regarding topics of the preparative, mechanistic or more biological bioorganic and bioinorganic chemistry and nanomedicine in an advanced level 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • can characterise and evaluate bioinorganic models • manage the preparation of bioorganic compounds and bioinorganic models as well as synthesis of functionalized nanoparticles, their characterization as well as their application in mechanistic studies • carry out bioorganic and bioinorganic research projects largely independently using a wide range of bioorganic and bioinorganic theories and are able to reflect upon the gained results 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lab module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total!) 	
10	Method of examination	<p>Practical achievement Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)</p>	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	Every semester	
13	Workload in clock hours	<p>Contact hours: 105 h Independent study: 45 h</p>	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	

1	Module name 46515	Sensory Sciences Lab	5 ECTS
2	Courses / lectures	<p>Projektseminar: Sensory Sciences Lab (for MSc Chemistry und MSc Molecular Science) (2 SWS) (WiSe 2026)</p> <p><u>Please note:</u></p> <ul style="list-style-type: none"> The module "Sensory Sciences Lab" is an interdisciplinary module with a limited number of participants: only 10 students from MSc Chemistry or MSc Molecular Science will be admitted! Students from Molecular LIFE Science will be given priority! The module "Sensory Sciences Lab" can be taken as part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" (20 ECTS in total, all modules have to be graded) or as an elective module (5 ECTS, ungraded)! 	5 ECTS
3	Lecturers	PD Dr. Helene Loos	

4	Module coordinator	Prof. Dr. Andrea Büttner
5	Contents	<p>The field of Sensory Sciences investigates how animals and humans sense, neurally process and perceive their environment. This research-oriented interdisciplinary course provides important concepts, theories, and methods of the Sensory Sciences to advanced students from three disciplines (Psychology, Medicine and Chemistry - Molecular Science). These topics comprise human auditory, visual and olfactory perception, chemocommunication, psychophysics, neuroimaging, molecular sensory receptors and technologies for neuromodulation (among others). The students apply these concepts in interdisciplinary teams to develop, conduct and analyse their own small research project. While engaging in an interdisciplinary collaboration and exchange, students can share and deepen their discipline-specific perspectives and competencies as well as learn about concepts and methods from other disciplines.</p>
6	Learning objectives and skills	<p>The students are able to</p> <ul style="list-style-type: none"> explain fundamental concepts and methods of the Sensory Sciences develop an own (limited) research question and design an empirical data collection to investigate the research question collect data in the laboratory and analyse the data present their findings in oral and written form communicate effectively in interdisciplinary teams understand and reflect the challenges and chances of interdisciplinary collaborations.
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3

9	Module compatibility	<p>Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> Part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" in M.Sc. Molecular Science (5 ECTS, graded; PO 2020) elective module (5 ECTS, ungraded; PO 2020)
10	Method of examination	<p>Seminar achievement</p> <p>Seminar achievement: Written project report (maximum 30 pages), oral project presentation, experimental work in the lab</p>
11	Grading procedure	<p>Seminar achievement (100%)</p> <p>Seminar achievement: 50% project reports, 25% project presentation, 25% lab work</p>
12	Module frequency	Only in winter semester
13	Workload in clock hours	<p>Contact hours: 30 h</p> <p>Independent study: 120 h</p>
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> Literature will be announced in the first session Scripts and materials will be provided via StudOn!

1	Module name 46516	Medical Life Sciences	5 ECTS
2	Courses / lectures	Vorlesung: Medical Life Sciences	5 ECTS
3	Lecturers	Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Ingrid Span
5	Contents	To lay the foundation for working in molecular disease research, the MedLife module provides knowledge in genetics, molecular biology, cell biology, nucleic acid and protein chemistry, enzymology, nuclear and nanomedicine.
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • explain the basic principles and techniques of the various fields of research • engage with scientific texts • will be able to apply the essential principles to practical biological and chemical problems
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total)* • Lecture module as Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) *If students want to take the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" as a whole, they have to choose 3 out of the 4 lecture modules, the laboratory module "Bio-Organic and Bio-Inorganic Chemistry - Lab" (No.46514) must be taken compulsorily!
10	Method of examination	Written examination (60 minutes)
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

Advances in homogeneous catalysis

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Quantum Chemistry*
- *Advances in Organic Chemistry*

1	Module name 46521	Organometallic Catalysis Organometallic catalysis	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organometallic Catalysis - Sustainable and Enantioselective Catalysis (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Sjoerd Harder Prof. Dr. Romano Dorta	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	<p>Knowledge on homogeneous organometallic catalysis will be deepened by discussing the latest scientific breakthroughs and developments in modern areas of catalysis. The module consists of two parts.</p> <p>Harder-Part <i>Sustainable Main Group Metal Catalysis</i></p> <ul style="list-style-type: none"> • Activation of strong bonds and small molecules with main group metal complexes (who needs d-orbitals?). Replacing expensive rare transition metals with cheap and abundant main group metals. • Bond activation using the ball-mill method. • Catalysis with main group metal complexes (discussion of elementary steps and catalytic cycles). • Activation of metals and making metal catalysts using Metal-Vapor-Synthesis. The evaporation of metals under vacuum and co-condensation in an organic matrix is a great way to make highly active metal catalysts. <p>Dorta-Part <i>Asymmetric Catalysis by Chiral Metal Complexes – Applications in Industry</i></p> <ul style="list-style-type: none"> • Importance of chiral molecules in nature and life science • Chiral pharmaceutical drugs and agrochemicals • Asymmetric catalysis: Chiral catalysts acting as a chirality multiplier • Advantages in comparison with racemate separation or stoichiometric enantioselective syntheses. • Historical development of chiral technology and industrial applications. • Synthesis of enantiopure ligands (famous industrial examples will be discussed). • Mechanistic aspects (catalytic cycles). 	
6	Learning objectives and skills	<p>The student</p> <ul style="list-style-type: none"> • will learn the latest developments at the forefront of organometallic catalysis • will understand that the field of catalysis needs to become sustainable and that we move away from expensive rare metals to abundant main group metals • will be able to critically discuss catalytic cycles and understand reaction mechanisms 	

		<ul style="list-style-type: none"> • will be taught new methods (practical and theoretical) to study molecular properties • will be understand the need for new methods of catalyst preparation • will be up-to-date with asymmetric catalysis and industrial applications
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	<p>Advances in Homogenous Catalysis Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective module "Advances in Homogenous Catalysis" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, ungraded)
10	Method of examination	<p>Oral</p> <p>O20(PL): Oral examination (20 minutes)</p>
11	Grading procedure	<p>Oral (100%)</p> <ul style="list-style-type: none"> • As part of the Compulsory Elective module "Advances in Homogenous Catalysis": O20, 100% • As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 • Early Main Group Metal Catalysis – Concepts and Reactions, Sjoerd Harder, Wiley-VCH, 2020 • Metal-catalysis in Industrial Organic Processes, Gian Paolo Chiusoli & Peter Maitlis, Royal Society of Chemistry, 2019 • Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 • Homogeneous Catalysis, Parshall & Ittel, 2nd ed., Wiley Interscience 1992

1	Module name 46522	Organocatalysis and Catalytic Reactions in Water Organocatalysis and catalytic reactions in water	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organocatalysis and Catalytic Reactions in Water (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Andriy Mokhir Prof. Dr. Svetlana Tsogoeva	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	<ul style="list-style-type: none"> History and basic principles of organocatalysis, Different types of organocatalysts, Enantioselective organocatalysis, Domino reactions and important examples. Catalytic reactions in water: Important reactions in biological chemistry, Metabolism of biomolecules and artificial bioorthogonal reactions. 	
6	Learning objectives and skills	<p>The student</p> <ul style="list-style-type: none"> can explain basic principles and underlying reaction mechanisms in organo- and organometallic catalysis and biological chemistry deepens her/his knowledge in special topics of homogeneous catalysis that are in the research focus of the involved research groups is able to construct important reaction mechanisms and catalytic cycles and can critically discuss each step. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Homogenous Catalysis Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <p>Organocatalysis and Catalytic Reactions in Water can be taken</p> <ul style="list-style-type: none"> within the Compulsory Elective Module "Advances in Homogenous Catalysis" (20 ECTS in total) as Elective Module (5 ECTS, not graded) 	
10	Method of examination	<p>Oral</p> <p>O20(PL): Oral examination (20 minutes)</p>	
11	Grading procedure	<p>Oral (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46523	Small Molecule Activation Small molecule activation	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Small Molecule Activation (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Karsten Meyer Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	Lecture: <ul style="list-style-type: none"> Inorganic coordination chemistry, catalytic and electrocatalytic transformation of abundant small molecules N₂, O₂, CO or CO₂, H₂O and NH₃, into value-added commodities, fine-chemicals, active pharmaceutical ingredients and polymers. Selective oxidation reactions. CO₂ reduction to CO and subsequent Fischer-Tropsch catalysis for the production of carbon-based fuels. Reductive activation of atmospheric N₂ to NH₃ and the production of H₂ from H₂O splitting catalysis. Applications, e.g. fuel-cell technologies for carbon-free energy production. 	
6	Learning objectives and skills	The student <ul style="list-style-type: none"> can explain basic principles and underlying reaction mechanisms in small molecule activation is able to construct important reaction mechanisms and catalytic cycles can critically discuss small-molecule activation 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> Lecture module within the Compulsory Elective Module (20 ECTS in total!) Lecture module as part of the Elective Module (5 ECTS, not graded) 	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46524	Homogeneous Catalysis - Lab Homogeneous catalysis - Lab	5 ECTS
2	Courses / lectures	Praktikum: Homogeneous Catalysis - Lab (7 SWS) <ul style="list-style-type: none"> Attendance in lab course is compulsory! Attendance at safety instructions is compulsory! A valid laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Prof. Dr. Sjoerd Harder Prof. Dr. Romano Dorta Prof. Dr. Andriy Mokhir Prof. Dr. Karsten Meyer Prof. Dr. Svetlana Tsogoeva	

4	Module coordinator	Prof. Dr. Sjoerd Harder
5	Contents	Students <ul style="list-style-type: none"> Students get in touch with modern research topics in the field of homogenous catalysis & research tools or advanced spectroscopic tools (Practical work in one of the involved research groups!) deepens her/his knowledge in special topics of homogeneous catalysis that are in the research focus of the involved research groups will be trained in the practical aspects of advanced homogeneous catalysis
6	Learning objectives and skills	Students <ul style="list-style-type: none"> get in touch with modern research topics in the field of homogenous catalysis & research tools or advanced spectroscopic tools manage the preparation and full characterization of catalysts learn methods to evaluate catalyst performance in a series of catalytic reactions and are able to discuss scope and relevance
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 <ul style="list-style-type: none"> Lab module within the Compulsory Elective Module (20 ECTS)!
10	Method of examination	Practical achievement Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)
11	Grading procedure	Practical achievement (100%)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 105 h

		Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

Advances in Interface Research and Catalysis A

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Quantum Chemistry*
- *Advances in Organic Chemistry*

1	Module name 46531	Theory of Catalytic Processes Theory of catalytic processes	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Theory of Catalytic Processes (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Bernd Meyer	
5	Contents	<ul style="list-style-type: none"> • Introduction: catalysis as interdisciplinary multiscale problem • Atomic structure and thermodynamic stability of surfaces and adsorbates, surface phase diagrams, active surface sites • Reactivity concepts: d-band model, Newns-Anderson model of adsorption, trends in the periodic table • Microkinetic modelling, rate equations, steady-state limit, rate determining step, transition state theory, kinetic Monte Carlo • Descriptors for catalytic activity, scaling relations (Bronsted-Evans-Polanyi), volcano plots • Accelerated molecular dynamics for sampling rare events 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • are familiar with the most common quantum-chemical methods for studies in heterogeneous catalysis • understand the principles of descriptor-based predictions of catalytic activity • can perform first quantum chemical calculations on their own and interpret the data 	
7	Prerequisites	Basic knowledge of quantum mechanics and quantum chemical calculations is strongly recommended!	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Within the Compulsory Elective Module Advances in Interfaces and Catalysis A (IntCat-3A) (20 ECTS in total)! • Module can also be taken as part of the elective module (5 ECTS, not graded)! <p>Please note: MSc Molecular NANO Science students have to attend the module "Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann!</p>	
10	Method of examination	Oral (20 minutes) PL: O 20 - oral examination (20 minutes)	
11	Grading procedure	Oral (100%) <ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Interfaces and Catalysis A (IntCat-3A)": 100% 	

		<ul style="list-style-type: none"> As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> J.K. Norsov, T. Bligaard, J. Rossmeisl, C.H. Christensen, Towards the computational design of sold catalysts, Nature Chemistry, 2009, 1, 37

1	Module name 46532	Surface and Interface Science Surface and interface science	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Surfaces and Interface Science (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr.-Ing. Marcus Bär	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	<ul style="list-style-type: none"> • Vacuum and pressure measurement (pumps, pressure and flow ranges) • Lab-based and synchrotron-based light sources (principles, optics, insertion devices, etc.) • Theory of photoemission and electronic structure • XPS (elemental / chemical sensitivity, cross sections, quantification, examples) • UPS (gas phase, adsorbates, 2D band structures, 3D band structures, orbital tomography) • IPES (probing of unoccupied states, energy level alignment determination) • HAXPES (depth-resolved photoemission measurements, examples) • PEEM (spatially-resolved photoemission measurements, examples) • NEXAFS (principle and examples) • XES & RIXS (principle and examples) • Structure of surfaces/ diffraction at surfaces (LEED, definitions and examples) • X-ray spectroscopy based materials research on energy conversion devices (examples from current research) 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • understand the principles of photoemission variants and their applications • can judge the quality of data evaluation and its pitfalls • can deliberately select an X-ray spectroscopic analysis method to address given scientific question and are able to evaluate the collected data 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Interfaces and Catalysis" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%)	

		As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg • Ertl, Küppers: Low Energy Electrons and Surface Chemistry" VCH Weinheim • D. Attwood, Soft X-rays and Extreme Ultraviolet Radiation, Cambridge University Press, 1999 • D. Briggs, M.P. Seah: Practical Surface Analysis: Auger and X-Ray Photoelectron Spectroscopy, Wiley, 1996 • M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. • A. Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989

1	Module name 46533	Heterogeneous Catalysis and Kinetics Heterogeneous catalysis and kinetics	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Heterogeneous Catalysis and Kinetics (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	<p>Syllabus:</p> <ul style="list-style-type: none"> • Concepts in heterogeneous catalysis: definition of terms, industrial processes • Characterization methods for real catalysts (in-situ and operando methods, TEM, SEM, XRD, EXAFS, XANES, XPS, SIMS, DRIFTS, Raman, TPR, etc.) • Surface Reaction Dynamics: dynamics of adsorption, reaction, desorption, molecular beam experiments, laser spectroscopies • Elementary Kinetics: microkinetics, transition-state theory, relaxation kinetics, rate-determining step; microkinetic experiments, TAP, SSITKA, etc. • Model Catalysis: growth processes, preparation and characterization of model catalysts; kinetics on nanostructured surfaces • Energy-related model catalysis (incl. examples from current research) 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • acquire the professional competence in heterogeneous catalysis and respective topics • obtain advanced knowledge in different experimental or theoretical models, their application to current problems, the corresponding data evaluation and interpretation using current research examples • get familiar with various modern experimental techniques and are able to apply them in a targeted manner 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective Module "Advances in Interface Science & Catalysis" in M. Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As part of the Elective Module in M.Sc. Chemistry/MSc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	Only in summer semester	

13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	see lecture notes

1	Module name 46534	Interfaces and Catalysis - Lab Interfaces and catalysis - Lab	5 ECTS
2	Courses / lectures	Praktikum: Interface & Catalysis LAB (7 SWS) <ul style="list-style-type: none"> • Attendance in lab course is compulsory! • Please check lab instructions (contact lab supervisor) • Laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Prof. Dr. Julien Bachmann Prof. Dr. Bernd Meyer Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda
5	Contents	<p>Practical introduction to state-of-the-art research in the fields of surface science, interface science, heterogeneous catalysis, electrocatalysis or materials characterization.</p> <p>Guided work on a current research project in a research group. Research topics may cover spectroscopy at surfaces, microscopy at surfaces, in-situ or operando spectroscopy, characterization of catalytic materials, in-situ methods in electrocatalysis, preparation and characterization of nanomaterials, modelling and simulation of interfaces and nanomaterials or similar.</p> <p>Practical laboratory experience to introduce state-of-the-art experimental tools in surface and catalysis research, among them:</p> <ul style="list-style-type: none"> • Electron spectroscopies • Vibrational spectroscopies • Microscopy at interfaces • Other characterization methods for surfaces / interfaces • In-situ and operando spectroscopy and microscopy • Characterization of nanomaterials • Electrochemical in-situ characterization • Photochemical / photoelectrochemical in-situ characterization • Modelling on processes at interfaces
6	Learning objectives and skills	<p>The students ...</p> <ul style="list-style-type: none"> • get familiar with the current state-of-knowledge for a specific research topic. • apply fundamental knowledge of physical chemistry to a specific research topic. • understand and test model-like descriptions for complex physicochemical problems. • operate complex state-of-the-art instrumentation. • get in contact with development of new methodologies to answer open questions in interface science and catalysis. • analyze data with state-of-the-art methodologies. • record, document, and analyze research data in in appropriate form.

		<ul style="list-style-type: none"> • present and discuss experimental results and develop interpretations. • present own results in written form and scientific style English language.
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Obligatory lab course module (5 ECTS) within the Compulsory elective module Interfaces and Catalysis within the degree programmes M.Sc. Chemistry or M.Sc. Molecular Science
10	Method of examination	<p>Practical achievement PL: Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)</p> <p>Please note: Module examination organized by supervising group!</p>
11	Grading procedure	Practical achievement (100%)
12	Module frequency	Every semester
13	Workload in clock hours	<p>Contact hours: 105 h Independent study: 45 h</p>
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	Will be provided by the supervising research group

Advances in energy materials

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Quantum Chemistry*
- *Advances in Organic Chemistry*

1	Module name 46541	Semiconductor Materials for Energy Applications Semiconductor materials for energy applications	5 ECTS
2	Courses / lectures	Seminar: Semiconductor Materials for Energy Applications - Seminar (1 SWS) Vorlesung: Semiconductor Materials for Energy Applications (2 SWS)	5 ECTS
3	Lecturers	Dr. Ryan Crisp	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> Fundamentals of semiconductors: Crystal structure, Electronic structure, Electrical transport, Interaction with light Semiconductor devices: Tunnelling, The pn junction, The transistor Photovoltaics: Principles, Types of solar cells The interface to a solution: Charged electrolytic interfaces, Electrocatalysis and photoelectrocatalysis 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> are familiar with the fundamentals and modern developments in semiconductor science and applications understand theoretical and practical aspects in state-of-the-art semiconductor devices can present, communicate and discuss scientific results with experts in English. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> within the Compulsory Elective Module "Advances in Energy Materials" in M. Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total) part of the Elective Module in M. Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Oral</p> <p>O20 (PL): Oral examination, 20 minutes</p>	
11	Grading procedure	<p>Oral (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46542	Advanced Electrochemistry Advanced electrochemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Electrochemistry (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Dr. Christian Ehli	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> • Comprehensive survey of the fundamentals for electrode processes (thermodynamics and kinetics) • Introduction to electrochemical techniques (e.g. cyclic voltammetry, rotating disk voltammetry, differential pulse voltammetry, spectroelectrochemistry, electrochemical impedance spectroscopy) • Applications of electrochemistry (e.g. corrosion prevention, batteries) • Seminars will be based on the discussion of practical aspects and electrochemical exercises 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • plan and perform own electrochemical experiments • characterize electroactive materials by common electrochemical methods • analyze, interpret and discuss electrochemical experimental results • discuss and evaluate current electrochemical publications 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	<p>Written examination (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	

16 **Bibliography**

- Allen J. Bard, Larry R. Faulkner: "Electrochemical Methods: Fundamentals and Applications", John Wiley & Sons, New York, NY
- Carl H. Hamann, Andrew Hamnett, Wolf Vielstich: "Electrochemistry", Wiley-VCH, Weinheim

For further literature, please see the current list on studon.

1	Module name 46543	Solar Energy Conversion Solar energy conversion	5 ECTS
2	Courses / lectures	<p>Vorlesung mit Übung: Solar Energy Conversion (3 SWS) (WiSe 2026)</p> <p>Please note: The lecture "Solar Energy Conversion" by Prof. Dirk M. Guldi was replaced by the lecture "Solar Energy Conversion - Inorganic Photochemistry" by Dr. Alejandro Cadranel in winter term 2025/26 due to a sabbatical of Prof. Guldi. Now the lecture of Prof. Guldi will be held again. Students of the master program "MSc Energietechnik" can attend the lecture again from winter term 2026/27!</p>	5 ECTS
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Dr. Alejandro Cadranel Prof. Dr. Dirk Michael Guldi
5	Contents	<ul style="list-style-type: none"> Fundamentals of Photochemistry: Light-matter interaction, electronic structure and properties of excited states, radiative and non-radiative transitions between states. Jablonski diagrams. Photoinduced energy and electron transfer. Inorganic vs. organic systems, molecular orbitals, MLCT, LMCT, LC and MC excited states. LLCT and MMCT excited states. MC states: Tanabe-Sugano diagrams Mixed-valence systems: ground-state and excited-state Photochemistry of [Ru(bpy)₃]²⁺, other d6 systems like Ru(II), Os(II), Re(I), Ir(III) Photochemistry of earth-abundant metals, Iron(II/III), Cu(I), Cr(III), Co(III), Ni(II), Zr(IV) Overview of instrumental techniques Applications: Solar-energy conversion, photoredox catalysis, photomedicine
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> are familiar with the fundamentals and modern applications of inorganic photochemistry. understand the design principles of photoactive transition-metal complexes. can present, communicate and discuss scientific results with experts in English.
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p>

		<ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Energy Materials" MSc Chemistry and Molecular Science (20 ECTS in total!) • Module can be taken as part of the Elective Module, too (5 ECTS, not graded)!
10	Method of examination	Written examination (60 minutes)
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • D. Bahnemann, A.O. Patrocínio, "Handbook of Inorganic Photochemistry", Springer 2022, ISBN 978-3-030-63712-5 (book) • V. Balzani, P. Ceroni, A. Juris, "Photochemistry and Photophysics", Wiley-VCH 2014, ISBN 978-3-527-33479-7 (book) • N.J. Turro, V. Ramamurthy, J.C. Scaiano, "Principles of Molecular Photochemistry", University Science Books 2009, ISBN 978-1891389573

1	Module name 46544	Energy Materials - Lab Energy materials - Lab	5 ECTS
2	Courses / lectures	Praktikum: Energy Materials - LAB (7 SWS) <u>Please note:</u> <ul style="list-style-type: none"> Attendance at safety instruction and lab course is compulsory! Students have to register for the module on StudOn at www.chemie.fau.de/pcfpraktikum - all organisational information due to the EnMat lab course is available there! The EnMat lab course takes place in one of the participating research groups! If you have any further questions, please visit the PC lab course office in person. It is located in the Physical Chemistry building (Egerlandstr. 3), room P 0.72. A valid laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Dr. Christian Ehli Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> Practical introduction to electrochemical techniques Guided work on the characterization of electroactive materials Attempts to solve independently a scientific problem Documentation of experimental results 	
6	Learning objectives and skills	Students <ul style="list-style-type: none"> plan and perform own electrochemical experiments characterize electroactive materials by common electrochemical methods analyze, interpret, and discuss electrochemical experimental results discuss and evaluate current electrochemical publications. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 <ul style="list-style-type: none"> Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! 	
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)	
11	Grading procedure	Practical achievement (100%)	

12	Module frequency	Every semester Please note: <ul style="list-style-type: none"> • EnMat lab course can be done in winter or summer term!!
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

Quantum chemistry

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Quantum Chemistry*
- *Advances in Organic Chemistry*

1	Module name 46507	Quantum Chemistry Quantum chemistry	10 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry 2 (3 SWS) (SoSe 2026)	5 ECTS
		Vorlesung mit Übung: Quantum Chemistry 1 (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Andreas Görling Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling
5	Contents	<ul style="list-style-type: none"> • Introduction to modern methods and current research issues in the field of quantum and computer chemistry • Hartree-Fock, DFT, Many Body Perturbation Theory • Configuration Interaction, Second Quantization, Coupled Cluster • TD-HF, TD-DFT, RPA
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • obtain sound knowledge in basic and advanced methods of quantum chemistry • are able to solve mathematical problems occurring in quantum chemistry • are able to understand and assess scientific reports in the field of quantum chemistry
7	Prerequisites	<p>Required Qualifications:</p> <ul style="list-style-type: none"> • good knowledge of basic quantum mechanics: axioms of QM, application to simple systems (particle in a box, harmonic oscillator, rigid rotator) • good knowledge in mathematics: differential calculus of functions of several variables, basic linear algebra
8	Integration in curriculum	semester: 1;2
9	Module compatibility	<p>Quantum Chemistry Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Core module „Quantum Chemistry“ in M. Sc. Chemistry • Lecture module within the Compulsory Elective Module in M.Sc. Chemistry (if not chosen as Core module) or M. Sc. Molecular Science
10	Method of examination	Oral (30 minutes)
11	Grading procedure	Oral (100%)
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 90 h Independent study: 210 h
14	Module duration	2 semester

15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none">• Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996• Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.)• Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.)

1	Module name 46508	Quantum Chemistry laboratory	10 ECTS
2	Courses / lectures	<p>Praktikum: Training in Applied Computational Chemistry (5 SWS) (SoSe 2026) or</p> <p>Praktikum: Training in Applied Computational Chemistry (5 SWS) (WiSe 2026)</p> <p>Praktikum: Quantum Chemistry - Lab / Scientific Programming 1 und 2 (5 SWS/WiSe) + (5 SWS/SoSe)</p> <p>Please note:</p> <ul style="list-style-type: none"> • Attendance in lab course is compulsory! • Module starts only in winter term (Duration: 2 semester)! • "Scientific Programming" starts only in winter term! • "Training in Applied Computational Chemistry" can be done in one of the Theoretical Chemistry groups (Profs Görling, Imhof, B. Meyer, Zahn), time and place by agreement, summer OR winter term! • Students have to register for the module in their first semester (check registration periods) • Registration / further information via StudOn 	<p>5 ECTS</p> <p>5 ECTS</p> <p>5 ECTS</p>
3	Lecturers	<p>Prof. Dr. Dirk Zahn</p> <p>Prof. Dr. Dirk Zahn</p> <p>Dr. Christian Neiß</p> <p>Dr. Christian Neiß</p>	

4	Module coordinator	Prof. Dr. Andreas Görling
5	Contents	<ul style="list-style-type: none"> • Introduction to Linux • Numerical Methods to solve equations and perform optimizations • Scientific programming in Python and Fortran • Using numerical and mathematical libraries/modules • Introduction to parallel computing • Exercises
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • get familiar with Linux as operating system for HPC • are able to create computer programs to solve numerical problems • can use numerical and mathematical libraries/modules • obtain knowledge about basic parallelization paradigms
7	Prerequisites	Good knowledge in quantum theory, theoretical chemistry, and/or quantum chemical calculations is highly recommended

8	Integration in curriculum	semester: 1;2
9	Module compatibility	Quantum Chemistry Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Lab module within the Core module „Quantum Chemistry“ in M. Sc. Chemistry • Lab module within the Compulsory Elective Module in M.Sc. Chemistry (if not chosen as Core module) or M. Sc. Molecular Science • or as Elective module:1. as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)2. as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science if 1. was already chosen as Elective Module (5 ECTS, not graded)
10	Method of examination	Practical achievement Successful implementation of the programming project (working program)
11	Grading procedure	Practical achievement (100%) 100% Graded Computer Program
12	Module frequency	Only in winter semester Start possible only in winter term
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h
14	Module duration	2 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Stephen J. Chapman: Fortran for Scientists and Engineers, McGraw Hill 2017 (4th ed.) • Bernd Klein: Einführung in Python 3, Hanser 2017 (3rd ed.) • Stefan Gerlach: Computerphysik, Springer Spektrum 2019 (2nd ed.)

Advances in Organic Chemistry

Please choose **only 1 mandatory elective module**(-package) out of entire program:

- *Advances in bio-organic and bio-inorganic chemistry*
- *Advances in homogeneous catalysis*
- *Advances in Interface Research and Catalysis A*
- *Advances in energy materials*
- *Quantum Chemistry*
- *Advances in Organic Chemistry*

1	Module name 46568	Organic Chemistry Spectroscopy Lab	5 ECTS
2	Courses / lectures	Praktikum: Organic Chemistry Spectroscopy Lab (7 SWS) <ul style="list-style-type: none"> • Attendance in lab course is compulsory! • Attendance at safety instructions is compulsory! • A valid laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Prof. Dr. Jürgen Schatz Prof. Dr. Andreas Hirsch Prof. Dr. Andriy Mokhir apl. Prof. Dr. Norbert Jux Prof. Dr. Svetlana Tsogoeva apl. Prof. Dr. Evgeny Kataev Prof. Dr. Henry Dube	

4	Module coordinator	apl. Prof. Dr. Norbert Jux	
5	Contents	The students will analyse one (or two, depending on complexity) organic compound prepared in one of the research groups mentioned above by <ul style="list-style-type: none"> • ¹H, ¹³C, and, if necessary, ¹⁹F, ³¹P NMR spectroscopies • Advanced NMR techniques such as various 2D methods, HETCOR, etc. • Mass spectrometries such as MALDI-TOF, ESI • UV/Vis absorption and fluorescence spectroscopies • IR and Raman spectroscopies • ECD (electronic circular dichroism) spektroskopie 	
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • understand the preparation of samples in terms of purity and concentration • are able to handle the spectrometers and analytical machinery • are able to process spectra and information from raw data • know how to properly store experimental data 	
7	Prerequisites	It is strongly recommended to have attended the lecture module " <i>Advanced Spectroscopy in Organic Chemistry</i> " as preparation!	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Organic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 Lab module within the Compulsory Elective Module " <i>Advances in Organic Chemistry</i> " (20 ECTS)!	
10	Method of examination	Practical achievement pÜL: <ul style="list-style-type: none"> • Analysis of one (or two, depending on complexity) organic molecule provided from the above-mentioned research groups 	

		<ul style="list-style-type: none"> Graded Lab protocol with processed spectra, 30-50 pages, including raw data documentation
11	Grading procedure	Practical achievement (100%)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> M. Hesse, H. Meier, B. Zeeh, Spektroskopische Methoden in der organischen Chemie, Thieme, 8. Edition, 2011, ISBN: 9783135761077. H. Friebolin, Basic one- and two-dimensional NMR spectroscopy, Wiley-VCH, 5th ed., 2010, ISBN: 9783527327829. T. N. Mitchell, B. Costisella, NMR – From spectra to structures, Springer, 2nd ed., 2007, ISBN: 9783540721956. D. H. Williams, I. Fleming, McGraw Hill, Spectroscopic methods in organic chemistry, Springer, 7th ed., 2019, ISBN: 9783030182519. K. M. Sanders, E. C. Constable, B. K. Hunter, C. M. Pearce, Modern NMR spectroscopy. A Workbook of Chemical Problems, Oxford University Press, 1989, ISBN: 9780198552871.

1	Module name 46569	Stereoselective Synthesis	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Stereoselective Synthesis (3 SWS)	5 ECTS
3	Lecturers	apl. Prof. Dr. Evgeny Kataev	

4	Module coordinator	apl. Prof. Dr. Evgeny Kataev	
5	Contents	<p>The students are introduced to advanced aspects of stereoselective synthesis:</p> <ul style="list-style-type: none"> • General/Fundamental principles of stereochemistry. • Determination of absolute configuration and enantiomeric composition. • Various types of stereoselective reactions, including alkylation, aldol-related reactions, asymmetric Diels-Alder, and cyclization reactions • Stereo- and enantioselective metal-catalyzed reactions, organocatalytic transformation, and enzymatic reactions. • Historical and recent developments in the field are given. • Exemplary stereoselective syntheses of essential drugs and other molecules. 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • understand mechanisms in terms of selectivities, conditions, additives, • are able to offer synthetic pathways to various chiral drugs and other molecules. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Organic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Written examination (60 minutes)</p> <p>W60 (PL): Written examination (60 minutes)</p>	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> • Written examination: 100 % • As part of the elective module: 0% - pass/fail 	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none">• Eliel, Wilen, Doyle, Basic Organic Stereochemistry, WileySons 2001;• Lin, Li, Chan, Principles and application of asymmetric synthesis, WileySons, 2001;• Carreira, Kvaerno, Classics in Stereoselective Synthesis, Wiley-VCH 2009;• Carey, F. A.; Sundberg, R. J.; Organic Chemistry Part A and B, Springer, Berlin 2007.

1	Module name 46571	Heterocyclic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Heterocyclic Chemistry (3 SWS)	5 ECTS
3	Lecturers	apl. Prof. Dr. Norbert Jux	

4	Module coordinator	apl. Prof. Dr. Norbert Jux	
5	Contents	<p>The students are introduced to fundamental and advanced aspects of heterocyclic chemistry:</p> <ul style="list-style-type: none"> Nitrogen-, oxygen- and sulphur-containing heterocycles are discussed. Syntheses, characterization and properties of heterocycles with ring sizes from three to seven atoms are presented. Examples from pharmacy (drugs) and materials science underline the importance of heterocycles. Historical and recent developments in the field are given. Retrosynthetic analyses of heterocycles and drugs are described. Named reactions and important contributors to the field are presented. 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> understand the formation reactions of heterocycles are able to analyse drugs and other molecules in terms of their components, in particular, heterocycles are able to offer synthetic pathways to various heterocycle-containing drugs and other molecules 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> Lecture module within the Compulsory elective module "Advances in Organic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Written examination (60 minutes)</p> <p>W60 (PL): Written examination (60 minutes)</p>	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> As part of the Compulsory Elective module "Advances in Organic Chemistry": W60, 100% As part of the elective module: 0% - pass/fail 	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	

14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • J. A. Joule, K. Mills, Heterocyclic Chemistry, Wiley-Blackwell, 5th Edition, 2010, ISBN-13: 978-1405133005. • T. L. Gilchrist, Heterocyclic Chemistry, Pearson Education Dorling Kindersley, 3rd Edition, 1997, ISBN: 978-8131707937. • Peter A. Jacobi, Introductory Heterocyclic Chemistry, Wiley, 1st Edition, 2018, ISBN: 978-1119417590. • T. Eicher, S. Hauptmann, A. Speicher, The Chemistry of Heterocycles, 3rd edition, Wiley 2013, ISBN: 978-3-527-66986-8.

1	Module name 46572	Advanced Spectroscopy in Organic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Spectroscopy in Organic Chemistry (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Dr. Alexander Scherer	

4	Module coordinator	Prof. Dr. Henry Dube	
5	Contents	<p>Fundamentals of spectroscopy in organic chemistry will be reviewed. More in-depth methods of molecular spectroscopy in organic chemistry are covered. Advanced methods of NMR-spectroscopy are also covered, such as NMR-spectroscopy of various nuclei (e.g., ^1H, ^{13}C, ^{19}F, ^{31}P). Two-dimensional methods of NMR-spectroscopy using scalar spin-spin couplings (e.g., HSQC, HMBC) are discussed. Furthermore, NMR-spectroscopic methods relying on interactions between coupling nuclear dipoles, which are transmitted directly through space are covered (e.g., NOESY).</p> <p>In addition, other optical spectroscopic methods will be reviewed and discussed in more depth (e.g., UV/Vis-, CD-, IR-spectroscopy and Mass spectrometry).</p>	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • master the reliable use and gain an understanding of spectroscopic methods in organic chemistry, which are used to elucidate organic molecules; • are able to characterize unknown molecules and to determine their structure as well as their dynamics and interactions (the correlations between the spectroscopic results and the characteristics of the molecules should become understandable and comprehensible); • discuss practical examples of spectroscopic results and the related correlations in the exercises and practice the structure elucidation on examples. 	
7	Prerequisites	Basic knowledge of spectroscopy is recommended!	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective module "Advances in Organic Chemistry" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> • As part of the Compulsory Elective module "Advances in Organic Chemistry": W60, 100% • As part of the elective module: 0% - pass/fail 	

12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • "Basic one- and two-dimensional NMR spectroscopy, edited by H. Friebolin, Wiley-VCH • "NMR From spectra to structures edited by T. N. Mitchell, B. Costisella, Springer • "Spectroscopic methods in organic chemistry, Edited by D. H. Williams, I. Flemming, McGraw Hill • "Modern NMR spectroscopy, Edited by J. K. M. Sanders, E. C. Constable, B. K. Hunter, C. M. Pearce, Oxford

Elective modules

Elective modules (15 ECTS in total):

specialist (please see below) or non-specialist modules have to be chosen, which may include, e.g., language courses, key skills, science courses, or others (the eligible courses are listed online in **Campo** in the **course catalog** of the current term under: "*Electives, soft skills and language courses*")

1	Module name 46511	Advanced Bio-Organic and Bio-Inorganic Chemistry Advanced bio-organic and bio-inorganic chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (2 SWS) (WiSe 2026) Seminar: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine - Seminar (1 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Nicolai Burzlaff	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	<p>The students</p> <ul style="list-style-type: none"> • are introduced into recent activities and achievements in the fields • of bioorganic and bioinorganic chemistry and metals in medicine: • porphyrins and heme enzymes • PDT • electron transfer (cofactors, ferredoxins, inner sphere mechanism, outer sphere mechanism, Marcus theory) • photosynthesis • copper containing proteins and enzymes • nitrogenases and other Mo containing enzymes • drugs based on Pt, Ru, Au and As • biominerals • implant materials and technology 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Oral</p> <p>O20 (PL): Oral examination (20 minutes)</p>	
11	Grading procedure	<p>Oral (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	Contact hours: 45 h	

		Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> W. Kaim, B. Schwederski, A. Klein, Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life, 2. Edition, John Wiley & Sons, Ltd, 2013

1	Module name 46512	Special Aspects in Bio-Organic Chemistry Special aspects in bio-organic chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Special Aspects in Bio-Organic Chemistry (2 SWS) (WiSe 2026) Übung: Special Aspects in Bio-Organic Chemistry: Seminar (1 SWS) (WiSe 2026) Please note: <ul style="list-style-type: none"> • Seminar: time and place by agreement! 	5 ECTS
3	Lecturers	Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff
5	Contents	The students <ul style="list-style-type: none"> • learn about: nucleic acid structure, synthesis, and reactivity; nucleic acid modifications; nucleic acid-protein interactions, nucleic acid-based drugs, nucleic acid biochemistry, modelling of nucleic acids
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • can explain, apply and reflect upon the theories, terminology, specialities, boundaries and different school of bioorganic and bioinorganic chemistry critically and in depth
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • A. Bloomfield, D.M. Crothers, I. Tinoco, Jr. „Nucleic Acids: structures, properties, and functions“, University Science Books, Sausalito, CA, USA 1999

- D. L. van Vranken, G. A. Weiss, Introduction to Bioorganic Chemistry and Chemical Biology, 1. Aufl., Garland Publishers 2012, ISBN: 978-0815342144
- W. Saenger, „Principles of Nucleic Acid Structure“, Springer-Verlag New York Inc. 1984
- T. Schlick: "Molecular modelling and simulation: An interdisciplinary guide", Springer New York Dordrecht Heidelberg London 2nd ed. 2010
- A. Vologodskii: "Biophysics of DNA", Cambridge University press, 2015, DOI 10.1017/CBO9781139542371

1	Module name 46521	Organometallic Catalysis Organometallic catalysis	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organometallic Catalysis - Sustainable and Enantioselective Catalysis (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Sjoerd Harder Prof. Dr. Romano Dorta	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	<p>Knowledge on homogeneous organometallic catalysis will be deepened by discussing the latest scientific breakthroughs and developments in modern areas of catalysis. The module consists of two parts.</p> <p>Harder-Part <i>Sustainable Main Group Metal Catalysis</i></p> <ul style="list-style-type: none"> • Activation of strong bonds and small molecules with main group metal complexes (who needs d-orbitals?). Replacing expensive rare transition metals with cheap and abundant main group metals. • Bond activation using the ball-mill method. • Catalysis with main group metal complexes (discussion of elementary steps and catalytic cycles). • Activation of metals and making metal catalysts using Metal-Vapor-Synthesis. The evaporation of metals under vacuum and co-condensation in an organic matrix is a great way to make highly active metal catalysts. <p>Dorta-Part <i>Asymmetric Catalysis by Chiral Metal Complexes – Applications in Industry</i></p> <ul style="list-style-type: none"> • Importance of chiral molecules in nature and life science • Chiral pharmaceutical drugs and agrochemicals • Asymmetric catalysis: Chiral catalysts acting as a chirality multiplier • Advantages in comparison with racemate separation or stoichiometric enantioselective syntheses. • Historical development of chiral technology and industrial applications. • Synthesis of enantiopure ligands (famous industrial examples will be discussed). • Mechanistic aspects (catalytic cycles). 	
6	Learning objectives and skills	<p>The student</p> <ul style="list-style-type: none"> • will learn the latest developments at the forefront of organometallic catalysis • will understand that the field of catalysis needs to become sustainable and that we move away from expensive rare metals to abundant main group metals • will be able to critically discuss catalytic cycles and understand reaction mechanisms 	

		<ul style="list-style-type: none"> • will be taught new methods (practical and theoretical) to study molecular properties • will be understand the need for new methods of catalyst preparation • will be up-to-date with asymmetric catalysis and industrial applications
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	<p>Advances in Homogenous Catalysis Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective module "Advances in Homogenous Catalysis" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, ungraded)
10	Method of examination	<p>Oral</p> <p>O20(PL): Oral examination (20 minutes)</p>
11	Grading procedure	<p>Oral (100%)</p> <ul style="list-style-type: none"> • As part of the Compulsory Elective module "Advances in Homogenous Catalysis": O20, 100% • As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 • Early Main Group Metal Catalysis – Concepts and Reactions, Sjoerd Harder, Wiley-VCH, 2020 • Metal-catalysis in Industrial Organic Processes, Gian Paolo Chiusoli & Peter Maitlis, Royal Society of Chemistry, 2019 • Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 • Homogeneous Catalysis, Parshall & Ittel, 2nd ed., Wiley Interscience 1992

1	Module name 46522	Organocatalysis and Catalytic Reactions in Water Organocatalysis and catalytic reactions in water	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organocatalysis and Catalytic Reactions in Water (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Andriy Mokhir Prof. Dr. Svetlana Tsogoeva	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	<ul style="list-style-type: none"> History and basic principles of organocatalysis, Different types of organocatalysts, Enantioselective organocatalysis, Domino reactions and important examples. Catalytic reactions in water: Important reactions in biological chemistry, Metabolism of biomolecules and artificial bioorthogonal reactions. 	
6	Learning objectives and skills	<p>The student</p> <ul style="list-style-type: none"> can explain basic principles and underlying reaction mechanisms in organo- and organometallic catalysis and biological chemistry deepens her/his knowledge in special topics of homogeneous catalysis that are in the research focus of the involved research groups is able to construct important reaction mechanisms and catalytic cycles and can critically discuss each step. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Homogenous Catalysis Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <p>Organocatalysis and Catalytic Reactions in Water can be taken</p> <ul style="list-style-type: none"> within the Compulsory Elective Module "Advances in Homogenous Catalysis" (20 ECTS in total) as Elective Module (5 ECTS, not graded) 	
10	Method of examination	<p>Oral</p> <p>O20(PL): Oral examination (20 minutes)</p>	
11	Grading procedure	<p>Oral (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46523	Small Molecule Activation Small molecule activation	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Small Molecule Activation (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Karsten Meyer Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	Lecture: <ul style="list-style-type: none"> Inorganic coordination chemistry, catalytic and electrocatalytic transformation of abundant small molecules N₂, O₂, CO or CO₂, H₂O and NH₃, into value-added commodities, fine-chemicals, active pharmaceutical ingredients and polymers. Selective oxidation reactions. CO₂ reduction to CO and subsequent Fischer-Tropsch catalysis for the production of carbon-based fuels. Reductive activation of atmospheric N₂ to NH₃ and the production of H₂ from H₂O splitting catalysis. Applications, e.g. fuel-cell technologies for carbon-free energy production. 	
6	Learning objectives and skills	The student <ul style="list-style-type: none"> can explain basic principles and underlying reaction mechanisms in small molecule activation is able to construct important reaction mechanisms and catalytic cycles can critically discuss small-molecule activation 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> Lecture module within the Compulsory Elective Module (20 ECTS in total!) Lecture module as part of the Elective Module (5 ECTS, not graded) 	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46531	Theory of Catalytic Processes Theory of catalytic processes	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Theory of Catalytic Processes (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Bernd Meyer	
5	Contents	<ul style="list-style-type: none"> • Introduction: catalysis as interdisciplinary multiscale problem • Atomic structure and thermodynamic stability of surfaces and adsorbates, surface phase diagrams, active surface sites • Reactivity concepts: d-band model, Newns-Anderson model of adsorption, trends in the periodic table • Microkinetic modelling, rate equations, steady-state limit, rate determining step, transition state theory, kinetic Monte Carlo • Descriptors for catalytic activity, scaling relations (Bronsted-Evans-Polanyi), volcano plots • Accelerated molecular dynamics for sampling rare events 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • are familiar with the most common quantum-chemical methods for studies in heterogeneous catalysis • understand the principles of descriptor-based predictions of catalytic activity • can perform first quantum chemical calculations on their own and interpret the data 	
7	Prerequisites	Basic knowledge of quantum mechanics and quantum chemical calculations is strongly recommended!	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Within the Compulsory Elective Module Advances in Interfaces and Catalysis A (IntCat-3A) (20 ECTS in total)! • Module can also be taken as part of the elective module (5 ECTS, not graded)! <p>Please note: MSc Molecular NANO Science students have to attend the module "Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann!</p>	
10	Method of examination	Oral (20 minutes) PL: O 20 - oral examination (20 minutes)	
11	Grading procedure	Oral (100%) <ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Interfaces and Catalysis A (IntCat-3A)": 100% 	

		<ul style="list-style-type: none"> As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> J.K. Norsov, T. Bligaard, J. Rossmeisl, C.H. Christensen, Towards the computational design of solid catalysts, Nature Chemistry, 2009, 1, 37

1	Module name 46532	Surface and Interface Science Surface and interface science	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Surfaces and Interface Science (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr.-Ing. Marcus Bär	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	<ul style="list-style-type: none"> • Vacuum and pressure measurement (pumps, pressure and flow ranges) • Lab-based and synchrotron-based light sources (principles, optics, insertion devices, etc.) • Theory of photoemission and electronic structure • XPS (elemental / chemical sensitivity, cross sections, quantification, examples) • UPS (gas phase, adsorbates, 2D band structures, 3D band structures, orbital tomography) • IPES (probing of unoccupied states, energy level alignment determination) • HAXPES (depth-resolved photoemission measurements, examples) • PEEM (spatially-resolved photoemission measurements, examples) • NEXAFS (principle and examples) • XES & RIXS (principle and examples) • Structure of surfaces/ diffraction at surfaces (LEED, definitions and examples) • X-ray spectroscopy based materials research on energy conversion devices (examples from current research) 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • understand the principles of photoemission variants and their applications • can judge the quality of data evaluation and its pitfalls • can deliberately select an X-ray spectroscopic analysis method to address given scientific question and are able to evaluate the collected data 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Interfaces and Catalysis" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%)	

		As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg • Ertl, Küppers: Low Energy Electrons and Surface Chemistry" VCH Weinheim • D. Attwood, Soft X-rays and Extreme Ultraviolet Radiation, Cambridge University Press, 1999 • D. Briggs, M.P. Seah: Practical Surface Analysis: Auger and X-Ray Photoelectron Spectroscopy, Wiley, 1996 • M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. • A. Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989

1	Module name 46533	Heterogeneous Catalysis and Kinetics Heterogeneous catalysis and kinetics	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Heterogeneous Catalysis and Kinetics (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	<p>Syllabus:</p> <ul style="list-style-type: none"> • Concepts in heterogeneous catalysis: definition of terms, industrial processes • Characterization methods for real catalysts (in-situ and operando methods, TEM, SEM, XRD, EXAFS, XANES, XPS, SIMS, DRIFTS, Raman, TPR, etc.) • Surface Reaction Dynamics: dynamics of adsorption, reaction, desorption, molecular beam experiments, laser spectroscopies • Elementary Kinetics: microkinetics, transition-state theory, relaxation kinetics, rate-determining step; microkinetic experiments, TAP, SSITKA, etc. • Model Catalysis: growth processes, preparation and characterization of model catalysts; kinetics on nanostructured surfaces • Energy-related model catalysis (incl. examples from current research) 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • acquire the professional competence in heterogeneous catalysis and respective topics • obtain advanced knowledge in different experimental or theoretical models, their application to current problems, the corresponding data evaluation and interpretation using current research examples • get familiar with various modern experimental techniques and are able to apply them in a targeted manner 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective Module "Advances in Interface Science & Catalysis" in M. Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As part of the Elective Module in M.Sc. Chemistry/MSc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	Only in summer semester	

13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	see lecture notes

1	Module name 46535	Neurotech: Physics and Chemistry of Neuromodulation Technologies Neurotech: Physics and chemistry of neuromodulation technologies	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Neurotech: Physics and Chemistry of Neuromodulation Technologies (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Danijela Gregurec	

4	Module coordinator	Prof. Dr. Danijela Gregurec	
5	Contents	<ul style="list-style-type: none"> • Introduction to neuromodulation technologies (definition, history, nervous anatomy, stimulation targets-ion channels, action potential) • Imaging and spectroscopic concepts (MRI, EEG, Calcium imaging, electrophysiology) • Current tech and principles (Invasive and noninvasive approaches, Deep brain stimulation, Transcranial magnetic stimulation, Pain management, BMI summary) • Organic materials and approaches (viral vectors, optogenetics, Chemogenetics (DREED)) • Micro- and macroscale materials (Mechanical properties and compatibility of neural implants, Electrodes (Utah arrays, Neuralink), Flexible electrodes, Optical fibers) • Nanomaterials (Nanomaterial properties leveraged for neuromodulation (Importance of surface chemistry in bio(nano)materials, Quantum confinement and quantum dots, Plasmons and photothermal neuromodulation, Magnetism, Magnetoelectric particles for electric stimulation, Magnetic nanoparticles for magnetothermal and magnetomechanical stimulation) 	
6	Learning objectives and skills	<p>Students will gain the knowledge, skills, and competences to be able to ...</p> <ul style="list-style-type: none"> • understand biophysical aspects of neuronal signaling and its correlation to cognition and behavior. • learn physical foundations, biological concepts, and chemical approaches crucial for materials used in neuromodulation and neurotechnology. • apply acquired knowledge to realize design criteria of technology that governs the modulation of neuronal signaling. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <p>Module compatibility:</p> <ul style="list-style-type: none"> • Lecture module within the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Oral (20 minutes)</p> <p>Oral examination (20 minutes, ungraded)</p>	
11	Grading procedure	Oral (pass/fail)	

		not graded - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Knotkova, Rasche; Springer-Verlag New York (2015) Textbook of Neuromodulation: Principles, Methods and Clinical Applications • Luan et al, Front Neuroeng 7 (2014) Neuromodulation: present and emerging methods • Frank, et al, Nat Biotech 37 (2019) Next-generation interfaces for studying neural function • Chen et al, Nat Rev Mat 2 (2017) Neural recording and modulation technologies <p> **Literature will be updated with leading peer-reviewed papers during lectures </p>

1	Module name 46536	Lebensmittelchemie Food chemistry	5 ECTS
2	Courses / lectures	<p>Seminar: Instrumentelle Analytik für Lebensmittelchemiker und das Wahlmodul Lebensmittelchemie /Master Molecular Science (1 SWS)</p> <p>Vorlesung: Chemie und Technologie der Lebensmittel IV - Bioaktive pflanzl. Lebensmittel (1 SWS)</p> <p>Vorlesung: Lebensmittelchemie V - Proteine (1 SWS)</p> <p>Bitte beachten:</p> <ul style="list-style-type: none"> • Die Vorlesung Lebensmittelchemie hat je nach Semester unterschiedliche inhaltliche Schwerpunkte (Lebensmittelchemie I-XI)! • Die Studierenden müssen 2 Vorlesungen (im Umfang von 2 SWS) und das Seminar (1 SWS) besuchen! 	<p>2,5 ECTS</p> <p>2,5 ECTS</p>
3	Lecturers	Prof. Dr. Monika Pischetsrieder	

4	Module coordinator	Prof. Dr. Monika Pischetsrieder	
5	Contents	<ul style="list-style-type: none"> • Es werden toxikologisch, technologisch und physiologisch relevante Inhaltsstoffe von Lebensmitteln ausführlich vorgestellt und diskutiert. • Ausgehend von den grundlegenden Kenntnissen der organischen Chemie werden Reaktionsmechanismen, die während der Prozessierung oder Entstehung von Lebensmitteln ablaufen, erläutert. • Ausgehend von grundlegenden Kenntnissen der analytischen Chemie werden die wichtigsten weiterführenden und aktuellen instrumentellanalytischen und bioanalytischen Analysemethoden besprochen. 	
6	Learning objectives and skills	<p>Die Studierenden</p> <ul style="list-style-type: none"> • erarbeiten sich die Sachkompetenz zur theoretischen Beurteilung und praktischen Handhabung wichtiger Fragestellungen der Lebensmittelchemie • sind in der Lage, die wichtigsten relevanten Arbeitstechniken aus dem Gebiet der Lebensmittelchemie und analytik und eines anderen Gebiets der Lebensmittelwissenschaften selbständig anzuwenden • können die wesentlichen Prinzipien der Lebensmittelanalytik auf praktische Probleme anwenden und kritisch reflektieren. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • kann als Teil des Wahlmoduls/Elective module belegt werden (5 ECTS/unbenotet) 	
10	Method of examination	Seminar achievement (20 minutes)	

		(SL): Seminarvortrag, 30 Minuten, unbenotet
11	Grading procedure	Seminar achievement (pass/fail) SL (0%): unbenotet bestanden/nicht bestanden
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	german
16	Bibliography	Wird von den Dozenten aktualisiert zur Verfügung gestellt.

1	Module name 46541	Semiconductor Materials for Energy Applications Semiconductor materials for energy applications	5 ECTS
2	Courses / lectures	Seminar: Semiconductor Materials for Energy Applications - Seminar (1 SWS) Vorlesung: Semiconductor Materials for Energy Applications (2 SWS)	5 ECTS
3	Lecturers	Dr. Ryan Crisp	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> Fundamentals of semiconductors: Crystal structure, Electronic structure, Electrical transport, Interaction with light Semiconductor devices: Tunnelling, The pn junction, The transistor Photovoltaics: Principles, Types of solar cells The interface to a solution: Charged electrolytic interfaces, Electrocatalysis and photoelectrocatalysis 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> are familiar with the fundamentals and modern developments in semiconductor science and applications understand theoretical and practical aspects in state-of-the-art semiconductor devices can present, communicate and discuss scientific results with experts in English. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> within the Compulsory Elective Module "Advances in Energy Materials" in M. Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total) part of the Elective Module in M. Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Oral</p> <p>O20 (PL): Oral examination, 20 minutes</p>	
11	Grading procedure	<p>Oral (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46542	Advanced Electrochemistry Advanced electrochemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Electrochemistry (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Dr. Christian Ehli	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> • Comprehensive survey of the fundamentals for electrode processes (thermodynamics and kinetics) • Introduction to electrochemical techniques (e.g. cyclic voltammetry, rotating disk voltammetry, differential pulse voltammetry, spectroelectrochemistry, electrochemical impedance spectroscopy) • Applications of electrochemistry (e.g. corrosion prevention, batteries) • Seminars will be based on the discussion of practical aspects and electrochemical exercises 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • plan and perform own electrochemical experiments • characterize electroactive materials by common electrochemical methods • analyze, interpret and discuss electrochemical experimental results • discuss and evaluate current electrochemical publications 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	<p>Written examination (100%)</p> <p>As part of the elective module: 0% - pass/fail</p>	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	

16	Bibliography	<ul style="list-style-type: none">• Allen J. Bard, Larry R. Faulkner: "Electrochemical Methods: Fundamentals and Applications", John Wiley & Sons, New York, NY• Carl H. Hamann, Andrew Hamnett, Wolf Vielstich: "Electrochemistry", Wiley-VCH, Weinheim <p>For further literature, please see the current list on studon.</p>
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1	Module name 46543	Solar Energy Conversion Solar energy conversion	5 ECTS
2	Courses / lectures	<p>Vorlesung mit Übung: Solar Energy Conversion (3 SWS) (WiSe 2026)</p> <p>Please note: The lecture "Solar Energy Conversion" by Prof. Dirk M. Guldi was replaced by the lecture "Solar Energy Conversion - Inorganic Photochemistry" by Dr. Alejandro Cadranel in winter term 2025/26 due to a sabbatical of Prof. Guldi. Next winter term the lecture of Prof. Guldi will be held again, and Students of the master program "MSc Energietechnik" can attend the lecture again from winter term 2026/27!</p>	5 ECTS
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Dr. Alejandro Cadranel Prof. Dr. Dirk Michael Guldi	
5	Contents	<ul style="list-style-type: none"> Fundamentals of Photochemistry: Light-matter interaction, electronic structure and properties of excited states, radiative and non-radiative transitions between states. Jablonski diagrams. Photoinduced energy and electron transfer. Inorganic vs. organic systems, molecular orbitals, MLCT, LMCT, LC and MC excited states. LLCT and MMCT excited states. MC states: Tanabe-Sugano diagrams Mixed-valence systems: ground-state and excited-state Photochemistry of [Ru(bpy)₃]²⁺, other d6 systems like Ru(II), Os(II), Re(I), Ir(III) Photochemistry of earth-abundant metals, Iron(II/III), Cu(I), Cr(III), Co(III), Ni(II), Zr(IV) Overview of instrumental techniques Applications: Solar-energy conversion, photoredox catalysis, photomedicine 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> are familiar with the fundamentals and modern applications of inorganic photochemistry. understand the design principles of photoactive transition-metal complexes. can present, communicate and discuss scientific results with experts in English. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Energy Materials Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p>	

		<ul style="list-style-type: none"> • Within the Compulsory Elective Module "Advances in Energy Materials" MSc Chemistry and Molecular Science (20 ECTS in total!) • Module can be taken as part of the Elective Module, too (5 ECTS, not graded)!
10	Method of examination	Written examination (60 minutes)
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • D. Bahnemann, A.O. Patrocínio, "Handbook of Inorganic Photochemistry", Springer 2022, ISBN 978-3-030-63712-5 (book) • V. Balzani, P. Ceroni, A. Juris, "Photochemistry and Photophysics", Wiley-VCH 2014, ISBN 978-3-527-33479-7 (book) • N.J. Turro, V. Ramamurthy, J.C. Scaiano, "Principles of Molecular Photochemistry", University Science Books 2009, ISBN 978-1891389573

1	Module name 46545	Organic thin films	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organic Thin Films (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Rainer Fink	

4	Module coordinator	Prof. Dr. Rainer Fink
5	Contents	<ul style="list-style-type: none"> • Molecular interactions and molecular self-organization • Influence of molecular geometry and functionalization • Thin film preparation techniques: Langmuir-Blodgett, Self-Assembled Monolayers (SAMs), other solvent-based techniques (e.g. spin-casting, doctor blading, etc.), vacuum sublimation • Analytical techniques, in-situ analysis • Effect of templates • Organic thin film applications • Seminars will be based on recent literature
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • get insight into the major preparation techniques of organic thin films • are able to evaluate to prepare organic thin film specimens • know to analyse thin film specimens with respect to structural and electronic properties • are aware of recent studies and modern applications of organic thin films
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • lecture module can be taken as part of the Elective Module!
10	Method of examination	Variable O20(PL): Oral examination (20 minutes)
11	Grading procedure	Variable (pass/fail) pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46546	Symmetry and Group Theory Symmetry and group theory	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Symmetry and Group Theory - Applications in Chemistry, Physics and Material Sciences (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	<ul style="list-style-type: none"> • Symmetry of Molecules (symmetry elements, operations, point groups, notations) • Symmetry of Crystals, Surfaces and Interfaces (symmetry in 1, 2 and 3 dimensional periodic structures, lattices, crystal classes, space groups) • Compact Course Group Theory (elements group theory, definitions, reducible and irreducible representations, orthogonality theorem, character tables) • Group Theory and Quantum Mechanics (representations, operators and symmetry, matrix elements, direct product functions, projection operators) • Symmetry of Organic Molecules: From Electronic Structure to Reactivity (symmetry adaption, cyclic groups, many electron systems, electronic transitions, configuration interaction, symmetry controlled reactions) • Symmetry in Anorganic Chemistry: From Atoms to Complexes (MO models, transition metal complexes, direct product groups, rotation inversion group, angular momentum coupling, crystal field splitting, vibronically allowed transitions) • Symmetry and Spectroscopy: Vibrational Spectroscopies (analysis of vibrational modes, normal coordinate analysis, symmetry of vibrational wave functions, vibrational spectroscopy, selection rules) • Symmetry in Crystal Physics: Tensor Description of Physical Properties (tensors, axial, polar, representations, transformation properties, intrinsic symmetry, Neumann's principle, Curie's principle) • Symmetry and Electronic Structure of Solids: Band Structures (translation group and irreps, reciprocal lattice, k-space, Bloch functions, Brillouin zones, symmetry of bands) 	
6	Learning objectives and skills	<p>Students</p> <ul style="list-style-type: none"> • acquire detailed understanding how to use symmetry properties and the mathematical tools of group theory in a broad range of application fields in chemistry, physics and materials science. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202	

		<ul style="list-style-type: none"> Lecture module can be taken as part of the Elective module (5 ECTS, not graded)
10	Method of examination	Variable O20(PL): Oral Examination (20 minutes)
11	Grading procedure	Variable (pass/fail) pass/fail
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46556	Scientific Programming Scientific programming	5 ECTS
2	Courses / lectures	Praktikum: Quantum Chemistry - Lab / Scientific Programming 2 (5 SWS) (SoSe 2026) Praktikum: Quantum Chemistry - Lab / Scientific Programming 1 (5 SWS) (WiSe 2026) Please note: <ul style="list-style-type: none"> Attendance in the lab course is mandatory! Module starts only in winter term (duration: 2 semesters) 	2,5 ECTS 2,5 ECTS
3	Lecturers	Dr. Christian Neiß Dr. Christian Neiß Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	<ul style="list-style-type: none"> Operating system Linux for high-performance computing (HPC) Scientific programming in Fortran and Python Using numerical and mathematical libraries/modules Introduction to parallel computing Exercises Programming project 	
6	Learning objectives and skills	Students <ul style="list-style-type: none"> get familiar with Linux as operating system for HPC are able to create computer programs for scientific purposes can use numerical and mathematical libraries/modules in home-made programs obtain knowledge about basic parallelization paradigms 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) - please note: the module cannot be combined with the module Quantum Chemistry - Lab 	
10	Method of examination	Practical achievement PL: Successful implementation of the programming project (working program), ungraded - module has to be passed	
11	Grading procedure	Practical achievement (pass/fail) not graded: pass/fail	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	Contact hours: 150 h Independent study: 0 h	
14	Module duration	2 semester	

15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none">• Stephen J. Chapman: Fortran for Scientists and Engineers, McGraw Hill 2017 (4th ed.)• Bernd Klein: Einführung in Python 3, Hanser 2017 (3rd ed.)• Stefan Gerlach: Computerphysik, Springer Spektrum 2019 (2nd ed.)

1	Module name 46557	Theory of Surface Phenomena Theory of surface phenomena	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Theory of Surface Phenomena / Theorie der Oberflächenphänomene (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Bernd Meyer	
5	Contents	<ul style="list-style-type: none"> • Brief introduction into quantum-chemical methods for surface science studies • Introduction of basic nomenclature of how to describe the atomic and electronic structure of surfaces • Basic concepts on how to understand the electronic properties of metal, semiconductor and insulator surfaces, such as surface states, dangling bonds, passivation, charge neutralization with respect to polar and nonpolar surfaces • Thermodynamic analysis of the stability of surface structures; surface phase diagrams • Methods for calculating STM and AFM data to support the analysis of experimental data from local probe measurements 	
6	Learning objectives and skills	<p>The students ...</p> <ul style="list-style-type: none"> • are familiar with the most common theoretical and experimental techniques for surface science studies • have a sound knowledge in basic principles governing surface structure and reactivity • can perform first quantum chemical calculations on their own and interpret the data 	
7	Prerequisites	Basic knowledge of quantum mechanics and quantum chemical calculations is strongly recommended	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Elective module (5 ECTS, ungraded) within the M.Sc. degree programme Chemistry or M.Sc. Molecular Science (especially Molecular NANO science) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (pass/fail) pass/fail	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46558	Modern X-ray Structure Determination Modern X-ray structure determination	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Modern X-ray structure determination of single crystals (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Dr. Frank Wilhelm Heinemann	

4	Module coordinator	Dr. Frank Wilhelm Heinemann	
5	Contents	<ul style="list-style-type: none"> • Fundamentals of crystallization and polymorphism • Structural description of single crystals, crystal systems, unit cell, symmetry and symmetry elements, space groups • Diffraction power of crystals, diffraction conditions, structure factor • Generation of X-rays, single crystal diffractometers, detection techniques • Structure solution techniques and refinement procedures, software, problems and pitfalls, interpretation of results • Anomalous dispersion and absolute structure • Graphical representations, use of data bases 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • get insight into thermodynamics of crystallization and crystallization techniques • get fundamentals of the theory behind crystal structure determination • get practice in crystal selection, mounting and measurement set-up • get hands-on training in structure solution and refinement using up-to-date software • are enabled to interpret and compare results of a single crystal structure determination 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Tutorial achievement (PL): Lab report in manuscript style (max. 2000 words plus raw data), not graded	
11	Grading procedure	Tutorial achievement (pass/fail) not graded: pass/fail	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	

16	Bibliography	<ul style="list-style-type: none">• Werner Massa: Kristallstrukturbestimmung. Teubner Studienbücher Chemie, Vieweg und Teubner, 6. Auflage, 2009, ISBN: 3834806498• William Clegg: Crystal Structure Determination. Oxford Chemistry Primers. Oxford University Press, 1998, ISBN: 0198559011• Further literature will be recommended in the course
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1	Module name 46559	Quantum Chemistry 1 Quantum chemistry 1	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry 1 (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	<ul style="list-style-type: none"> Mathematical concepts and current research issues in the field of quantum and computer chemistry Hartree-Fock, DFT 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> obtain sound knowledge in basic methods of quantum chemistry are able to solve mathematical problems occurring in quantum chemistry are able to understand and assess scientific reports in the field of quantum chemistry 	
7	Prerequisites	<p>Strongly recommended Qualifications:</p> <ul style="list-style-type: none"> good knowledge of basic quantum mechanics: axioms of QM, application to simple systems (particle in a box, harmonic oscillator, rigid rotator) good knowledge in mathematics: differential calculus of functions of several variables, linear algebra 	
8	Integration in curriculum	semester: 1;3	
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Oral (20 minutes)</p> <p>O20 (PL): Oral Examination (20 minutes, not graded: pass/fail)</p>	
11	Grading procedure	<p>Oral (pass/fail)</p> <p>not graded: pass/fail</p>	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	<ul style="list-style-type: none"> Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996 Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.) Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.) 	

1	Module name 46561	Quantum Chemistry 2 Quantum chemistry 2	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry 2 (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	<ul style="list-style-type: none"> • Many-Body Perturbation Theory • Configuration Interaction, Second Quantization, Coupled Cluster • TD-HF, TD-DFT, RPA 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • obtain sound knowledge in advanced methods of quantum chemistry • are able to solve mathematical problems occurring in quantum chemistry • are able to understand and assess scientific reports in the field of quantum chemistry 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;4	
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) if Quantum Chemistry I was already chosen as Elective Module! 	
10	Method of examination	<p>Oral (20 minutes) O20 (PL): Oral Examination (20 minutes, not graded: pass/fail)</p>	
11	Grading procedure	<p>Oral (pass/fail) not graded: pass/fail</p>	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h Independent study: 105 h</p>	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	<ul style="list-style-type: none"> • Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996 • Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.) • Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.) 	

1	Module name 46562	Biological and Synthetic Molecular Switches and Machines Biological and synthetic molecular switches and machines	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Molecular Switches and Molecular Machines (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Henry Dube	

4	Module coordinator	Prof. Dr. Henry Dube	
5	Contents	Examination of molecular triggers, switches and machines in biology and in synthetic systems as foundation of nanotechnology; working mechanisms; types of systems; design principles; seminal contributions; historical backgrounds are given. The material is ordered in introduction and context, basic principles, triggers, switches, machines, integrated systems, future prospective. The course will be updated to implement the newest developments yearly	
6	Learning objectives and skills	<p>The Students ...</p> <ul style="list-style-type: none"> • acquire a fundamental understanding in the working mechanisms and design principles of molecular triggers, switches, and machines. • will be able to develop strategies for implementing responsiveness into nanostructured biological or synthetic systems and will be equipped with an exhaustive overview of historical developments and current state of the art in the field by discussing representative examples in depth. • will therefore be educated in one of the most prominent fields of modern (bio)chemistry and nanosciences. <p>Compulsory attendance will be necessary. The skills will be appropriate for Masters level and will partially repeat and build on knowhow from supramolecular, biological, and photochemistry as well as on fundamental physical organic chemistry, biochemistry, nanotechnology, and spectroscopy.</p>	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Elective modules Master of Science Molecular Science 20202</p> <p>Module compatibility:</p> <ul style="list-style-type: none"> • as Elective Module in MSc Molecular Life Science (not applicable for Molecular Nanoscience), 5 ECTS/not graded • as Elective Module in MSc Chemistry, 5 ECTS/not graded 	
10	Method of examination	<p>Tutorial achievement</p> <p>O20 (SL): 20 min oral examination in the form of a seminar talk presenting the content of a seminal original publication (Non-graded seminar presentation)</p>	
11	Grading procedure	<p>Tutorial achievement (pass/fail)</p> <p>pass/fail</p>	
12	Module frequency	Only in summer semester	

		Seminar: time and place by arrangement!
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • "Molecular Switches, edited by Ben L. Feringa, Wiley-VCH • "From Non-Covalent Assemblies to Molecular Machines, Edited by Jean-Pierre Sauvage & Pierre Gaspard, Wiley-VCH • "Molecular Machines and Motors Recent Advances and Perspectives edited by Alberto Credi, Serena Silvi, Margherita Venturi, Springer

1	Module name 46564	Modern Methods in Mass Spectrometry Modern methods in mass spectrometry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Modern Methods in Mass Spectrometry (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Prof. Dr. Thomas Drewello	

4	Module coordinator	Prof. Dr. Thomas Drewello	
5	Contents	<ul style="list-style-type: none"> Advanced aspects of Soft Ionization methods (MALDI, ESI, related Atmospheric Pressure Ionization methods) Advanced aspects of mass analyzers (FT-ICR, orbitrap, hybrids: multi sector, BEqQ, QToF, QIT-ToF) Ion Activation (CID, BIRD, IRMPD, SID, ECID) Applications (thermochemistry, kinetic method, equilibrium methods, ion/molecule-reactions) "Omics (proteomics, petroleomics, metabolomics) Further applications: ¹⁴C dating, accelerator MS, stable isotope MS, MS in space, ICP-MS. Seminars in form of problem solving classes 	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> gain insight into the different ion formation processes are able to decide on an ionization method for a given compound class are able to evaluate the use of different mass spectrometers gain understanding of ion activation processes in the gas phase 	
7	Prerequisites	The course builds on the master module "Applied Spectroscopy (Compulsory Module Physical Chemistry)	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> Lecture module as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (pass/fail) pass/fail	
12	Module frequency	Only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	<ul style="list-style-type: none"> Jürgen H. Groß: "Mass Spectrometry, a textbook Springer, Heidelberg Edmond De Hoffmann: "Mass Spectrometry, principles and applications, Wiley 	

1	Module name 46567	HZB Photon School HZB photon school	5 ECTS
2	Courses / lectures	Vorlesung: HZB Photon School - Advanced synchrotron-based photon science methods (2 SWS) (WiSe 2026)	2,5 ECTS
		Praktikum: HZB Photon School - Practical training at synchrotron facility (2 SWS) (WiSe 2026) <u>Please note:</u> The HZB Photon School usually takes place during the lecture-free period between the winter and summer terms!	2,5 ECTS
3	Lecturers	Prof. Dr.-Ing. Marcus Bär	

4	Module coordinator	Prof. Dr.-Ing. Marcus Bär	
5	Contents	<p>Students will be introduced to the</p> <ul style="list-style-type: none"> • functionality of a synchrotron (storage ring-based light sources, x-ray optics and beamlines for synchrotron radiation experiments), • fundamentals of the interaction of x-rays with matter, and • laboratory-based characterization methods that probe the molecular structure, function, and dynamics of complex material systems such as x-ray absorption spectroscopy, x-ray photoemission spectroscopy, magnetic spectroscopy and scattering, and x-ray microscopy. <p>The two-week program consists of basic and specialized lectures including exercises, given by experienced teachers who are HZB scientists or BESSY II super-users from HZB partner universities in the first week. The second week is dedicated to hands-on training at several BESSY II experimental stations and X-ray laboratories. Participants will get a vivid experience by joining up to two experiments to work in small groups. The second week's activities will teach participants how to collect, analyse, interpret, and present own data amongst peers and specialists.</p> <p>For further information, visit hz-b.de/photonschool</p>	
		6	Learning objectives and skills

		<ul style="list-style-type: none"> • represent their measurement results • are able to write a compelling beamtime proposal
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3;4
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)
10	Method of examination	Oral (20 minutes) PL (O20): Oral examination = Poster presentation (20 minutes)
11	Grading procedure	Oral (pass/fail) Ungraded, module has to be passed only
12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 60 h Independent study: 90 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg • Ertl, Küppers: Low Energy Electrons and Surface Chemistry", VCH Weinheim • D. Attwood, Soft X-rays and Extreme Ultraviolet Radiation, Cambridge University Press, 1999 • D. Briggs, M.P. Seah: Practical Surface Analysis: Auger and X-Ray Photoelectron Spectroscopy, Wiley, 1996 • M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag, GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. • Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989

1	Module name 46515	Sensory Sciences Lab	5 ECTS
2	Courses / lectures	<p>Projektseminar: Sensory Sciences Lab (for MSc Chemistry und MSc Molecular Science) (2 SWS) (WiSe 2026)</p> <p><u>Please note:</u></p> <ul style="list-style-type: none"> The module "Sensory Sciences Lab" is an interdisciplinary module with a limited number of participants: only 10 students from MSc Chemistry or MSc Molecular Science will be admitted! Students from Molecular LIFE Science will be given priority! The module "Sensory Sciences Lab" can be taken as part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" (20 ECTS in total, all modules have to be graded) or as an elective module (5 ECTS, ungraded)! 	5 ECTS
3	Lecturers	PD Dr. Helene Loos	

4	Module coordinator	Prof. Dr. Andrea Büttner
5	Contents	<p>The field of Sensory Sciences investigates how animals and humans sense, neurally process and perceive their environment. This research-oriented interdisciplinary course provides important concepts, theories, and methods of the Sensory Sciences to advanced students from three disciplines (Psychology, Medicine and Chemistry - Molecular Science). These topics comprise human auditory, visual and olfactory perception, chemocommunication, psychophysics, neuroimaging, molecular sensory receptors and technologies for neuromodulation (among others). The students apply these concepts in interdisciplinary teams to develop, conduct and analyse their own small research project. While engaging in an interdisciplinary collaboration and exchange, students can share and deepen their discipline-specific perspectives and competencies as well as learn about concepts and methods from other disciplines.</p>
6	Learning objectives and skills	<p>The students are able to</p> <ul style="list-style-type: none"> explain fundamental concepts and methods of the Sensory Sciences develop an own (limited) research question and design an empirical data collection to investigate the research question collect data in the laboratory and analyse the data present their findings in oral and written form communicate effectively in interdisciplinary teams understand and reflect the challenges and chances of interdisciplinary collaborations.
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3

9	Module compatibility	<p>Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" in M.Sc. Molecular Science (5 ECTS, graded; PO 2020) • elective module (5 ECTS, ungraded; PO 2020)
10	Method of examination	<p>Seminar achievement</p> <p>Seminar achievement: Written project report (maximum 30 pages), oral project presentation, experimental work in the lab</p>
11	Grading procedure	<p>Seminar achievement (100%)</p> <p>Seminar achievement: 50% project reports, 25% project presentation, 25% lab work</p>
12	Module frequency	Only in winter semester
13	Workload in clock hours	<p>Contact hours: 30 h</p> <p>Independent study: 120 h</p>
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • Literature will be announced in the first session • Scripts and materials will be provided via StudOn!

1	Module name 46568	Organic Chemistry Spectroscopy Lab	5 ECTS
2	Courses / lectures	Praktikum: Organic Chemistry Spectroscopy Lab (7 SWS) <ul style="list-style-type: none"> • Attendance in lab course is compulsory! • Attendance at safety instructions is compulsory! • A valid laboratory insurance is mandatory for participation in the lab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Prof. Dr. Jürgen Schatz Prof. Dr. Andreas Hirsch Prof. Dr. Andriy Mokhir apl. Prof. Dr. Norbert Jux Prof. Dr. Svetlana Tsogoeva apl. Prof. Dr. Evgeny Kataev Prof. Dr. Henry Dube	

4	Module coordinator	apl. Prof. Dr. Norbert Jux	
5	Contents	The students will analyse one (or two, depending on complexity) organic compound prepared in one of the research groups mentioned above by <ul style="list-style-type: none"> • ¹H, ¹³C, and, if necessary, ¹⁹F, ³¹P NMR spectroscopies • Advanced NMR techniques such as various 2D methods, HETCOR, etc. • Mass spectrometries such as MALDI-TOF, ESI • UV/Vis absorption and fluorescence spectroscopies • IR and Raman spectroscopies • ECD (electronic circular dichroism) spektroskopie 	
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • understand the preparation of samples in terms of purity and concentration • are able to handle the spectrometers and analytical machinery • are able to process spectra and information from raw data • know how to properly store experimental data 	
7	Prerequisites	It is strongly recommended to have attended the lecture module " <i>Advanced Spectroscopy in Organic Chemistry</i> " as preparation!	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Organic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 Lab module within the Compulsory Elective Module " <i>Advances in Organic Chemistry</i> " (20 ECTS)!	
10	Method of examination	Practical achievement pÜL: <ul style="list-style-type: none"> • Analysis of one (or two, depending on complexity) organic molecule provided from the above-mentioned research groups 	

		<ul style="list-style-type: none"> Graded Lab protocol with processed spectra, 30-50 pages, including raw data documentation
11	Grading procedure	Practical achievement (100%)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> M. Hesse, H. Meier, B. Zeeh, Spektroskopische Methoden in der organischen Chemie, Thieme, 8. Edition, 2011, ISBN: 9783135761077. H. Friebolin, Basic one- and two-dimensional NMR spectroscopy, Wiley-VCH, 5th ed., 2010, ISBN: 9783527327829. T. N. Mitchell, B. Costisella, NMR – From spectra to structures, Springer, 2nd ed., 2007, ISBN: 9783540721956. D. H. Williams, I. Fleming, McGraw Hill, Spectroscopic methods in organic chemistry, Springer, 7th ed., 2019, ISBN: 9783030182519. K. M. Sanders, E. C. Constable, B. K. Hunter, C. M. Pearce, Modern NMR spectroscopy. A Workbook of Chemical Problems, Oxford University Press, 1989, ISBN: 9780198552871.

1	Module name 46569	Stereoselective Synthesis	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Stereoselective Synthesis (3 SWS)	5 ECTS
3	Lecturers	apl. Prof. Dr. Evgeny Kataev	

4	Module coordinator	apl. Prof. Dr. Evgeny Kataev	
5	Contents	<p>The students are introduced to advanced aspects of stereoselective synthesis:</p> <ul style="list-style-type: none"> • General/Fundamental principles of stereochemistry. • Determination of absolute configuration and enantiomeric composition. • Various types of stereoselective reactions, including alkylation, aldol-related reactions, asymmetric Diels-Alder, and cyclization reactions • Stereo- and enantioselective metal-catalyzed reactions, organocatalytic transformation, and enzymatic reactions. • Historical and recent developments in the field are given. • Exemplary stereoselective syntheses of essential drugs and other molecules. 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • understand mechanisms in terms of selectivities, conditions, additives, • are able to offer synthetic pathways to various chiral drugs and other molecules. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Organic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Written examination (60 minutes)</p> <p>W60 (PL): Written examination (60 minutes)</p>	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> • Written examination: 100 % • As part of the elective module: 0% - pass/fail 	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none">• Eliel, Wilen, Doyle, Basic Organic Stereochemistry, WileySons 2001;• Lin, Li, Chan, Principles and application of asymmetric synthesis, WileySons, 2001;• Carreira, Kvaerno, Classics in Stereoselective Synthesis, Wiley-VCH 2009;• Carey, F. A.; Sundberg, R. J.; Organic Chemistry Part A and B, Springer, Berlin 2007.

1	Module name 46571	Heterocyclic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Heterocyclic Chemistry (3 SWS)	5 ECTS
3	Lecturers	apl. Prof. Dr. Norbert Jux	

4	Module coordinator	apl. Prof. Dr. Norbert Jux	
5	Contents	<p>The students are introduced to fundamental and advanced aspects of heterocyclic chemistry:</p> <ul style="list-style-type: none"> Nitrogen-, oxygen- and sulphur-containing heterocycles are discussed. Syntheses, characterization and properties of heterocycles with ring sizes from three to seven atoms are presented. Examples from pharmacy (drugs) and materials science underline the importance of heterocycles. Historical and recent developments in the field are given. Retrosynthetic analyses of heterocycles and drugs are described. Named reactions and important contributors to the field are presented. 	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> understand the formation reactions of heterocycles are able to analyse drugs and other molecules in terms of their components, in particular, heterocycles are able to offer synthetic pathways to various heterocycle-containing drugs and other molecules 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> Lecture module within the Compulsory elective module "Advances in Organic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	<p>Written examination (60 minutes)</p> <p>W60 (PL): Written examination (60 minutes)</p>	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> As part of the Compulsory Elective module "Advances in Organic Chemistry": W60, 100% As part of the elective module: 0% - pass/fail 	
12	Module frequency	Only in summer semester	
13	Workload in clock hours	<p>Contact hours: 45 h</p> <p>Independent study: 105 h</p>	

14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • J. A. Joule, K. Mills, Heterocyclic Chemistry, Wiley-Blackwell, 5th Edition, 2010, ISBN-13: 978-1405133005. • T. L. Gilchrist, Heterocyclic Chemistry, Pearson Education Dorling Kindersley, 3rd Edition, 1997, ISBN: 978-8131707937. • Peter A. Jacobi, Introductory Heterocyclic Chemistry, Wiley, 1st Edition, 2018, ISBN: 978-1119417590. • T. Eicher, S. Hauptmann, A. Speicher, The Chemistry of Heterocycles, 3rd edition, Wiley 2013, ISBN: 978-3-527-66986-8.

1	Module name 46572	Advanced Spectroscopy in Organic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Spectroscopy in Organic Chemistry (3 SWS) (WiSe 2026)	5 ECTS
3	Lecturers	Dr. Alexander Scherer	

4	Module coordinator	Prof. Dr. Henry Dube	
5	Contents	<p>Fundamentals of spectroscopy in organic chemistry will be reviewed. More in-depth methods of molecular spectroscopy in organic chemistry are covered. Advanced methods of NMR-spectroscopy are also covered, such as NMR-spectroscopy of various nuclei (e.g., ^1H, ^{13}C, ^{19}F, ^{31}P). Two-dimensional methods of NMR-spectroscopy using scalar spin-spin couplings (e.g., HSQC, HMBC) are discussed. Furthermore, NMR-spectroscopic methods relying on interactions between coupling nuclear dipoles, which are transmitted directly through space are covered (e.g., NOESY).</p> <p>In addition, other optical spectroscopic methods will be reviewed and discussed in more depth (e.g., UV/Vis-, CD-, IR-spectroscopy and Mass spectrometry).</p>	
6	Learning objectives and skills	<p>The students</p> <ul style="list-style-type: none"> • master the reliable use and gain an understanding of spectroscopic methods in organic chemistry, which are used to elucidate organic molecules; • are able to characterize unknown molecules and to determine their structure as well as their dynamics and interactions (the correlations between the spectroscopic results and the characteristics of the molecules should become understandable and comprehensible); • discuss practical examples of spectroscopic results and the related correlations in the exercises and practice the structure elucidation on examples. 	
7	Prerequisites	Basic knowledge of spectroscopy is recommended!	
8	Integration in curriculum	semester: 1;2;3	
9	Module compatibility	<p>Advances in Organic Chemistry Master of Science Molecular Science 20202</p> <p>Elective modules Master of Science Molecular Science 20202</p> <ul style="list-style-type: none"> • Lecture module within the Compulsory Elective module "Advances in Organic Chemistry" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, not graded) 	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	<p>Written examination (100%)</p> <ul style="list-style-type: none"> • As part of the Compulsory Elective module "Advances in Organic Chemistry": W60, 100% • As part of the elective module: 0% - pass/fail 	

12	Module frequency	Only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	<ul style="list-style-type: none"> • "Basic one- and two-dimensional NMR spectroscopy, edited by H. Friebolin, Wiley-VCH • "NMR From spectra to structures edited by T. N. Mitchell, B. Costisella, Springer • "Spectroscopic methods in organic chemistry, Edited by D. H. Williams, I. Flemming, McGraw Hill • "Modern NMR spectroscopy, Edited by J. K. M. Sanders, E. C. Constable, B. K. Hunter, C. M. Pearce, Oxford

1	Module name 46548	Economics	5 ECTS
2	Courses / lectures	Seminar: Economics - Chemistry (Blockveranstaltung) (5 SWS) Please note: <ul style="list-style-type: none"> The module takes place in mid/late September. 	5 ECTS
3	Lecturers	Dr. Andreas Späth	

4	Module coordinator	Dr. Andreas Späth
5	Contents	Tools for Strategic Assessment and Analysis for Entrepreneurship, Introduction to Project Management Techniques (with a focus on R&D), Introduction to Tools for Innovation Management
6	Learning objectives and skills	Students <ul style="list-style-type: none"> gain insights into various aspects of modern entrepreneurship are able to develop informed strategic decisions based on structured assessments understand various approaches to project management are able to transfer complex project guidelines into clear schedules and budget plans.
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3;4
9	Module compatibility	Elective modules Master of Science Molecular Science 20202
10	Method of examination	Variable SL: Preparation and presentation of a project study (presentation approx. 15 - 20 min)
11	Grading procedure	Variable (pass/fail) SL: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46516	Medical Life Sciences	5 ECTS
2	Courses / lectures	Vorlesung: Medical Life Sciences	5 ECTS
3	Lecturers	Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Ingrid Span
5	Contents	To lay the foundation for working in molecular disease research, the MedLife module provides knowledge in genetics, molecular biology, cell biology, nucleic acid and protein chemistry, enzymology, nuclear and nanomedicine.
6	Learning objectives and skills	The students <ul style="list-style-type: none"> • explain the basic principles and techniques of the various fields of research • engage with scientific texts • will be able to apply the essential principles to practical biological and chemical problems
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2;3
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 <ul style="list-style-type: none"> • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total)* • Lecture module as Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) *If students want to take the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" as a whole, they have to choose 3 out of the 4 lecture modules, the laboratory module "Bio-Organic and Bio-Inorganic Chemistry - Lab" (No.46514) must be taken compulsorily!
10	Method of examination	Written examination (60 minutes)
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	Only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

Research module

Students are responsible for finding a place in one of the working groups of the Department of Chemistry according to their specialization, their interests and the current availability.

The Research Module is intended as preparation for the subsequent Master's thesis.

1	Module name 63069	Research Lab - Drug Discovery Research lab - Drug discovery	15 ECTS
2	Courses / lectures	Praktikum: Research Module - Drug Discovery (23 SWS)	15 ECTS
3	Lecturers	Prof. Dr. Andreas Burkovski Prof. Dr. Peter Gmeiner Prof. Dr. Jutta Eichler Prof. Dr. Monika Pischetsrieder Prof. Dr. Markus Heinrich Prof. Dr. Susanne Mühlich Prof. Dr. Andrea Büttner Prof. Dr. Dagmar Fischer	

4	Module coordinator	Prof. Dr. Peter Gmeiner	
5	Contents	Research lab project full time (about 6 weeks: 21SWS LAB/2SWS Seminar) in a work group of the students choice at a research group in Medicinal Chemistry, Food Chemistry, or the disciplines of Biology involved in teaching at the Department of Chemistry and Pharmacy	
6	Learning objectives and skills	<p>Students ...</p> <ul style="list-style-type: none"> • are provided with the up-to-date practical and operative know-how suitable for future scientific and/or applied work in research institutes, pharmaceutical/food industry, medical care laboratories, bio-technological, bio-analytical and environmental branches • get an advanced theoretical background and an overview of emerging trends in life sciences (chemistry, biology and medicine) • look at living systems through the lens of basic chemical principles • are prepared to work in interdisciplinary environment and participate in national and international development of forefront fields such as translational medicine. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 3	
9	Module compatibility	Research module Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol (approx. 20 pages plus raw data documentation)	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	Every semester	
13	Workload in clock hours	Contact hours: 345 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	

1	Module name 63074	Research Lab - Molecular Nanoscience Research lab - Molecular nanoscience	15 ECTS
2	Courses / lectures	Praktikum: Research Module - Molecular Nanoscience (23 SWS)	15 ECTS
3	Lecturers	Prof. Dr. Dirk Zahn Prof. Dr. Julien Bachmann Prof. Dr. Rainer Fink Prof. Dr. Bernd Meyer Prof. Dr. Franziska Gröhn Prof. Dr. Karl Mandel Prof. Dr. Henry Dube Dr. Ryan Crisp	

4	Module coordinator	Prof. Dr. Rainer Fink
5	Contents	research lab project (23 SWS) fulltime in one of the research groups of Molecular Nanoscience
6	Learning objectives and skills	Students ... <ul style="list-style-type: none"> • are provided with the up-to-date practical and operative know-how suitable for future scientific and/or applied work in research institutes and nanoscience-labs • get an advanced theoretical background and an overview of emerging trends in nano sciences • look at living systems through the lens of basic chemical principles • are prepared to work in interdisciplinary environment and participate in national and international development of forefront fields.
7	Prerequisites	None
8	Integration in curriculum	semester: 3
9	Module compatibility	Research module Master of Science Molecular Science 20202
10	Method of examination	Practical achievement
11	Grading procedure	Practical achievement (100%) pÜL: Graded Lab Protocol (approx. 20 pages plus raw data documentation)
12	Module frequency	Every semester
13	Workload in clock hours	Contact hours: 345 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	