

Module description

for the degree programme

Master of Science Molecular Science

(Version of examination regulation: 20202)

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1	Module name 1999	Masterarbeit (M.Sc. Molecular Science 20202) Master's thesis	30 ECTS
2	Courses / lectures	No courses / lectures available for this module! Please note: • The master's thesis will be written at the student's conormally in one of the research groups of the Depa of Chemistry and Pharmacy or the Department of B • Students must independently apply for a master's the of the department's research groups!	rtment iology!
3	Lecturers	No lecturers available since there are no courses / lectures for this module!	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff/Supervising Professor	
5	Contents	Written elaboration in form of a scientific manuscript with a length of approx. 20,000 words. It describes the scientific findings as well as the way leading to these findings. It contains justifications for decisions regarding chosen methods for the thesis and discarded alternatives. The student's own substantial contribution to the achieved results has to be evident. In addition, the student presents his work in a seminar, in which the scientific quality and the scientific independence of his achievements are evaluated. Workload: 900h	
6	Learning objectives and skills	 Students demonstrate their ability to perform independent scientific work focusing on an adequately challenging research topic. rank their own research results in the context of current literature and research papers in the field and record their results in appropriate scientific writing and documentation style. give oral and written presentations of the results and acquired knowledge in an appropriate scientific style in English language. 	
7	Prerequisites	Before commencing work on the Master's thesis, students must have successfully completed the two subject-related compulsory elective modules, the supplementary compulsory elective module and the research module, coming to a total of 75 ECTS credits.	
8	Integration in curriculum	semester: 4	
9	Module compatibility	Pflichtmodul Master of Science Molecular Science 20202	
10	Method of examination	Written (6 Monate) • Thesis (2 hard copies in bound form + electronic version) with a length of approx. 20,000 words; • Referee report, 2 experts	
11	Grading procedure	Written (100%) Averaged grade of the two reports	
12	Module frequency	every semester	
13	Resit examinations	The exams of this moduls can only be resit once.	

14	Workload in clock hours	Contact hours: 675 h Independent study: 225 h
15	Module duration	1 semester
16	Teaching and examination language	english
17	Bibliography	

Studienrichtung Drug Discovery

1	Module name 63066	Biology-informed Drug Discovery	10 ECTS
2	Courses / lectures	Vorlesung: Biology-informed drug discovery (2 SWS)	
3	Lecturers	Prof. Dr. Thomas Winkler Prof. Dr. Christian Koch Prof. Dr. Lars Nitschke Prof. Dr. Steffen Backert Prof. Dr. Anja Lux Prof. Dr. Markus Albert Prof. Dr. Andreas Burkovski Prof. Dr. Yves Muller Prof. Dr. Gregor Fuhrmann	

4	Module coordinator	Prof. Dr. Yves Muller
5	Contents	Basic biology concepts of importance for the drug discovery process. The lectures cover three main topics, namely (i) selected experimental methods, (ii) the use of model organisms in drug discovery and (iii) the discussion of the biology of specific classes of drugs, namely antibodies and natural products. Part (i) focusses on omics methods (genomics, transcriptomics, proteomics, metabolomics) in drug target identification and drug validation as well as on the biological readout of high-throughput screening techniques. In addition, the question how to gain atomic insight into macromolecular drug targets is addressed by covering the topics of molecular modelling, nuclear magnetic resonance spectroscopy, X-ray crystallography and cryo-electron microscopy. In part (ii) basic methods in molecular biology are reviewed and the following model organisms are being addressed: Yeast as a not yet fully explored model organism for the identification of novel drug targets. The production of recombinant proteins and the synthesis of small molecule drugs by yeasts and aspergillus species are discussed. Plants as even less well characterized organisms for drug discovery, target screening and drug production as well as the third model organism mice and how the latter are exploited as disease model systems. Topics are: xenograft tumour models in mice and genetic mouse models for inherited diseases and tumour diseases. In part (iii) the isolation, production and use of monoclonal antibodies are being discussed together with natural products. A particular focus of the latter are novel antibiotic drugs and their role in the fight against increasingly prevalent bacterial infections. In addition, several concepts to enhance the targeted delivery of natural products are evaluated, using model dispositions, such as coeliac disease.
6	Learning objectives and skills	Learning objectives are the acquaintance of novel insights, concepts, and methods in biology-informed drug discovery. To be able to understand state-of-the-art methods/concepts in drug discovery and to critically assess the potential and limitations of the individual methods/concepts. To be able to independently develop working hypotheses

		and to design experimental approaches to further the drug discovery process. As an integral part of the module, each student will have to present the content of a primary research article in a seminar presentation of approximately 20 minutes. Each student acquires the skill to understand, present, and critically discuss primary research reports in the field of drug discovery and to defend the findings before an audience. Attendance of the student presentations is mandatory. Only regular attendance of these student presentations will lead to a successful completion of the module. Justification: The learning objectives of this part can only be reached if the students actively discuss their literature findings with fellow students.
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202
10	Method of examination	Written examination (90 minutes) PL: Written examination, 90 minutes
11	Grading procedure	Written examination (100%) (Please note: The grade of module "Biology-informed Drug Discovery", former: Concepts in Biology, is only a part of the total grade of module package "Drug Discovery"!) Grading procedure of "Drug Discovery": Medicinal Chemistry 25% + Biology-informed Drug Discovery (former: Concepts in Biology) 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%
12	Module frequency	every semester
13	Workload in clock hours	Contact hours: 60 h Independent study: 240 h
14	Module duration	2 semester
15	Teaching and examination language	english
16	Bibliography	Course material is provided via the StudOn platform

1	Module name 63068	Drug Discovery - Lab	10 ECTS
		The teaching units in the module are only offered in the summer semester.	
2	Courses / lectures	 Lab course Biomolecular simulations (5 SWS) Lab course Pharmacopeia-based drug analysis (5 SWS) 	3)
_		3)Lab course Instrumental and Bioanalytics (5 SWS)	,
		(Please note: Attendance in all lab courses and safety inscompulsory!)	tructions is
3	Lecturers	Prof. Dr. Monika Pischetsrieder	

4	Module coordinator	Prof. Dr. Monika Pischetsrieder	
5	Contents	3 independent practical courses focusing on biomolecular simulations (block, 5 SWS), Pharmacopoeia-based drug analysis (block, 5 SWS) and instrumental analytical techniques (block, 5 SWS)	
6	Learning objectives and skills	 The students are able to understand the basic and advanced principles of Pharmacopoeia-based drug analysis, instrumental and bioanalytics as well as computer chemistry-based applications in the field of modern drug research to utilize modern instrumental analytical techniques for characterisation of samples and quantification of target analytes to apply modern simulation techniques for the modeling of biomolecules to interpret and to critically summarize experimental results in written form (lab report) 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)	
11	Grading procedure	Practical achievement (100%) (Please note: The grade of module "Drug Discovery - Lab" is only a part of the total grade of module package "Drug Discovery"!) Grading procedure of "Drug Discovery": Medicinal Chemistry 25% + Concepts in Biology 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	Manuscripts are available on StudON

	1	Module name 63065	Medicinal Chemistry	10 ECTS
ĺ	2	Courses / lectures	Vorlesung mit Übung: Medicinal Chemistry (3 SWS)	
	3	Lecturers	Prof. Dr. Nicolai Burzlaff Prof. Dr. Andriy Mokhir Prof. Dr. Peter Gmeiner Dr. Stefan Löber Prof. Dr. Markus Heinrich Prof. Dr. Monika Pischetsrieder Dr. Dorothée Weikert Prof. Dr. Jutta Eichler Dr. Jürgen Einsiedel Prof. Dr. Susanne Mühlich	

4	Module coordinator	Prof. Dr. Peter Gmeiner	
5	Contents	Genomics, transcriptomics, proteomics; in-vitro assay systems, assay technology; target screening and drug production in plants, drug screening and production in yeast; experimental structural biology; chemoinformatics; molecular modeling: molecular dynamics simulation, force-fields, modeling of proteins, proteinligand docking; drug synthesis and combinatorial chemistry; redox-active metal complexes, metalloenzyme inhibitors; stereochemistry in drug design; organic reactions in medicinal chemistry; drug metabolism; peptidomimetics;	
6	Learning objectives and skills	The students are able to understand the basic and advanced principles of medicinal chemical, molecular biological and Computer chemistry based applications in the field of modern drug design research	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202	
10	Method of examination	Written examination (90 minutes)	
11	Grading procedure	Written examination (100%) (Please note: The grade of module "Medicinal Chemistry" is only a part of the total grade of module package "Drug Discovery"!) Grading procedure of "Drug Discovery": Medicinal Chemistry 25% + Concepts in Biology 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 90 h Independent study: 210 h	
14	Module duration	2 semester	
15	Teaching and examination language	english	

16	Bibliography	G. Klebe, Drug Design: Methodology, Concepts, and Mode-of-Action, Springer 2013
		Springer 2013

	1	Module name 63067	Molecular Modelling	10 ECTS
ĺ	2	Courses / lectures	Vorlesung mit Übung: Modelling of Complex Systems (2V/1UE) (3 SWS)	
	3	Lecturers	Prof. Dr. Dirk Zahn Prof. Dr. Petra Imhof	

4	Module coordinator	Prof. Dr. Petra Imhof	
5	Contents	 *WS:* Rationalizing Complex Systems from Statistics: Probability distributi-ons, Concepts of Information and Entropy; Thermodynamics of Monte-Carlo simulation and state-of-the art ana-lyses of transition pathways; Molecular dynamics simulations of complex systems: data evaluation and reduction to key information. *SS:* Model building and setup of first-guess configurations; Enhanced sampling techniques to tackle manifolds of configurations and to find reaction paths; Machine-learning and prediction of real-world properties from molecular data. 	
6	Learning objectives and skills	* master the basics of molecular modelling, * are able to select and apply from a wide range of different simulation techniques the one that is relevant and suitable for different problems, * can analyse, evaluate and present data and results, * monitor and control their own progress.	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Drug Discovery Master of Science Molecular Science 20202	
10	Method of examination	Written examination (90 minutes)	
11	Grading procedure	Written examination (100%) (Please note: The grade of module "Molecular Modelling" is only a part of the total grade of module package "Drug Discovery"!) Grading procedure of "Drug Discovery": Medicinal Chemistry 25% + Concepts in Biology 25% + Molecular Modeling 25% + Drug Discovery-Lab (Graded lab protocol) 25%	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 90 h Independent study: 210 h	
14	Module duration	2 semester	
15	Teaching and examination language	english	

16	Bibliography	B. Smit, D. Frenkel: Understanding Molecular Simulation: From Algo-rithms to Applications
	2.209)	A. Leach: Molecular Modelling: Principles and Applications

Studienrichtung Molecular Nanoscience

1	Module name 63071	Molecular Nanoscience I	15 ECTS
	Courses / lectures	Vorlesung mit Übung: Self-Assembly: Molecular, Particulate and Hybrid Nanostructures (3 SWS)	
		Seminar: Supramolecular Chemistry and Molecular Materials (2 SWS)	
2		Seminar: Nanoprobes I (2 SWS)	
		Seminar: From 2D assemblies to bulk (2 SWS)	
		Seminar: Molecular Nano Science (winter semester) (2 SWS)	
	Lecturers	Prof. Dr. Franziska Gröhn	
		Prof. Dr. Andreas Hirsch Dr. Tao Wei	
		Dr. Maria Eugenia Perez-Ojeda Rodriguez	
3		Prof. Dr. Rainer Fink	
		Prof. Dr. Julien Bachmann Prof. Dr. Norbert Jux	
		Prof. Dr. Dirk Michael Guldi	
		Prof. Dr. Karl Mandel	

4	Module coordinator	Prof. Dr. Rainer Fink Prof. Dr. Andreas Hirsch
5	Contents	 Concepts in supramolecular chemistry; host-guest chemistry; energetics of supramolecular complexes: experimental methods; templates and self-assembly. Molecular devices. Supramolecular catalysis: principles of supramolecular catalysis, supramolecular metal catalysis, self-assembled catalysts, metal-free catalysis, enzyme mimics, antibodies, imprinted polymers. Nanoscaled systems, general issues of microscopic techniques; experimental techniques with nanometer resolution: resolution determination; image processing techniques, light microscopic techniques with light in the visible and IR range, confocal techniques, scanning probes (STM,AFM and related scanning probes). Specific topics in synthesis and analysis of specific molecule-based nanoscale objects 2D film fabrication techniques (PVD, ALD, etc.)
6	Learning objectives and skills	 to explain the fundamental chemical and physical properties of nano-scale materials to distinguish and to compare some properties, structure and applications of different nanomaterials to describe and to evaluate the major concepts in supramolecular chemistry, molecular self-assembly and nano-scaled 2D materials and transfer knowledge to related topics

		to judge on potential nano-analytical tools with optimized image contrast
7	Prerequisites	None
8	Integration in curriculum	semester: 1;2
9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202
10	Method of examination	Oral O30 (PL): Oral Examination, 30 Minutes
11	Grading procedure	Oral (100%) (Please note: The grade of module "Molecular Nanoscience I" is only a part of the total grade of module package "Molecular Nanoscience"!)
12	Module frequency	only in winter semester
13	Workload in clock hours	Contact hours: 150 h Independent study: 300 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 63072	Molecular Nanoscience II	15 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Andreas Hirsch, etc.	

4	Module coordinator	Prof. Dr. Andreas Hirsch	
5	Contents	 Concepts in supramolecular chemistry; molecular switches, molecular machines; Nanoanalytical tools with charged particle: ion microscopy (FIM, HeIM, Nano-SIMS), electron microscopy (PEEM, LEEM, SEM, (S)TEM), aberration correction, x-ray tools, hybrid type microscopies; coherent imaging; 3D imaging (tomography, laminography) Carbon allotropes: from molecules to carbon nanodots, fullerenes, carbon nanotubes and nanohorns; synthesis, properties and applications. Inorganic nanoparticles and industrial applications, synthesis methods, sol-gel science and technology, concept of inorganic supraparticles & practical assembly techniques; characterization methods for supraparticulate systems; applications of supraparticles. Modelling of molecular materials, molecular assembly, molecular dynamics Specific topics in molecular nanoscience 	
6	Learning objectives and skills	 to explain the fundamental chemical and physical properties of nano-scale materials to distinguish and to compare some properties, structure and applications of different nanomaterials to describe and to evaluate the major concepts in supramolecular chemistry, molecular machines, self-assembly and supraparticules, or carbon allotropes and transfer this knowledge to related topics to judge on potential nano-analytical tools with optimized image contrast 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202	
10	Method of examination	Oral O30 (PL): Oral examination, 30 Minutes	
11	Grading procedure	Oral (100%) (Please note: The grade of module "Molecular Nanoscience II" is only a part of the total grade of module package "Molecular Nanoscience"!)	
12	Module frequency	only in summer semester	

13	Workload in clock hours	Contact hours: 180 h Independent study: 270 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	E. Katz: Molecular and Supramolecular Information Processing", Wiley-VCH D. Attwood: Soft x-rays and extreme ultraviolet radiation" (Cambridge); C. Jacobsen: X-ray microscopy" (Cambridge) L. Reimer: Scanning electron Microscopy (Springer, 2020) and others

1	Module name 63073	Molecular Nanoscience - Lab	10 ECTS
2	Courses / lectures	 Praktikum: Molecular Nanoscience - Lab (7 SWS) Attendance in lab course is compulsory! Attendance at safety instruction is compulsory! Attendance in winter or summer term possible! A valid laboratory insurance is mandatory for participate course - see: www.laborversicherung.de 	pation in the
3	Lecturers	Prof. Dr. Rainer Fink Prof. Dr. Julien Bachmann Prof. Dr. Andreas Hirsch	

4	Module coordinator	Prof. Dr. Rainer Fink	
5	Contents	Research practicals in selected modern fields of molecular nanoscience: Topics according to an annually updated list of experiments, which includes molecule or nanoparticle synthesis, spectroscopic and microscopic analysis, thin-film and device preparation, characterization	
6	Learning objectives and skills	 to use their theoretical and practical background to plan and perform advanced research experiments under supervision of experienced scientists to synthesize nanoscaled materials to interprete spectroscopic or microscopic data taking advantage of modern research tools to provide a state-of-the-art documentation and discussion of their experimental results to present, communicate and discuss scientific results in scientific english. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Studienrichtung Molecular Nanoscience Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	Lecture notes from the various course lectures in Molecular Nanoscience" (see module descriptions "Molecular Nanoscience-I" and "Molecular Nanoscience-II")	

Advances in bioorganic and bio-inorganic chemistry

1	Module name 46511	Advanced Bio-Organic and Bio-Inorganic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (2 SWS) Seminar: Seminar Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (1 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Norbert Jux Prof. Dr. Nicolai Burzlaff	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	 The students are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine: porphyrins and heme enzymes PDT electron transfer (cofactors, ferredoxins, inner sphere mechanism, outer sphere mechanism, Marcus theory) photosynthesis copper containing proteins and enzymes nitrogenases and other Mo containing enzymes drugs based on Pt, Ru, Au and As biominerals implant materials and technology 	
6	Learning objectives and skills	The students are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	S Contact hours: 45 h	

		Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	W. Kaim, B. Schwederski, A. Klein, Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life, 2. Edition, John Wiley & Sons, Ltd, 2013

1	Module name 46514	Bio-Organic and Bio-Inorganic Chemistry - Lab	5 ECTS
2	Courses / lectures	Praktikum: Bio-Organic & Bio-Inorganic LAB (7 SWS) Please note: Attendance at lab course is compulsory!	5 ECTS
3	Lecturers	Prof. Dr. Norbert Jux Prof. Dr. Ingrid Span Prof. Dr. Petra Imhof Prof. Dr. Nicolai Burzlaff Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	The students deepen their knowledge in special topics of bioorganic and bioinorganic chemistry and nanomedicine that are in the research focus of the involved research groups of the department depending on their own choice perform practical studies and small research projects regarding topics of the preparative, mechanistic or more biological bioorganic and bioinorganic chemistry and nanomedicine in an advanced level	
6	Learning objectives and skills	The students	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 • Lab module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total!)	
10	Method of examination	Practical achievement Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	

1	Module name 46516	Medical Life Sciences Metallic nanoparticles in medicine	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Ingrid Span	
5	Contents	To lay the foundation for working in molecular disease research, the MedLife module provides knowledge in genetics, molecular biology, cell biology, nucleic acid and protein chemistry, enzymology, nuclear and nanomedicine.	
6	Learning objectives and skills	The students explain the basic principles and techniques of the various fields of research engage with scientific texts will be able to apply the essential principles to practical biological and chemical problems	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total)* • Lecture module as Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) *If students want to take the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" as a whole, they have to choose 3 out of the 4 lecture modules, the laboratory module "Bio-Organic and Bio-Inorganic Chemistry - Lab" (No.46514) must be taken compulsorily!	
10	Method of examination	Oral - O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46515	Sensory Sciences Lab	5 ECTS
2	Courses / lectures	Projektseminar: Sensory Sciences Lab (for MSc Chemistry und MSc Molecular Science) (2 SWS) Please note: The module "Sensory Sciences Lab" is an interdisc module with a limited number of participants: only 1 from MSc Chemistry or MSc Molecular Science will admitted! Students from Molecular LIFE Science will admitted! Students from Molecular LIFE Science will priority! The module "Sensory Sciences Lab" can be taken a the compulsory elective module "Advances in Bio-C Bio-Inorganic Chemistry" (20 ECTS in total, all mod be graded) or as an elective module (5 ECTS, ungr	O students be II be given as part of Organic and ules have to
3	Lecturers	Prof. Dr. Danijela Gregurec Prof. Dr. Tim Rohe Prof. Dr. Jessica Freiherr Dr. Helene Loos	

4	Module coordinator	Prof. Dr. Andrea Büttner	
5	Contents	The field of Sensory Sciences investigates how animals and humans sense, neurally process and perceive their environment. This research-oriented interdisciplinary course provides important concepts, theories, and methods of the Sensory Sciences to advanced students from three disciplines (Psychology, Medicine and Chemistry - Molecular Science). These topics comprise human auditory, visual and olfactory perception, chemocommunication, psychophysics, neuroimaging, molecular sensory receptors and technologies for neuromodulation (among others). The students apply these concepts in interdisciplinary teams to develop, conduct and analyse their own small research project. While engaging in an interdisciplinary collaboration and exchange, students can share and deepen their discipline-specific perspectives and competencies as well as learn about concepts and methods from other disciplines.	
6	Learning objectives and skills	The students are able to explain fundamental concepts and methods of the Sensory Sciences develop an own (limited) research question and design an empirical data collection to investigate the research question collect data in the laboratory and analyse the data present their findings in oral and written form communicate effectively in interdisciplinary teamsunderstand and reflect the challenges and chances of interdisciplinary collaborations.	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2;3	

9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202	
10	Method of examination	Seminar achievement Written project report (20-30 pages)	
11	Grading procedure	Seminar achievement (100%) 100% project report • Part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" in M.Sc. Molecular Science (5 ECTS, graded; PO 2020) • elective module (5 ECTS, ungraded; PO 2020)	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 30 h Independent study: 120 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Literature will be announced in the first session Scripts and materials will be provided via StudOn! 	

1	Module name 46512	Special Aspects in Bio-Organic Chemistry	5 ECTS
		Vorlesung: Special Aspects in Bio-Organic Chemistry (2 SWS)	5 ECTS
2	Courses / lectures	Übung: Special Aspects in Bio-Organic Chemistry: Seminar (1 SWS)	
		Please note: • Seminar: time and place by agreement!	
3	Lecturers	Prof. Dr. Petra Imhof Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	The students • learn about: nucleic acid structure, synthesis, and reactivity; nucleic acid modifications; nucleic acid-protein interactions, nucleic acid-based drugs, nucleic acid biochemistry, modelling of nucleic acids	
6	Learning objectives and skills	The students - can explain, apply and reflect upon the theories, terminology, specialities, boundaries and different school of bioorganic and bioinorganic chemistry critically and in depth	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	A. Bloomfield, D.M. Crothers, I. Tinoco, Jr. "Nucleic Acids: structures, properties, and functions", University Science Books, Sausalito, CA, USA 1999	

- D. L. van Vranken, G. A. Weiss, Introduction to Bioorganic Chemistry and Chemical Biology, 1. Aufl., Garland Publishers 2012, ISBN: 978-0815342144
- W. Saenger, Principles of Nucleic Acid Structure", Springer-Verlag New York Inc. 1984
- T. Schlick: "Molecular modelling and simulation: An interdisciplinary guide", Springer New York Dordrecht Heidelberg London 2nd ed. 2010
- A. Vologodskii: "Biophysics of DNA", Cambridge University press, 2015, DOI 10.1017/CBO9781139542371

Advances in homogeneous catalysis

1	Module name 46524	Homogeneous Catalysis - Lab	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester. • Attendance in lab course is compulsory! • Attendance at safety instructions is compulsory! • A valid laboratory insurance is mandatory for particulation of the course is compulsory.	pation in the
3	Lecturers	Prof. Dr. Sjoerd Harder, etc.	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	Students Students get in touch with modern research topics in the field of homogenous catalysis & research tools or advanced spectroscopic tools (Practical work in one of the involved research groups!) deepens her/his knowledge in special topics of homogeneous catalysis that are in the research focus of the involved research groups will be trained in the practical aspects of advanced homogeneous catalysis	
6	Learning objectives and skills	get in touch with modern research topics in the field of homogenous catalysis & research tools or advanced spectroscopic tools manage the preparation and full characterization of catalysts learn methods to evaluate catalyst performance in a series of catalytic reactions and are able to discuss scope and relevance	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Lab module within the Compulsory Elective Module (20 ECTS)! Lab module as part of the elective module (5 ECTS, not graded)!	
10	Method of examination	Practical achievement Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	

1	Module name 46521	Organometallic Catalysis	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Sjoerd Harder, Prof. Dr. Romano Dorta	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 Harder: Introduction and basic principles of advanced catalysis, homogeneous vs. heterogeneous catalysis, mechanisms of substrate activation (C=C, C=O), control by ligand and metal choice, classical catalytic cycles: Wilkinson catalyst, C-C cross coupling, hydroformylation, polymerization catalysis, catalysis with main group metals. Dorta: Review of basic organometallic mechanisms & reactivity (elementary reactions, isolobality); Hydrogenation of alkenes and aromatics; CC cross coupling; Atom-economical CC formation (alkene hydroformylation & hydrocyanation, alkene oligomerization, methanol carbonylation); Synthetic fuels form CO (methanol, MTG, Fischer-Tropsch); Alkene metathesis; Selective CO bond formation (epoxidations and Wacker-type oxidations) 	
6	Learning objectives and skills	 The student understands basic principles and underlying reaction mechanisms in organometallic catalysis is able to outline reaction mechanisms and catalytic cycles knows the tools for the characterization of catalysts and how to evaluate catalyst performance acquires knowledge on topics of current interest and recent breakthroughs in organometallic catalysis 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective module "Advances in Homogenous Catalysis" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, ungraded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h	

		Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 Early Main Group Metal Catalysis – Concepts and Reactions, Sjoerd Harder, Wiley-VCH, 2020 Metal-catalysis in Industrial Organic Processes, Gian Paolo Chiusoli & Peter Maitlis, Royal Society of Chemistry, 2019 Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 Homogeneous Catalysis, Parshall & Ittel, 2nd ed., Wiley Interscience 1992 	

1	Module name 46522	Organocatalysis and Catalytic Reactions in Water	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Svetlana Tsogoeva, Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 History and basic principles of organocatalysis, Different types of organocatalysts, Enantioselective organocatalysis, Domino reactions and important examples. Catalytic reactions in water: Important reactions in biological chemistry, Metabolism of biomolecules and artificial bioorthogonal reactions. 	
6	Learning objectives and skills	The student	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 Organocatalysis and Catalytic Reactions in Water can be taken • within the Compulsory Elective Module "Advances in Homogenous Catalysis" (20 ECTS in total) • as Elective Module (5 ECTS, not graded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46523	Small Molecule Activation	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Karsten Meyer	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 Lecture: Inorganic coordination chemistry, catalytic and electrocatalytic transformation of abundant small molecules N2, O2, CO or CO2, H2O and NH3, into value-added commodities, fine-chemicals, ac-tive pharmaceutical ingredients and polymers. Selective oxidation reactions. CO2 reduction to CO and subsequent Fischer-Tropsch catalysis for the production of carbon-based fuels. Reductive activation of atmospheric N2 to NH3 and the production of H2 from H2O splitting catalysis. Applications, e.g. fuel-cell technologies for carbon-free energy production. 	
6	Learning objectives and skills	The student	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective Module (20 ECTS in total!) • Lecture module as part of the Elective Module (5 ECTS, not graded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

Advances in Interface Research and Catalysis A

1	Module name 46533	Heterogenous Catalysis and Kinetics	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda
5	Contents	 Syllabus: Concepts in heterogeneous catalysis: definition of terms, industrial processes Characterization methods for real catalysts (in-situ and operando methods, TEM, SEM, XRD, EXAFS, XANES, XPS, SIMS, DRIFTS, Raman, TPR, etc.) Surface Reaction Dynamics: dynamics of adsorption, reaction, desorption, molecular beam experiments, laser spectroscopies Elementary Kinetics: microkinetics, transition-state theory, relaxation kinetics, rate-determining step; microkinetic experiments, TAP, SSITKA, etc. Model Catalysis: growth processes, preparation and characterization of model catalysts; kinetics on nanostructured surfaces Energy-related model catalysis (incl. examples from current research)
6	Learning objectives and skills	Students acquire the professional competence in heterogeneous catalysis and respective topics obtain advanced knowledge in different experimental or theoretical models, their application to current problems, the corresponding data evaluation and interpretation using current research examples get familiar with various modern experimental techniques and are able to apply them in a targeted manner
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective Module "Advances in Interface Science & Catalysis" in M. Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As part of the Elective Module in M.Sc. Chemistry/MSc. Molecular Science (5 ECTS, not graded)
10	Method of examination	Oral (20 minutes)
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	only in summer semester

13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	see lecture notes

1	Module name 46534	Interfaces and Catalysis - Lab	5 ECTS
2	Courses / lectures	 Praktikum: Interface & Catalysis LAB (7 SWS) Attendance in lab course is compulsory! Please check lab instructions (contact lab supervisor) Laboratory insurance is mandatory for participation course - see: www.laborversicherung.de 	,
3	Lecturers	Prof. Dr. Jörg Libuda Prof. Dr. Julien Bachmann Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Jörg Libuda		
5	Contents	Practical introduction to state-of-the-art research in the fields of surface science, interface science, heterogeneous catalysis, electrocatalysis or materials characterization. Guided work on a current research project in a research group. Research topics may cover spectroscopy at surfaces, microscopy at surfaces, in-situ or operando spectroscopy, characterization of catalytic materials, in-situ methods in electrocatalysis, preparation and characterization of nanomaterials, modelling and simulation of interfaces and nanomaterials or similar. Practical laboratory experience to introduce state-of-the-art experimental tools in surface and catalysis research, among them: • Electron spectroscopies • Vibrational spectroscopies • Wicroscopy at interfaces • Other characterization methods for surfaces / interfaces • In-situ and operando spectroscopy and microscopy • Characterization of nanomaterials • Electrochemical in-situ characterization • Photochemical / photoelectrochemical in-situ characterization • Modelling on processes at interfaces		
6	Learning objectives and skills	 The students get familiar with the current state-of-knowledge for a specific research topic. apply fundamental knowledge of physical chemistry to a specific research topic. understand and test model-like descriptions for complex physicochemical problems. operate complex state-of-the-art instrumentation. get in contact with development of new methodologies to answer open questions in interface science and catalysis. analyze data with state-of-the-art methodologies. record, document, and analyze research data in in appropriate form. present and discuss experimental results and develop interpretations. 		

		present own results in written form and scientific style English language.
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 • Obligatory lab course module (5 ECTS) within the Compulsory elective module Interfaces and Catalysis within the degree programmes M.Sc. Chemistry or M.Sc. Molecular Science
10	Method of examination	Practical achievement PL: Graded Lab Protocol of 30 - 50 pages (plus raw data documentation) Please note: Module examination organized by supervising group!
11	Grading procedure	Practical achievement (100%)
12	Module frequency	every semester
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	Will be provided by the supervising research group

1	Module name 46532	Surface and Interface Science	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. DrIng. Marcus Bär	

4	Module coordinator	Prof. Dr. Jörg Libuda		
5	Contents	 Vacuum and pressure measurement (pumps, pressure and flow ranges) Lab-based and synchrotron-based light sources (principles, optics, insertion devices, etc.) Theory of photoemission and electronic structure XPS (elemental / chemical sensitivity, cross sections, quantification, examples) UPS (gas phase, adsorbates, 2D band structures, 3D band structures, orbital tomography) IPES (probing of unoccupied states, energy level alignment determination) HAXPES (depth-resolved photoemission measurements, examples) PEEM (spatially-resolved photoemission measurements, examples) NEXAFS (principle and examples) XES & RIXS (principle and examples) Structure of surfaces/ diffraction at surfaces (LEED, definitions and examples) X-ray spectroscopy based materials research on energy conversion devices (examples from current research) 		
6	Learning objectives and skills	 students understand the principles of photoemission variants and their applications can judge the quality of data evaluation and its pitfalls can deliberately select an X-ray spectroscopic analysis method to address given scientific question and are able to evaluate the collected data 		
7	Prerequisites	None		
8	Integration in curriculum	semester: 2;3		
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Interfaces and Catalysis" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)		
10	Method of examination	Oral (20 minutes)		
11	Grading procedure	Oral (100%)		

		As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg Ertl, Küppers: Low Energy Electrons and Surface Chemistry" VCH Weinheim D. Attwood, Soft X-rays and Extreme Ultraviolett Radiation, Cambridge University Press, 1999 D. Briggs, M.P. Seah: PracticalSurface Analysis: AugerandX-Ray PhotoelectronSpectroscopy, Wiley, 1996 M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. A. Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989 	

1	Module name 46531	Theory of Catalytic Processes	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Theory of Catalytic Processes (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Jörg Libuda
5	Contents	The module focuses on physical, chemical or technological aspects of modification, manipulation and characterization of interfaces. These aspects relate to the research of ideal model systems (surfaces and adsorbates on single crystal surfaces) or real systems, in which the interface plays a crucial role for the respective properties. In all cases, the local electronic and chemical interactions at the interface affect the geometric structure (e.g. adsorption geometry) and consequently the chemical and physical properties.
6	Learning objectives and skills	deepen their knowledge in experimental methods and theoretical aspects to describe and characterize interface phenomena are able to perform experiments independently and to analyse the data are familiar with the model-type description of experimental data
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module Advances in Interfaces and Catalysis A (IntCat-3A) (20 ECTS in total)! • Module can also be taken as part of the elective module (5 ECTS, not graded)! Please note: MSc Molecular NANO Science students have to attend the module
		"Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann!
10	Method of examination	"Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof.
10	Method of examination Grading procedure	"Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann!
		"Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann! Oral (20 minutes) Oral (100%)

		Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

Advances in energy materials

1	Module name 46542	Advanced Electrochemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Electrochemistry (3 SWS)	
3	Lecturers	Dr. Christian Ehli	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	 Comprehensive survey of the fundamentals for electrode processes (thermodynamics and kinetics) Introduction to electrochemical techniques (e.g. cyclic voltammetry, rotating disk voltammetry, differential pulse voltammetry, spectroelectrochemistry, electrochemical impedance spectroscopy) Applications of electrochemistry (e.g. corrosion prevention, batteries) Seminars will be based on the discussion of practical aspects and electrochemical exercises 	
6	Learning objectives and skills	Plan and perform own electrochemical experiments characterize electroactive materials by common electrochemical methods analyze, interpret and discuss electrochemical experimental results discuss and evaluate current electrochemical publications	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	

16	Bibliography	 Allen J. Bard, Larry R. Faulkner: "Electrochemical Methods: Fundamentals and Applications", John Wiley & Sons, New York, NY Carl H. Hamann, Andrew Hamnett, Wolf Vielstich: "Electrochemistry", Wiley-VCH, Weinheim
		For further literature, please see the current list on studon.

1	Module name 46544	Energy Materials - Lab	5 ECTS
2	Courses / lectures	 Praktikum: Energy Materials - LAB (7 SWS) Attendance at lab course is compulsory! Attendance at safety instructions is compulsory! A valid laboratory insurance is mandatory for particilab course - see: www.laborversicherung.de 	5 ECTS
3	Lecturers	Dr. Christian Ehli Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi
5	Contents	 Practical introduction to electrochemical techniques Guided work on the characterization of electroactive materials Attempts to solve independently a scientific problem Documentation of experimental results
6	Learning objectives and skills	Students
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • The module can be taken as part of the Elective Module (5 ECTS, not graded)!
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol of 30 - 50 pages (plus raw data documentation)
11	Grading procedure	Practical achievement (100%)
12	Module frequency	every semester
13	Workload in clock hours	Contact hours: 105 h Independent study: 45 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46541	Semiconductor Materials for Energy Applications	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi
5	Contents	 Fundamentals of semiconductors: Crystal structure, Electronic structure, Electrical transport, Interaction with light Semiconductor devices: Tunnelling, The pn junction, The transistor Photovoltaics: Principles, Types of solar cells The interface to a solution: Charged electrolytic interfaces, Electrocatalysis and photoelectrocatalysis
6	Learning objectives and skills	The students
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • within the Compulsory Elective Module "Advances in Energy Materials" in M. Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total) • part of the Elective Module in M. Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)
10	Method of examination	Oral O20 (PL): Oral examination, 20 minutes
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46543	Solar Energy Conversion	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Solar Energy Conversion (3 SWS)	
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi
5	Contents	 Demand and supply of energy Solar cells: Silicon solar cells, dye-sensitized solar cells, organic solar cells, perovskite solar cells, singlet fission Fundamentals of Electron Transfer Photosynthesis: natural photosynthesis, artificial photosynthesis
6	Learning objectives and skills	 The students are familiar with the fundamentals and modern applications in solar energy research and applications understand design principles in solar energy devices and can transfer this knowledge to related topics can present, communicate and discuss scientific results with experts in English.
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" MSc Chemistry and Molecular Science (20 ECTS in total!) • Module can be taken as part of the Elective Module, too (5 ECTS, not graded)!
10	Method of examination	Written examination (60 minutes)
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

Quantum chemistry

1	Module name 46508	Quantum Chemistry laboratory	10 ECTS
	Courses / lectures	Praktikum: Training in Applied Computational Chemistry (5 SWS)	
		Vorlesung mit Übung: Quantum Chemistry - Lab / Scientific Programming (10 SWS)	
2		 Attendance in lab course is compulsory! Internship in one of the Theoretical Chemistry group Görling, Imhof, B. Meyer, Zahn), time and place by Training has to be taken only once: in winter or in statement (time and place by agreement)! Module starts only in winter term (Duration: 2 seme 	agreement! ummer term
3	Lecturers	Prof. Dr. Andreas Görling Prof. Dr. Dirk Zahn Prof. Dr. Petra Imhof Prof. Dr. Bernd Meyer Dr. Christian Neiß	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	 Operating system Linux for high-performance computing (HPC) Scientific programming in Fortran and Python Using numerical and mathematical libraries/modules Introduction to parallel computing Exercises Programming project Training in applied computational chemistry 	
6	Learning objectives and skills	Students get familiar with Linux as operating system for HPC are able to create computer programs for scientific purposes can use numerical and mathematical libraries/modules in home-made programs obtain knowledge about basic parallelization paradigms apply quantum chemical methods to scientific questions under guidance	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Quantum Chemistry Master of Science Molecular Science 20202 Lab module within the Core module "Quantum Chemistry" in M. Sc. Chemistry Lab module within the Compulsory Elective Module in M.Sc. Chemistry (if not chosen as Core module) or M. Sc. Molecular Science	
10	Method of examination	Practical achievement Graded Lab Protocol: Successful implementation of the programming project (working program) including Lab report (ca. 5 pages)	

11	Grading procedure	Practical achievement (100%) 100% Graded Computer Program	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 225 h Independent study: 75 h	
14	Module duration	2 semester	
15	Teaching and examination language	english	
16	Bibliography	 Stephen J. Chapman: Fortran for Scientists and Engineers, McGraw Hill 2017 (4th ed.) Bernd Klein: Einführung in Python 3, Hanser 2017 (3rd ed.) Stefan Gerlach: Computerphysik, Springer Spektrum 2019 (2nd ed.) 	

1	Module name 46507	Quantum Chemistry	10 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry 2 (3 SWS, SoSe 2024) Vorlesung mit Übung: Quantum Chemistry 1 (3 SWS, WiSe 2023)	
		Please note: • The core module "Quantum Chemistry" starts only i term!	n winter
3	Lecturers	Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	 Introduction to modern methods and current research issues in the field of quantum and computer chemistry Hartree-Fock, DFT, Many Body Perturbation Theory Configuration Interaction, Second Quantization, Coupled Cluster TD-HF, TD-DFT, RPA 	
6	Learning objectives and skills	Students obtain sound knowledge in basic and advanced methods of quantum chemistry are able to solve mathematical problems occurring in quantum chemistry are able to understand and assess scientific reports in the field of quantum chemistry	
7	Prerequisites	Required Qualifications: good knowledge of basic quantum mechanics: axioms of QM, application to simple systems (particle in a box, harmonic oscillator, rigid rotator) good knowledge in mathematics: differential calculus of functions of several variables, basic linear algebra	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Quantum Chemistry Master of Science Molecular Science 20202 Lecture module within the Core module "Quantum Chemistry" in M. Sc. Chemistry Lecture module within the Compulsory Elective Module in M.Sc. Chemistry (if not chosen as Core module) or M. Sc. Molecular Science	
10	Method of examination	Oral (30 minutes) O30 (PL): Oral Examination (30 minutes)	
11	Grading procedure	Oral (100%)	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 90 h Independent study: 210 h	
14	Module duration	2 semester	

15	Teaching and examination language	english	
16	Bibliography	 Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996 Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.) Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.) 	

Research module

1	Module name 63069	Research Lab - Drug Discovery	15 ECTS
2	Courses / lectures	Praktikum: Research Module - Drug Discovery (23 SWS)	15 ECTS
3	Lecturers	Prof. Dr. Peter Gmeiner	

4	Module coordinator	Prof. Dr. Peter Gmeiner	
5	Contents	Research lab project full time (about 6 weeks: 21SWS LAB/2SWS Seminar) in a work group of the students choice at a research group in Medicinal Chemistry, Food Chemistry, or the disciplines of Biology involved in teaching at the Department of Chemistry and Pharmacy	
6	Learning objectives and skills	 are provided with the up-to-date practical and operative know-how suitable for future scientific and/or applied work in research institutes, pharmaceutical/food industry, medical care laboratories, bio-technological, bio-analytical and environmental branches get an advanced theoretical background and an overview of emerging trends in life sciences (chemistry, biology and medicine) look at living systems through the lens of basic chemical principles are prepared to work in interdisciplinary environment and participate in national and international development of forefront fields such as translational medicine. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 3	
9	Module compatibility	Research module Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement pÜL: Graded Lab Protocol (approx. 20 pages plus raw data documentation)	
11	Grading procedure	Practical achievement (100%)	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 345 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 63074	Research Lab - Molecular Nanoscience	15 ECTS
2	Courses / lectures	Praktikum: Research Module - Molecular Nanoscience (23 SWS)	15 ECTS
3	Lecturers	Prof. Dr. Karl Mandel Prof. Dr. Henry Dube Prof. Dr. Julien Bachmann Prof. Dr. Dirk Zahn Prof. Dr. Rainer Fink Prof. Dr. Franziska Gröhn Dr. Ryan Crisp Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Rainer Fink	
5	Contents	research lab project (23 SWS) fulltime in one of the research groups of Molecular Nanoscience	
6	Learning objectives and skills	 are provided with the up-to-date practical and operative know-how suitable for future scientific and/or applied work in research institutes and nanoscience-labs get an advanced theoretical background and an overview of emerging trends in nano sciences look at living systems through the lens of basic chemical principles are prepared to work in interdisciplinary environment and participate in national and international development of forefront fields. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 3	
9	Module compatibility	Research module Master of Science Molecular Science 20202	
10	Method of examination	Practical achievement	
11	Grading procedure	Practical achievement (100%) pÜL: Graded Lab Protocol (approx. 20 pages plus raw data documentation)	
12	Module frequency	every semester	
13	Workload in clock hours	Contact hours: 345 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

Elective modules

1	Module name 46511	Advanced Bio-Organic and Bio-Inorganic Chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (2 SWS) Seminar: Seminar Advanced Bioinorganic Chemistry, Metalloenzymes and Metals in Medicine (1 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Norbert Jux Prof. Dr. Nicolai Burzlaff	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	 The students are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine: porphyrins and heme enzymes PDT electron transfer (cofactors, ferredoxins, inner sphere mechanism, outer sphere mechanism, Marcus theory) photosynthesis copper containing proteins and enzymes nitrogenases and other Mo containing enzymes drugs based on Pt, Ru, Au and As biominerals implant materials and technology 	
6	Learning objectives and skills	The students • are introduced into recent activities and achievements in the fields of bioorganic and bioinorganic chemistry and metals in medicine	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h	

		Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	W. Kaim, B. Schwederski, A. Klein, Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life, 2. Edition, John Wiley & Sons, Ltd, 2013

1	Module name 46542	Advanced Electrochemistry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Advanced Electrochemistry (3 SWS)	
3	Lecturers	Dr. Christian Ehli	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	 Comprehensive survey of the fundamentals for electrode processes (thermodynamics and kinetics) Introduction to electrochemical techniques (e.g. cyclic voltammetry, rotating disk voltammetry, differential pulse voltammetry, spectroelectrochemistry, electrochemical impedance spectroscopy) Applications of electrochemistry (e.g. corrosion prevention, batteries) Seminars will be based on the discussion of practical aspects and electrochemical exercises 	
6	Learning objectives and skills	Students plan and perform own electrochemical experiments characterize electroactive materials by common electrochemical methods analyze, interpret and discuss electrochemical experimental results discuss and evaluate current electrochemical publications	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded)	
9	Module compatibility Method of examination	20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc.	
		 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) 	
10	Method of examination	20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) Written examination (60 minutes) Written examination (100%)	
10	Method of examination Grading procedure	20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) Written examination (60 minutes) Written examination (100%) As part of the elective module: 0% - pass/fail	
10 11 12	Method of examination Grading procedure Module frequency	20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" in M.Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total)! • as part of the Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) Written examination (60 minutes) Written examination (100%) As part of the elective module: 0% - pass/fail only in winter semester Contact hours: 45 h	

16	Bibliography	 Allen J. Bard, Larry R. Faulkner: "Electrochemical Methods: Fundamentals and Applications", John Wiley & Sons, New York, NY Carl H. Hamann, Andrew Hamnett, Wolf Vielstich: "Electrochemistry", Wiley-VCH, Weinheim
		For further literature, please see the current list on studon.

	1	Module name 46563	Advanced Spectroscopic Techniques	5 ECTS
ĺ	2	Courses / lectures	Vorlesung mit Übung: Advanced Spectroscopy in Organic Chemistry (3 SWS)	5 ECTS
	3	Lecturers	Prof. Dr. Henry Dube Dr. Alexander Scherer	

4	Module coordinator	Prof. Dr. Henry Dube	
5	Contents	Fundamentals of spectroscopy in organic chemistry will be reviewed. More in-depth methods of molecular spectroscopy in organic chemistry are covered. Advanced methods of NMR-spectroscopy are also covered, such as NMR-spectroscopy of various nuclei (e.g., 1H, 13C, 19F, 31P). Two-dimensional methods of NMR-spectroscopy using scalar spin-spin couplings (e.g., HSQC, HMBC) are discussed. Furthermore, NMR-spectroscopic methods relying on interactions between coupling nuclear dipoles, which are transmitted directly through space are covered (e.g., NOESY). In addition, other optical spectroscopic methods will be reviewed and discussed in more depth (e.g., UV/Vis-, CD-, IR-spectroscopy and Mass spectrometry).	
6	Learning objectives and skills	 The students master the reliable use and gain an understanding of spectroscopic methods in organic chemistry, which are used to elucidate organic molecules; are able to characterize unknown molecules and to determine their structure as well as their dynamics and interactions (the correlations between the spectroscopic results and the characteristics of the molecules should become understandable and comprehensible); discuss practical examples of spectroscopic results and the related correlations in the exercises and practice the structure elucidation on examples. 	
7	Prerequisites	Basic knowledge of spectroscopy is recommended.	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202	
10	Method of examination	Tutorial achievement SL: practical exercise performance (15- 20 minutes), non-graded	
11	Grading procedure	Tutorial achievement (0%) pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	

15	Teaching and examination language	english
16	Bibliography	 "Basic one- and two-dimensional NMR spectroscopy, edited by H. Friebolin, Wiley-VCH "NMR From spectra to structures edited by T. N. Mitchell, B. Costisella, Springer "Spectroscopic methods in organic chemistry, Edited by D. H. Williams, I. Flemming, McGraw Hill "Modern NMR spectroscopy, Edited by J. K. M. Sanders, E. C. Constable, B. K. Hunter, C. M. Pearce, Oxford

1	Module name 46562	Biological and Synthetic Molecular Switches and Machines	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Henry Dube	

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4	Module coordinator	Prof. Dr. Henry Dube	
5	Contents	Examination of molecular triggers, switches and machines in biology and in synthetic systems as foundation of nanotechnology; working mechanisms; types of systems; design principles; seminal contributions; historical backgrounds are given. The material is ordered in introduction and context, basic principles, triggers, switches, machines, integrated systems, future prospective. The course will be updated to implement the newest developments yearly	
6	Learning objectives and skills	 acquire a fundamental understanding in the working mechanisms and design principles of molecular triggers, switches, and machines. will be able to develop strategies for implementing responsiveness into nanostructured biological or synthetic systems and will be equipped with an exhaustive overview of historical developments and current state of the art in the field by discussing representative examples in depth. will therefore be educated in one of the most prominent fields of modern (bio)chemistry and nanosciences. Compulsory attendance will be necessary. The skills will be appropriate for Masters level and will partially repeat and build on knowhow from supramolecular, biological, and photochemistry as well as on fundamental physical organic chemistry, biochemistry, nanotechnology, and spectroscopy. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 Module compatibility:	
10	Method of examination	Tutorial achievement O20 (SL): 20 min oral examination in the form of a seminar talk presenting the content of a seminal original publication (Non-graded seminar presentation)	
11	Grading procedure	Tutorial achievement (0%) pass/fail	
12	Module frequency	only in summer semester	

13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 "Molecular Switches, edited by Ben L. Feringa, Wiley-VCH "From Non-Covalent Assemblies to Molecular Machines, Edited by Jean-Pierre Sauvage & Pierre Gaspard, Wiley-VCH "Molecular Machines and Motors Recent Advances and Perspectives edited by Alberto Credi, Serena Silvi, Margherita Venturi, Springer 	

1	Module name 46549	Biomolecules and metal ions - evolution, biological functions, and biomedicine	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Ingrid Span	
5	Contents	 Roles of metal ions in biology with focus on photosynthesis & oxygen transport Metalloproteins Metal ions in evolution, extremophile organisms Metal ions in biomedicine (imaging and therapy) Fundamentals of protein crystallography Fundamentals of spectroscopic techniques for characterizing metalloproteins Seminars in form of presenting scientific research articles on metallobiochemistry 	
6	Learning objectives and skills	 can explain the fundamental properties of biomolecules, the occurrence and the role of metals in biological systems, and the chemistry of life. gain a better understanding of the relevance of metals in evolution and the application of metals in biomedicine. get insight into different techniques that can be used to analyse metal-binding biomolecules. are able to transfer the acquired knowledge to solve unrelated scientific problems. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 The module (5 ECTS, not graded) can be taken as part of the elective module (in total 15 ECTS) within the Master programme Chemistry or Molecular Science	
10	Method of examination	Written examination PL: Written exam, 60 minutes, not graded	
11	Grading procedure	Written examination (0%) PL: W60 (0%) pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	

16 Bibliography	 Bioinorganic Chemistry - Inorganic Elements in the Chemistry of Life: An Introduction and Guide (Second Edition 2013) Wolfgang Kaim, Brigitte Schwederski, Axel Klein Bioanorganische Chemie - Metalloproteine, Methoden und Konzepte (1. Auflage August 2017) Sonja Herres-Pawlis, Peter Klüfers
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1	Module name 46536	Lebensmittelchemie Food chemistry	5 ECTS
2	Courses / lectures	Vorlesung: Chemie und Technologie der Lebensmittel, I- Vitamine (1 SWS) Vorlesung: Chemie und Technologie der Lebensmittel, II - Lipide (1 SWS)	
		Seminar: Bioanalytik-Seminar für Lebensmittelchemiker und Molecular Science (1 SWS) Bitte beachten: • Die Vorlesung Lebensmittelchemie hat je nach Semester unterschiedliche inhaltliche Schwerpunkte (Lebensmittelchemie I-XI)! • Die Studierenden müssen 2 Vorlesungen (im Umfang von 2 SWS) und das Seminar (1 SWS) besuchen!	
3	Lecturers	Prof. Dr. Monika Pischetsrieder	

4	Module coordinator	Prof. Dr. Monika Pischetsrieder	
5	Contents	 Es werden toxikologisch, technologisch und physiologisch relevante Inhaltsstoffe von Lebensmitteln ausführlich vorgestellt und diskutiert. Ausgehend von den grundlegenden Kenntnissen der organischen Chemie werden Reaktionsmechanismen, die während der Prozessierung oder Entstehung von Lebensmitteln ablaufen, erläutert. Ausgehend von grundlegenden Kenntnissen der analytischen Chemie werden die wichtigsten weiterführenden und aktuellen instrumentellanalytischen und bioanalytischen Analysenmethoden besprochen. Die Studierenden erarbeiten sich die Sachkompetenz zur theoretischen Beurteilung und praktischen Handhabung wichtiger Fragestellungen der Lebensmittelchemie sind in der Lage, die wichtigsten relevanten Arbeitstechniken aus dem Gebiet der Lebensmittelchemie und analytik und eines anderen Gebiets der Lebensmittelwissenschaften selbständig anzuwenden können die wesentlichen Prinzipien der Lebensmittelanalytik auf praktische Probleme anwenden und kritisch reflektieren. 	
6	Learning objectives and skills		
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202	
10	Method of examination	Seminar achievement (20 minutes) (SL): Seminarvortrag, 30 Minuten, unbenotet	
11	Grading procedure	Seminar achievement (0%)	

		SL (0%): unbenotet bestanden/nicht bestanden
12	Module frequency	every semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	german
16	Bibliography	Wird von den Dozenten aktualisiert zur Verfügung gestellt.

1	Module name 46533	Heterogenous Catalysis and Kinetics	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	 Syllabus: Concepts in heterogeneous catalysis: definition of terms, industrial processes Characterization methods for real catalysts (in-situ and operando methods, TEM, SEM, XRD, EXAFS, XANES, XPS, SIMS, DRIFTS, Raman, TPR, etc.) Surface Reaction Dynamics: dynamics of adsorption, reaction, desorption, molecular beam experiments, laser spectroscopies Elementary Kinetics: microkinetics, transition-state theory, relaxation kinetics, rate-determining step; microkinetic experiments, TAP, SSITKA, etc. Model Catalysis: growth processes, preparation and characterization of model catalysts; kinetics on nanostructured surfaces Energy-related model catalysis (incl. examples from current research) 	
6	Learning objectives and skills	 Students acquire the professional competence in heterogeneous catalysis and respective topics obtain advanced knowledge in different experimental or theoretical models, their application to current problems, the corresponding data evaluation and interpretation using current research examples get familiar with various modern experimental techniques and are able to apply them in a targeted manner 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective Module "Advances in Interface Science & Catalysis" in M. Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As part of the Elective Module in M.Sc. Chemistry/MSc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	

13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	see lecture notes

1	Module name 46567	HZB Photon School	5 ECTS
2	Courses / lectures	Vorlesung: HZB Photon School - Advanced synchrotron-based photon science methods (2 SWS)	2,5 ECTS
		Praktikum: HZB Photon School - Practical training at synchrotron facility (lab course) (2 SWS)	2,5 ECTS
3	Lecturers	Prof. DrIng. Marcus Bär	

4	Module coordinator	Prof. DrIng. Marcus Bär	
5	Contents	 Students will be introduced to the functionality of a synchrotron (storage ring-based light sources, x-ray optics and beamlines for synchrotron radiation experiments), fundamentals of the interaction of x-rays with matter, and laboratory-based characterization methods that probe the molecular structure, function, and dynamics of complex material systems such as x-ray absorption spectroscopy, x-ray photoemission spectroscopy, magnetic spectroscopy and scattering, and x-ray microscopy. The two-week program consists of basic and specialized lectures including exercises, given by experienced teachers who are HZB scientists or BESSY II super-users from HZB partner universities in the first week. The second week is dedicated to hands-on training at several BESSY II experimental stations and X-ray laboratories. Participants will get a vivid experience by joining up to two experiments to work in small groups. The second week's activities will teach participants how to collect, analyse, interpret, and present own data amongst peers and specialists. For further information, visit hz-b.de/photonschool 	
6	Learning objectives and skills	 describe the working principle of a synchrotron-based light source explain basic photon-matter interactions describe the principles of photoemission, diffraction, scattering based analysis method variants, photon-in - photon-out spectroscopic and microscopic techniques and their applications deliberately select an X-ray based analysis method to address given scientific question and are able to evaluate the collected data judge the quality of data evaluation and its pitfalls represent their measurement results are able to write a compelling beamtime proposal 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2;3;4	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202	

		as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes) PL (O20): Oral examination = Poster presentation (20 minutes)	
11	Grading procedure	Oral (0%) Ungraded, module has to be passed only	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 60 h Independent study: 90 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg Ertl, Küppers: Low Energy Electrons and Surface Chemistry", VCH Weinheim D. Attwood, Soft X-rays and Extreme Ultraviolett Radiation, Cambridge University Press, 1999 D. Briggs, M.P. Seah: PracticalSurface Analysis: Augerand X-Ray Photoelectron Spectroscopy, Wiley, 1996 M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag, GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989 	

1	Module name 46516	Medical Life Sciences Metallic nanoparticles in medicine	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Ingrid Span	

4	Module coordinator	Prof. Dr. Ingrid Span	
5	Contents	To lay the foundation for working in molecular disease research, the MedLife module provides knowledge in genetics, molecular biology, cell biology, nucleic acid and protein chemistry, enzymology, nuclear and nanomedicine.	
6	Learning objectives and skills	The students explain the basic principles and techniques of the various fields of research engage with scientific texts will be able to apply the essential principles to practical biological and chemical problems	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total)* • Lecture module as Elective Module in M.Sc. Chemistry or M.Sc. Molecular Science (5 ECTS, not graded) *If students want to take the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" as a whole, they have to choose 3 out of the 4 lecture modules, the laboratory module "Bio-Organic and Bio-Inorganic Chemistry - Lab" (No.46514) must be taken compulsorily!	
10	Method of examination		
11	Grading procedure	As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46521	Organometallic Catalysis	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Sjoerd Harder, Prof. Dr. Romano Dorta	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 Harder: Introduction and basic principles of advanced catalysis, homogeneous vs. heterogeneous catalysis, mechanisms of substrate activation (C=C, C=O), control by ligand and metal choice, classical catalytic cycles: Wilkinson catalyst, C-C cross coupling, hydroformylation, polymerization catalysis, catalysis with main group metals. Dorta: Review of basic organometallic mechanisms & reactivity (elementary reactions, isolobality); Hydrogenation of alkenes and aromatics; CC cross coupling; Atom-economical CC formation (alkene hydroformylation & hydrocyanation, alkene oligomerization, methanol carbonylation); Synthetic fuels form CO (methanol, MTG, Fischer-Tropsch); Alkene metathesis; Selective CO bond formation (epoxidations and Wacker-type oxidations) 	
6	Learning objectives and skills	 The student understands basic principles and underlying reaction mechanisms in organometallic catalysis is able to outline reaction mechanisms and catalytic cycles knows the tools for the characterization of catalysts and how to evaluate catalyst performance acquires knowledge on topics of current interest and recent breakthroughs in organometallic catalysis 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective module "Advances in Homogenous Catalysis" in M. Sc. Chemistry or Molecular Science (20 ECTS in total, graded) • As Elective Module in M. Sc. Chemistry or Molecular Science (5 ECTS, ungraded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h	

		Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 Early Main Group Metal Catalysis – Concepts and Reactions, Sjoerd Harder, Wiley-VCH, 2020 Metal-catalysis in Industrial Organic Processes, Gian Paolo Chiusoli & Peter Maitlis, Royal Society of Chemistry, 2019 Fundamentals of Organometallic Catalysis, Dirk Steinborn, Wiley-VCH, 2012 Homogeneous Catalysis, Parshall & Ittel, 2nd ed., Wiley Interscience 1992 	

1	Module name 46564	Modern Methods in Mass Spectrometry	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Modern Methods in Mass Spectrometry (3 SWS)	
3	Lecturers	Prof. Dr. Thomas Drewello	

4	Module coordinator	Prof. Dr. Thomas Drewello	
5	Contents	 Advanced aspects of Soft Ionization methods (MALDI, ESI, related Atmospheric Pressure Ionization methods) Advanced aspects of mass analyzers (FT-ICR, orbitrap, hybrids: multi sector, BEqQ, QToF, QIT-ToF) Ion Activation (CID, BIRD, IRMPD, SID, ECID) Applications (thermochemistry, kinetic method, equilibrium methods, ion/molecule-reactions) "Omics (proteomics, petroleomics, metabolomics) Further applications: 14C dating, accelerator MS, stable isotope MS, MS in space, ICP-MS. Seminars in form of problem solving classes 	
6	Learning objectives and skills	 Students gain insight into the different ion formation processes are able to decide on an ionization method for a given compound class are able to evaluate the use of different mass spectrometers gain understanding of ion activation processes in the gas phase 	
7	Prerequisites	The course builds on the master module "Applied Spectroscopy (Compulsory Module Physical Chemistry)	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 Lecture module as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (0%) pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Jürgen H. Groß: "Mass Spectrometry, a textbook Springer, Heidelberg Edmond De Hoffmann: "Mass Spectrometry, principles and applications, Wiley 	

1	Module name 46558	Modern X-ray Structure Determination	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Modern X-ray structure determination of single crystals (3 SWS)	5 ECTS
3	Lecturers	Dr. Frank Wilhelm Heinemann	

4	Module coordinator	Dr. Frank Wilhelm Heinemann	
5	Contents	 Fundamentals of crystallization and polymorphism Structural description of single crystals, crystal systems, unit cell, symmetry and symmetry elements, space groups Diffraction power of crystals, diffraction conditions, structure factor Generation of X-rays, single crystal diffractometers, detection techniques Structure solution techniques and refinement procedures, software, problems and pitfalls, interpretation of results Anomalous dispersion and absolute structure Graphical representations, use of data bases 	
6	Learning objectives and skills	Students get insight into thermodynamics of crystallization and crystallization techniques get fundamentals of the theory behind crystal structure determination get practice in crystal selection, mounting and measurement set-up get hands-on training in structure solution and refinement using up-to-date software are enabled to interpret and compare results of a single crystal structure determination	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Tutorial achievement (PL): Lab report in manuscript style (max. 2000 words plus raw data), not graded	
11	Grading procedure	Tutorial achievement (0%) not graded: pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	

		 Werner Massa: Kristallstrukturbestimmung. Teubner Studienbücher Chemie, Vieweg und Teubner, 6. Auflage, 2009, ISBN: 3834806498
16	Bibliography	 William Clegg: Crystal Structure Determination. Oxford Chemistry Primers. Oxford University Press, 1998, ISBN: 0198559011
		Further literature will be recommended in the course

1	Module name 46535	Neurotech: Physics and Chemistry of Neuromodulation Technologies	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Danijela Gregurec	

4	Module coordinator	Prof. Dr. Danijela Gregurec	
5	Contents	 Introduction to neuromodulation technologies (definition, history, nervous anatomy, stimulation targets-ion channels, action potential) Imaging and spectroscopic concepts (MRI, EEG, Calcium imaging, electrophysiology) Current tech and principles (Invasive and noninvasive approaches, Deep brain stimulation, Transcranial magnetic stimulation, Pain management, BMI summary) Organic materials and approaches (viral vectors, optogenetics, Chemogenetics (DREED) Micro- and macroscale materials (Mechanical properties and compatibility of neural implants, Electrodes (Utah arrays, Neuralink), Flexible electrodes, Optical fibers) Nanomaterials (Nanomaterial properties leveraged for neuromodulation (Importance of surface chemistry in bio(nano)materials, Quantum confinement and quantum dots, Plasmons and photothermal neuromodulation, Magnetic nanoparticles for magnetothermal and magnetomechanical stimulation 	
6	Learning objectives and skills	 Students will gain the knowledge, skills, and competences to be able to understand biophysical aspects of neuronal signaling and its correlation to cognition and behavior. learn physical foundations, biological concepts, and chemical approaches crucial for materials used in neuromodulation and neurotechnology. apply acquired knowledge to realize design criteria of technology that governs the modulation of neuronal signaling. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 Module compatibility: • Lecture module within the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes) Oral examination (20 minutes, ungraded)	
11	Grading procedure	Oral (0%)	

		not graded - pass/fail
12	Module frequency	only in summer semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	Knotkova, Rasche; Springer-Verlag New York (2015) Textbook of Neuromodulation: Principles, Methods and Clinical Applications Luan et al, Front Neuroeng 7 (2014) Neuromodulation: present and emerging methods Frank, et al, Nat Biotech 37 (2019) Next-generation interfaces for studying neural function Chen et al, Nat Rev Mat 2 (2017) Neural recording and modulation technologies **Literature will be updated with leading peer-reviewed papers during lectures

1	Module name 46545	Organic thin films	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Organic Thin Films (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Rainer Fink	

4	Module coordinator	Prof. Dr. Rainer Fink	
5	Contents	 Molecular interactions and molecular self-organization Influence of molecualr geometry and functionalization Thin film preparation techniques: Langmuir-Blodgett, Self-Assembled Monolaers (SAMs), other solvent-based techniques (e.g. spin-casting, doctor blading, etc.), vacuum sublimation Analytical techniques, in-situ analysis Effect of templates Organic thin film applications Seminars will be based on recent literature 	
6	Learning objectives and skills	Students	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 • lecture module can be taken as part of the Elective Module!	
10	Method of examination	Variable O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Variable (0%) pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46522	Organocatalysis and Catalytic Reactions in Water	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Svetlana Tsogoeva, Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 History and basic principles of organocatalysis, Different types of organocatalysts, Enantioselective organocatalysis, Domino reactions and important examples. Catalytic reactions in water: Important reactions in biological chemistry, Metabolism of biomolecules and artificial bioorthogonal reactions. 	
6	Learning objectives and skills	The student	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 Organocatalysis and Catalytic Reactions in Water can be taken • within the Compulsory Elective Module "Advances in Homogenous Catalysis" (20 ECTS in total) • as Elective Module (5 ECTS, not graded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46559	Quantum Chemistry 1	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry 1 (3 SWS)	
3	Lecturers	Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	Mathematical concepts and current research issues in the field of quantum and computer chemistry Hartree-Fock, DFT	
6	Learning objectives and skills	Students obtain sound knowledge in basic methods of quantum chemistry are able to solve mathematical problems occurring in quantum chemistry are able to understand and assess scientific reports in the field of quantum chemistry	
7	Prerequisites	Strongly recommended Qualifications: • good knowledge of basic quantum mechanics: axioms of QM, application to simple systems (particle in a box, harmonic oscillator, rigid rotator) • good knowledge in mathematics: differential calculus of functions of several variables, linear algebra	
8	Integration in curriculum	semester: 1;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes) O20 (PL): Oral Examination (20 minutes, not graded: pass/fail)	
11	Grading procedure	Oral (0%) not graded: pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996 Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.) Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.) 	

1	Module name 46561	Quantum Chemistry 2	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Andreas Görling	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	 Many-Body Perturbation Theory Configuration Interaction, Second Quantization, Coupled Cluster TD-HF, TD-DFT, RPA 	
6	Learning objectives and skills	Students obtain sound knowledge in advanced methods of quantum chemistry are able to solve mathematical problems occurring in quantum chemistry are able to understand and assess scientific reports in the field of quantum chemistry	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;4	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) if Quantum Chemistry I was already chosen as Elective Module!	
10	Method of examination	Oral (20 minutes) O20 (PL): Oral Examination (20 minutes, not graded: pass/fail)	
11	Grading procedure	Oral (0%) not graded: pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Attila Szabo, Neil S. Ostlund: Modern Quantum Chemistry, Dover 1996 Frank Jensen: Introduction to Computational Chemistry, Wiley 2017 (3rd ed.) Ira N. Levine: Quantum Chemistry, Pearson 2016 (7th ed.) 	

1	Module name 46556	Scientific Programming	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Quantum Chemistry - Lab / Scientific Programming (10 SWS, SoSe 2024) Please note: • Attendance in the lab course is mandatory! • Module starts only in winter term (duration: 2 semes	sters)
3	Lecturers	Prof. Dr. Andreas Görling Dr. Christian Neiß	

4	Module coordinator	Prof. Dr. Andreas Görling	
5	Contents	 Operating system Linux for high-performance computing (HPC) Scientific programming in Fortran and Python Using numerical and mathematical libraries/modules Introduction to parallel computing Exercises Programming project 	
6	Learning objectives and skills	Students • get familiar with Linux as operating system for HPC • are able to create computer programs for scientific purposes • can use numerical and mathematical libraries/modules in home-made programs • obtain knowledge about basic parallelization paradigms	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 as Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded) - please note: the module cannot be combined with the module Quantum Chemistry - Lab	
10	Method of examination	Practical achievement PL: Successful implementation of the programming project (working program), ungraded - module has to be passed	
11	Grading procedure	Practical achievement (0%) not graded: pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 150 h Independent study: 0 h	
14	Module duration	2 semester	
15	Teaching and examination language	english	
16	Bibliography	 Stephen J. Chapman: Fortran for Scientists and Engineers, McGraw Hill 2017 (4th ed.) Bernd Klein: Einführung in Python 3, Hanser 2017 (3rd ed.) 	

Stefan Gerlach: Computerphysik, Springer Spektrum 2019
(2nd ed.)

1	Module name 46541	Semiconductor Materials for Energy Applications	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	 Fundamentals of semiconductors: Crystal structure, Electronic structure, Electrical transport, Interaction with light Semiconductor devices: Tunnelling, The pn junction, The transistor Photovoltaics: Principles, Types of solar cells The interface to a solution: Charged electrolytic interfaces, Electrocatalysis and photoelectrocatalysis 	
6	Learning objectives and skills	The students	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • within the Compulsory Elective Module "Advances in Energy Materials" in M. Sc. Chemistry or M.Sc. Molecular Science (20 ECTS in total) • part of the Elective Module in M. Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral O20 (PL): Oral examination, 20 minutes	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46515	Sensory Sciences Lab	5 ECTS
2	Courses / lectures	Projektseminar: Sensory Sciences Lab (for MSc Chemistry und MSc Molecular Science) (2 SWS) Please note: • The module "Sensory Sciences Lab" is an interdisc module with a limited number of participants: only 1 from MSc Chemistry or MSc Molecular Science will admitted! Students from Molecular LIFE Science will priority! • The module "Sensory Sciences Lab" can be taken at the compulsory elective module "Advances in Bio-C Bio-Inorganic Chemistry" (20 ECTS in total, all module graded) or as an elective module (5 ECTS, ungr	O students be II be given as part of Organic and ules have to
3	Lecturers	Prof. Dr. Danijela Gregurec Prof. Dr. Tim Rohe Prof. Dr. Jessica Freiherr Dr. Helene Loos	

4	Module coordinator	Prof. Dr. Andrea Büttner	
5	Contents	The field of Sensory Sciences investigates how animals and humans sense, neurally process and perceive their environment. This research-oriented interdisciplinary course provides important concepts, theories, and methods of the Sensory Sciences to advanced students from three disciplines (Psychology, Medicine and Chemistry - Molecular Science). These topics comprise human auditory, visual and olfactory perception, chemocommunication, psychophysics, neuroimaging, molecular sensory receptors and technologies for neuromodulation (among others). The students apply these concepts in interdisciplinary teams to develop, conduct and analyse their own small research project. While engaging in an interdisciplinary collaboration and exchange, students can share and deepen their discipline-specific perspectives and competencies as well as learn about concepts and methods from other disciplines.	
6	Learning objectives and skills	The students are able to • explain fundamental concepts and methods of the Sensory Sciences develop an own (limited) research question and design an empirical data collection to investigate the research question • collect data in the laboratory and analyse the data present their findings in oral and written form • communicate effectively in interdisciplinary teamsunderstand and reflect the challenges and chances of interdisciplinary collaborations.	
7	Prerequisites	None	
8	Integration in curriculum	semester: 1;2;3	

9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202	
10	Method of examination	Seminar achievement Written project report (20-30 pages)	
11	Grading procedure	Seminar achievement (100%) 100% project report • Part of the compulsory elective module "Advances in Bio-Organic and Bio-Inorganic Chemistry" in M.Sc. Molecular Science (5 ECTS, graded; PO 2020) • elective module (5 ECTS, ungraded; PO 2020)	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 30 h Independent study: 120 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Literature will be announced in the first session Scripts and materials will be provided via StudOn! 	

1	Module name 46523	Small Molecule Activation	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Karsten Meyer	

4	Module coordinator	Prof. Dr. Sjoerd Harder	
5	Contents	 Lecture: Inorganic coordination chemistry, catalytic and electrocatalytic transformation of abundant small molecules N2, O2, CO or CO2, H2O and NH3, into value-added commodities, fine-chemicals, ac-tive pharmaceutical ingredients and polymers. Selective oxidation reactions. CO2 reduction to CO and subsequent Fischer-Tropsch catalysis for the production of carbon-based fuels. Reductive activation of atmospheric N2 to NH3 and the production of H2 from H2O splitting catalysis. Applications, e.g. fuel-cell technologies for carbon-free energy production. 	
6	Learning objectives and skills	The student	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Homogenous Catalysis Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory Elective Module (20 ECTS in total!) • Lecture module as part of the Elective Module (5 ECTS, not graded)	
10	Method of examination	Oral O20(PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46543	Solar Energy Conversion	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Solar Energy Conversion (3 SWS)	
3	Lecturers	Prof. Dr. Dirk Michael Guldi	

4	Module coordinator	Prof. Dr. Dirk Michael Guldi	
5	Contents	 Demand and supply of energy Solar cells: Silicon solar cells, dye-sensitized solar cells, organic solar cells, perovskite solar cells, singlet fission Fundamentals of Electron Transfer Photosynthesis: natural photosynthesis, artificial photosynthesis 	
6	Learning objectives and skills	 The students are familiar with the fundamentals and modern applications in solar energy research and applications understand design principles in solar energy devices and can transfer this knowledge to related topics can present, communicate and discuss scientific results with experts in English. 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Energy Materials Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module "Advances in Energy Materials" MSc Chemistry and Molecular Science (20 ECTS in total!) • Module can be taken as part of the Elective Module, too (5 ECTS, not graded)!	
10	Method of examination	Written examination (60 minutes)	
11	Grading procedure	Written examination (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		

1	Module name 46512	Special Aspects in Bio-Organic Chemistry	5 ECTS
		Vorlesung: Special Aspects in Bio-Organic Chemistry (2 SWS)	5 ECTS
2	Courses / lectures	Übung: Special Aspects in Bio-Organic Chemistry: Seminar (1 SWS)	
		Please note: • Seminar: time and place by agreement!	
3	Lecturers	Prof. Dr. Petra Imhof Prof. Dr. Andriy Mokhir	

4	Module coordinator	Prof. Dr. Nicolai Burzlaff	
5	Contents	The students • learn about: nucleic acid structure, synthesis, and reactivity; nucleic acid modifications; nucleic acid-protein interactions, nucleic acid-based drugs, nucleic acid biochemistry, modelling of nucleic acids	
6	Learning objectives and skills	The students - can explain, apply and reflect upon the theories, terminology, specialities, boundaries and different school of bioorganic and bioinorganic chemistry critically and in depth	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Bio-Organic and Bio-Inorganic Chemistry Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Bioorganic and Bioinorganic Chemistry" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • As Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral O20 (PL): Oral examination (20 minutes)	
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail	
12	Module frequency	only in winter semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	A. Bloomfield, D.M. Crothers, I. Tinoco, Jr. "Nucleic Acids: structures, properties, and functions", University Science Books, Sausalito, CA, USA 1999	

- D. L. van Vranken, G. A. Weiss, Introduction to Bioorganic Chemistry and Chemical Biology, 1. Aufl., Garland Publishers 2012, ISBN: 978-0815342144
- W. Saenger, Principles of Nucleic Acid Structure", Springer-Verlag New York Inc. 1984
- T. Schlick: "Molecular modelling and simulation: An interdisciplinary guide", Springer New York Dordrecht Heidelberg London 2nd ed. 2010
- A. Vologodskii: "Biophysics of DNA", Cambridge University press, 2015, DOI 10.1017/CBO9781139542371

1	Module name 46532	Surface and Interface Science	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	 Vacuum and pressure measurement (pumps, pressure and flow ranges) Lab-based and synchrotron-based light sources (principles, optics, insertion devices, etc.) Theory of photoemission and electronic structure XPS (elemental / chemical sensitivity, cross sections, quantification, examples) UPS (gas phase, adsorbates, 2D band structures, 3D band structures, orbital tomography) IPES (probing of unoccupied states, energy level alignment determination) HAXPES (depth-resolved photoemission measurements, examples) PEEM (spatially-resolved photoemission measurements, examples) NEXAFS (principle and examples) XES & RIXS (principle and examples) Structure of surfaces/ diffraction at surfaces (LEED, definitions and examples) X-ray spectroscopy based materials research on energy conversion devices (examples from current research) 	
6	Learning objectives and skills	 Students understand the principles of photoemission variants and their applications can judge the quality of data evaluation and its pitfalls can deliberately select an X-ray spectroscopic analysis method to address given scientific question and are able to evaluate the collected data 	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Lecture module within the Compulsory elective module "Advances in Interfaces and Catalysis" in M.Sc. Chemistry or M. Sc. Molecular Science (20 ECTS in total) • Lecture module as part of the Elective Module in M.Sc. Chemistry or M. Sc. Molecular Science (5 ECTS, not graded)	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (100%)	

		As part of the elective module: 0% - pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography	 Hüfner: Photoelectron Spectroscopy: Principles and Applications" Springer-Verlag Berlin Heidelberg Ertl, Küppers: Low Energy Electrons and Surface Chemistry" VCH Weinheim D. Attwood, Soft X-rays and Extreme Ultraviolett Radiation, Cambridge University Press, 1999 D. Briggs, M.P. Seah: PracticalSurface Analysis: AugerandX-Ray PhotoelectronSpectroscopy, Wiley, 1996 M. Bär, L. Weinhardt, and C. Heske: Advanced Characterization Techniques for Thin Film Solar Cells, edited by D. Abou-Ras, T. Kirchartz, and U. Rau (Wiley VCH Verlag GmbH & Co KGaA, ISBN: 978-3-527-33992-1), 2nd Extended Edition, Volume 2, Chap. 18. A. Meisel, G. Leonhardt, R. Szargan: X-ray Spectra and Chemical Binding, Springer, 1989 	

1	Module name 46546	Symmetry and Group Theory	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Symmetry and Group Theory - Applications in Chemistry, Physics and Material Sciences (3 SWS)	
3	Lecturers	Prof. Dr. Jörg Libuda	

4	Module coordinator	Prof. Dr. Jörg Libuda	
5	Contents	 Symmetry of Molecules (symmetry elements, operations, point groups, notations) Symmetry of Crystals, Surfaces and Interfaces (symmetry in 1, 2 and 3 dimensional periodic strutures, lattices, crystal classes, space groups) Compact Course Group Theory (elements group theory, definitions, reducible and irreducible representations, orthogonality theorem, character tables) Group Theory and Quantum Mechanics (representations, operators and symmetry, matrix elements, direct product functions, projection operators) Symmetry of Organic Molecules: From Electronic Structure to Reactivity (symmetry adaption, cyclic groups, many electron systems, electronic transitions, configuration interaction, symmetry controlled reactions Symmetry in Anorganic Chemistry: From Atoms to Complexes (MO models, transition metal complexes, direct product groups, rotation inversion group, angular momentum coupling, crystal field splitting, vibronically allowed transitions) Symmetry and Spectroscopy: Vibrational Spectroscopies (analysis of vibrational modes, normal coordinate analysis, symmetry of vibrational wave functions, vibrational spectroscopy, selection rules) Symmetry in Crystal Physics: Tensor Description of Physical Properties (tensors, axial, polar, representations, transformation properties, intrinsic symmetry, Neumann's principle, Curie's principle) Symmetry and Electronic Structure of Solids: Band Structures (translation group and irreps, reciprocal lattice, k-space, Bloch functions, Brillouin zones, symmetry of bands) 	
6	Learning objectives and skills	acquire detailed understanding how to use symmetry properties and the mathematical tools of group theory in a broad range of application fields in chemistry, physics and materials science.	
7	Prerequisites	None	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202	

		Lecture module can be taken as part of the Elective module (5 ECTS, not graded)
10	Method of examination	Variable O20(PL): Oral Examination (20 minutes)
11	Grading procedure	Variable (0%) pass/fail
12	Module frequency	only in winter semester
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46531	Theory of Catalytic Processes	5 ECTS
2	Courses / lectures	Vorlesung mit Übung: Theory of Catalytic Processes (3 SWS)	5 ECTS
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Jörg Libuda
5	Contents	The module focuses on physical, chemical or technological aspects of modification, manipulation and characterization of interfaces. These aspects relate to the research of ideal model systems (surfaces and adsorbates on single crystal surfaces) or real systems, in which the interface plays a crucial role for the respective properties. In all cases, the local electronic and chemical interactions at the interface affect the geometric structure (e.g. adsorption geometry) and consequently the chemical and physical properties.
6	Learning objectives and skills	deepen their knowledge in experimental methods and theoretical aspects to describe and characterize interface phenomena are able to perform experiments independently and to analyse the data are familiar with the model-type description of experimental data
7	Prerequisites	None
8	Integration in curriculum	semester: 2;3
		Advances in Interface Research and Catalysis A Master of Science Molecular Science 20202 Elective modules Master of Science Molecular Science 20202 • Within the Compulsory Elective Module Advances in Interfaces and Catalysis A (IntCat-3A) (20 ECTS in total)! • Module can also be taken as part of the elective module (5 ECTS, not graded)!
9	Module compatibility	Please note:
		MSc Molecular NANO Science students have to attend the module "Advances in Interfaces and Catalysis A (IntCat-3A)" / MSc Chemistry students can choose between the module "Advances in Interfaces and Catalysis A (IntCat-3A)" and the module "Nanostructured Materials and Interfaces B (IntCat-3B)" of Prof. Bachmann!
10	Method of examination	Oral (20 minutes)
11	Grading procedure	Oral (100%) As part of the elective module: 0% - pass/fail
12	Module frequency	only in winter semester
13	Workload in clock hours	Contact hours: 45 h

		Independent study: 105 h
14	Module duration	1 semester
15	Teaching and examination language	english
16	Bibliography	

1	Module name 46557	Theory of Surface Phenomena	5 ECTS
2	Courses / lectures	The teaching units in the module are only offered in the summer semester.	
3	Lecturers	Prof. Dr. Bernd Meyer	

4	Module coordinator	Prof. Dr. Bernd Meyer	
5	Contents	 Brief introduction into quantum-chemical methods for surface science studies Introduction of basic nomenclature of how to describe the atomic and electronic structure of surfaces Basic concepts on how to understand the electronic properties of metal, semiconductor and insulator surfaces, such as surface states, dangling bonds, passivation, charge neutralization with respect to polar and nonpolar surfaces Thermodynamic analysis of the stability of surface structures; surface phase diagrams Methods for calculating STM and AFM data to support the analysis of experimental data from local probe measurements 	
6	Learning objectives and skills	The students are familiar with the most common theoretical and experimental techniques for surface science studies have a sound knowledge in basic principles governing surface structure and reactivity can perform first quantum chemical calculations on their own and interprete the data	
7	Prerequisites	Basic knowledge of quantum mechanics and quantum chemical calculations is strongly recommended	
8	Integration in curriculum	semester: 2;3	
9	Module compatibility	Elective modules Master of Science Molecular Science 20202 Elective module (5 ECTS, ungraded) within the M.Sc. degree programme Chemistry or M.Sc. Molecular Science (especially Molecular NANO science)	
10	Method of examination	Oral (20 minutes)	
11	Grading procedure	Oral (0%) pass/fail	
12	Module frequency	only in summer semester	
13	Workload in clock hours	Contact hours: 45 h Independent study: 105 h	
14	Module duration	1 semester	
15	Teaching and examination language	english	
16	Bibliography		