

Nanoprobes in Materials Science, Physics & Chemistry

Week 2

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Erlangen-Nürnberg

Physical Chemistry 2 (surface & interface science)



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■Introduction and Motivation

- Length scales, physical properties, quantization effects
- Nano-objects, "bottom-up" / "top-down" approach

■Optical methods

- Visible light microscopy (VLM)
- Confocal laser-scanning microscopy(CLSM)
- Scanning near-field optical microscopy (SNOM)

■Scanning probes

- Scanning-tunneling microscopy / -spectroscopy
- Scanning-force microscopy
- Other scanning probe techniques

■Electron probes

- Field-electron- / field-ion-microscopy (FEM / FIM)
- Electron microscopy (SEM, TEM, SAM, LEEM, PEEM)

■Ion probes

- He Ion Microscopy / FIB / AtomProbe Tomography

■X-ray techniques

- Basics (Synchrotron radiation, Photoelectron spectr., x-ray absorption, x-ray fluorescence)
- X-ray microscopy using soft and hard x-rays
- Small-angle x-ray scattering (SAXS) – compare to neutrons

Electron microscopy

EMITTED ELECTRONS:

- Field electron microscopy (FEM)
- Field ion microscopy (FIM)
- Photo electron emission microscopy (PEEM)

INCIDENT ELECTRONS:

- Scanning-electron microscopy (SEM, SAM, ...)
- Transmission electron microscopy (TEM)
- Low-energy electron microscopy (LEEM)

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Electron Microscopy

$$E = \hbar\omega, p = \hbar k = h/\lambda$$

$$E_k = \frac{1}{2}mv^2 = eU, |\vec{p}| = mv$$

$$\lambda_{vac} = \frac{h}{\sqrt{2meU}}, \lambda_{med} = \frac{h}{\sqrt{2m(eU - E_p)}}$$

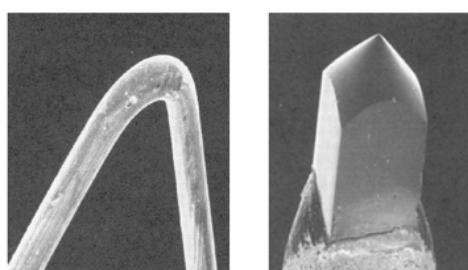
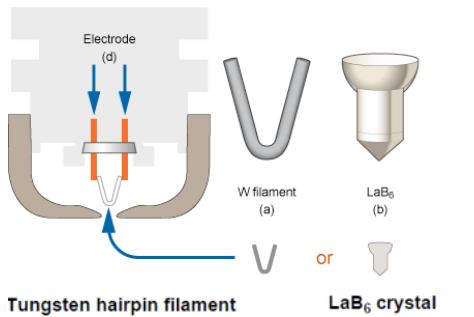
$$n = \frac{\lambda_{vac}}{\lambda_{med}} = \sqrt{1 - \frac{E_p}{eU}}$$

U/V	v/c	λ/nm
10^{-1}	$6.3 \cdot 10^{-4}$	3.9
1	$2.0 \cdot 10^{-3}$	1.2
10^1	$6.3 \cdot 10^{-3}$	$3.9 \cdot 10^{-1}$
10^2	$2.0 \cdot 10^{-2}$	$1.2 \cdot 10^{-1}$
10^4	0.19	$1.2 \cdot 10^{-2}$
10^6	0.94	$8.7 \cdot 10^{-4}$

Significantly shorter wavelength compared to visible light will offer better resolution !

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Creating electrons, e.g., by thermal emission



Thermal emission according to **Richardson law**:

$$\text{Emission current density } j = A T^2 \exp(-\Phi / k_B T)$$

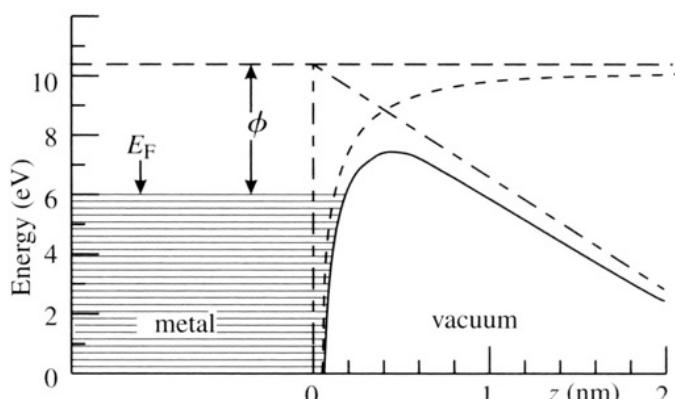
A is a material constant, e.g. AW = 120 A/(cm K)² Φ_W = 4.5 eV

Sharp tip spot-welded to the hairpin filament increases j by a factor of 4

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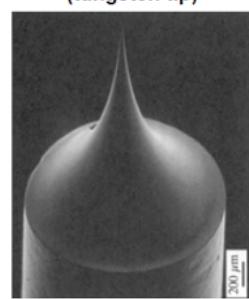
From: Williams/Carter
Transmission electron microscopy

Creating electrons, e.g., by cold field emission



$$J(T) = AE^2 \exp(-B\phi^{3/2}/E)$$

Field emission source
(tungsten tip)



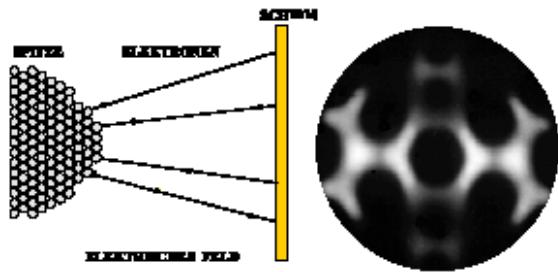
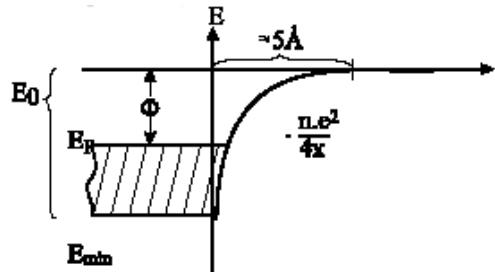
From: Williams/Carter
Transmission electron
microscopy

Fowler/Nordheim equation (1928) describes current density J emitted by a filament under a high electric field, where plot of log(J/E²) vs. 1/E gives straight line.

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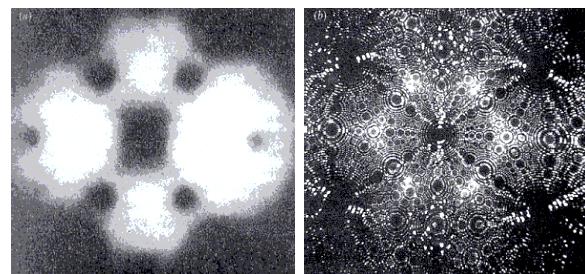
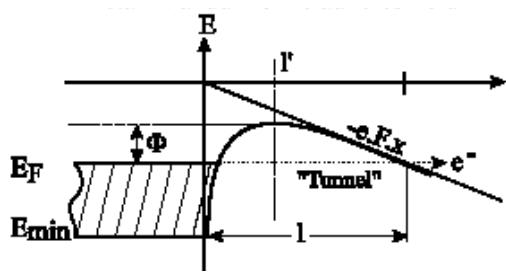
Field electron microscopy (FEM)

1D electrostatic potential at a metal surface without field



FEM scheme

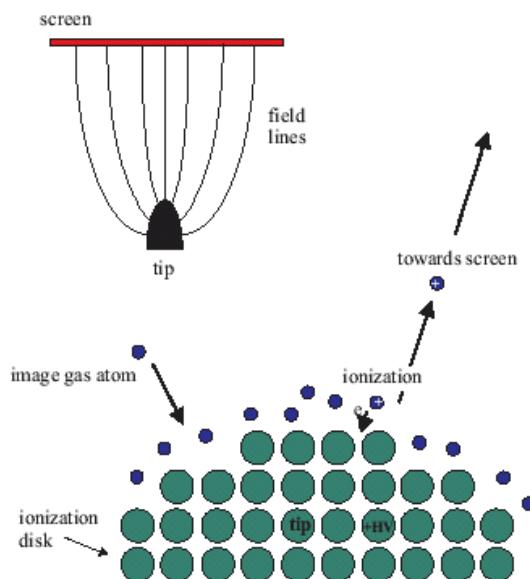
... with external electric field



$\Delta x = 2-3 \text{ nm}$

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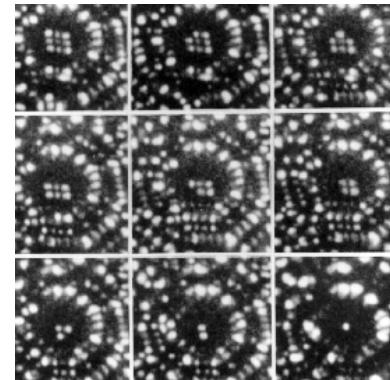
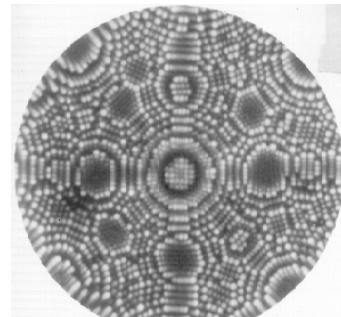
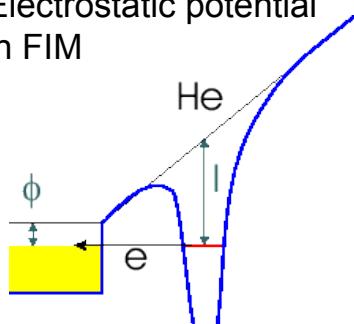
Field Ion Microscopy (FIM)



E. Müller, Univ. Erlangen, 1950's

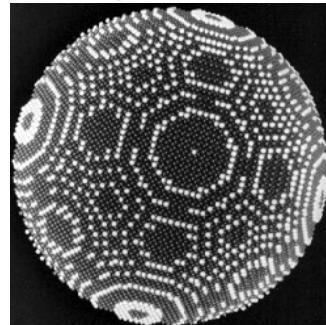
Field ion microscopy (FIM)

Electrostatic potential
in FIM



Experiment

Due to **the** higher mass of the gas atoms (compared to electrons) and thus smaller Brownian motion, the resolution in FIM is higher than in FEM



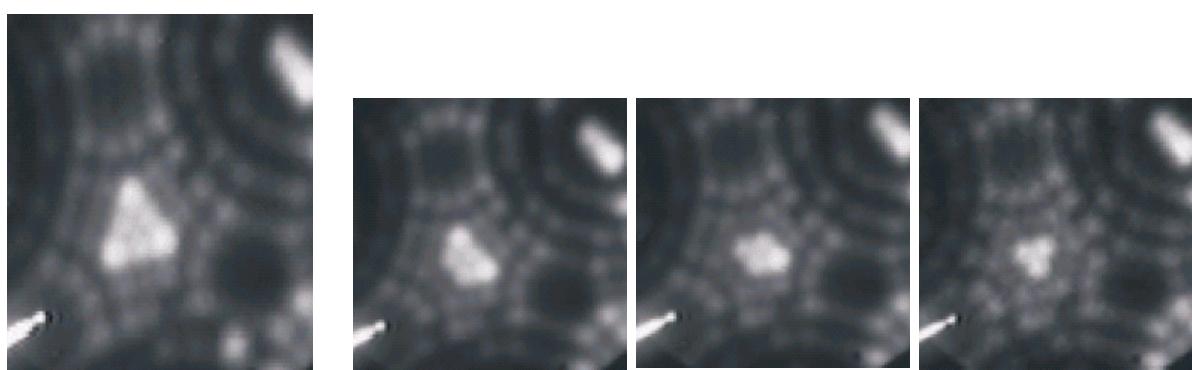
Removal of individual atoms from FIM tip

$$\Delta x = 0.1 \text{ nm}$$

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Simulation

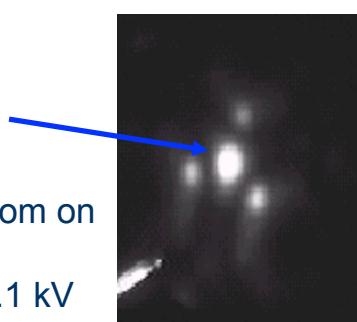
Atomic manipulations in FIM: W(111) tip



Imaging at 5.0 kV
Manipulating at 6.0 kV

In addition:
Field desorption
Field evaporation

Single Au atom on
W(111) tip
imaged at 2.1 kV



Electron microscopy

EMITTED ELECTRONS:

- Field electron microscopy (FEM)
- Field ion microscopy (FIM)
- Photo electron emission microscopy (PEEM)

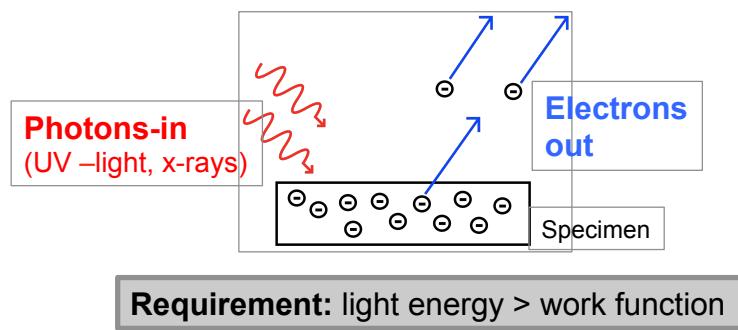
INCIDENT ELECTRONS:

- Scanning-electron microscopy (SEM, SAM)
- Transmission electron microscopy (TEM)
- Low-energy electron microscopy (LEEM)

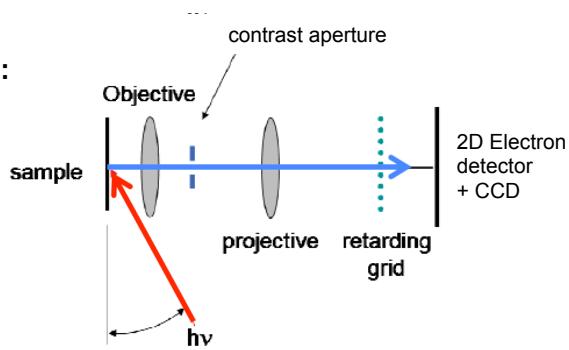
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PEEM = Photo Electron Emission Microscopy

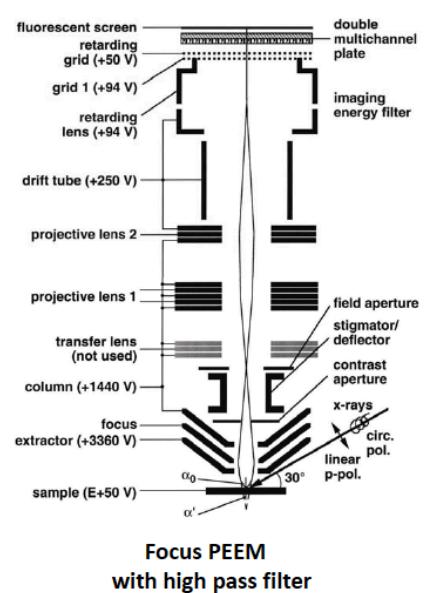
Concept: Photoelectric Effect (Einstein, 1905)



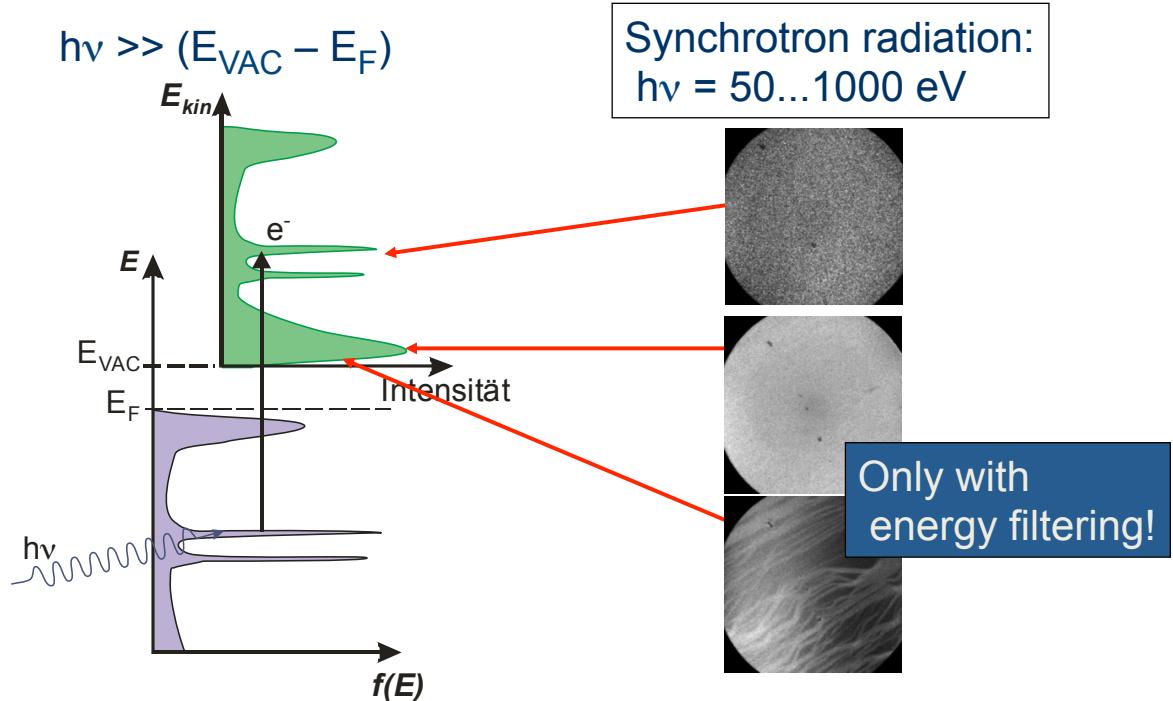
Setup:



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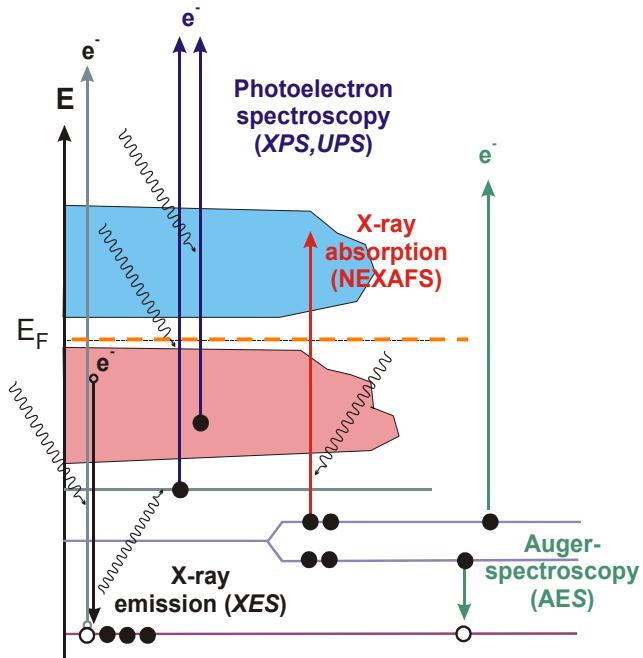
Principle of PEEM/XPEEM



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Appendix: Core-level electron spectroscopies

Example: p-type semiconductor DOS



Energy dispersive **electron detection** (i.e. inelastic mean free path is limited)
→ **surface sensitive**
(PES, NEXAFS, AES)

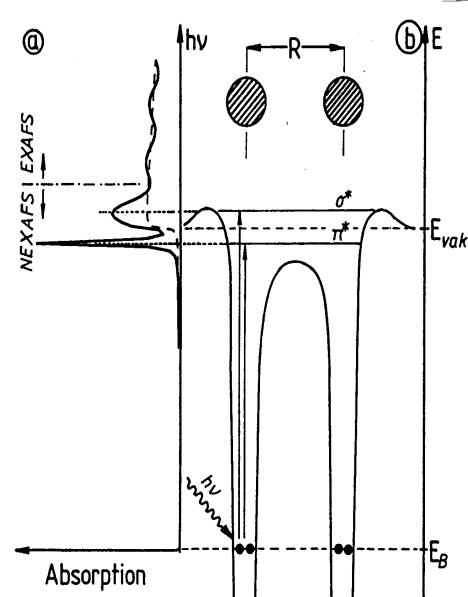
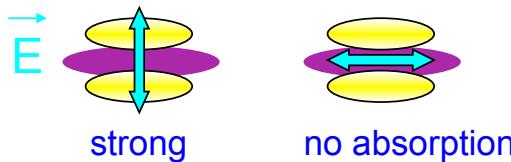
Energy dispersive
photon detection (i.e., higher penetration depth)
→ **bulk sensitive**
(XES)

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NEXAFS = Near-Edge X-Ray Absorption Fine Structure

core excitation into unoccupied molecular orbitals (π^* -resonances) [LUMO]
and into delocalized states (σ^* -resonances)
local chemical probe
→ *chemical bonds*
polarization dependence (*dichroism*)
→ molecular orientation

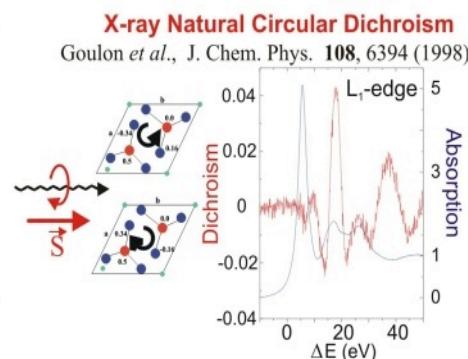
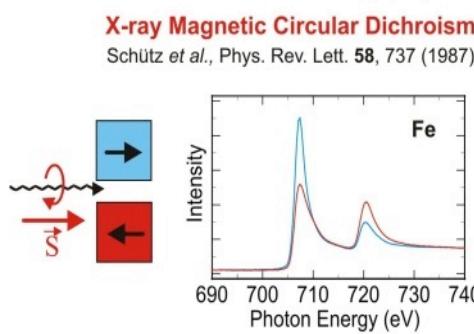
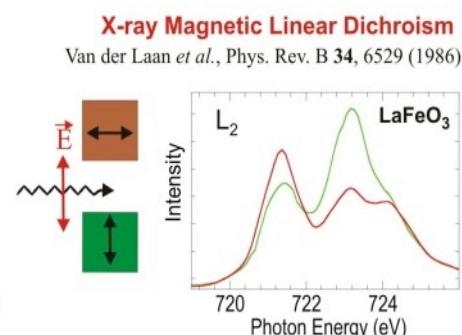
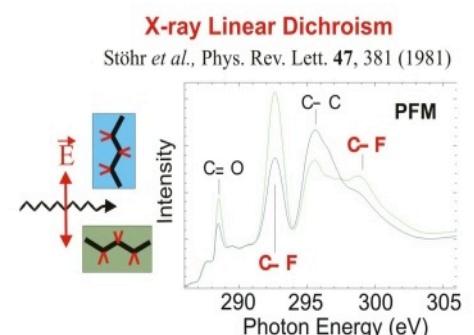
$$W_{if} \propto | \langle f | \mathbf{p} \cdot \mathbf{e} | i \rangle |^2 \cdot \rho(f)$$



many distinct resonances → model calculations

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Polarization effects in NEXAFS

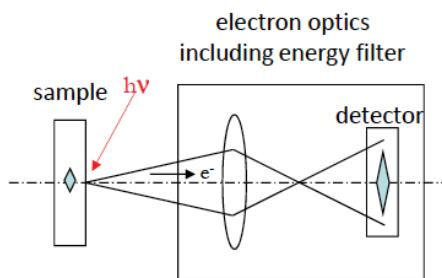


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Taken from: J. Stöhr, Stanford

Method characteristics (PEEM / XPEEM)

X-ray photo emission electron microscopy (XPEEM)



- ❖ Direct imaging, parallel detection
- ❖ Dynamic processes ok!
- ❖ Lateral resolution is determined by electron optics (10-50 nm); with aberration correction: few nm will be possible.
- ❖ Requires smooth sample morphology.
- ❖ Combination with LEEM/LEED
- ❖ Spectroscopic PEEM! Intermediate spectroscopic ability(200 meV).
- ❖ Diffraction imaging possible.
- ❖ Sensitive in plane magnetisation!
- ❖ Vacuum better than $1 \cdot 10^{-5}$ mbar

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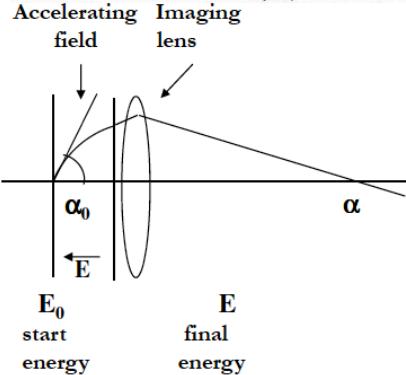
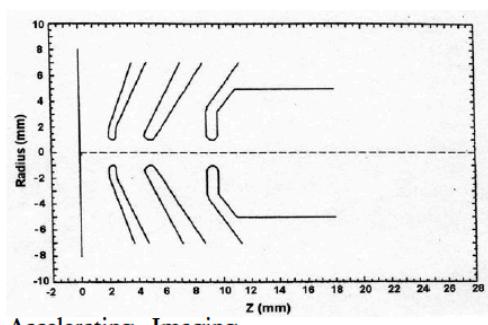
Cathode lens in PEEM (LEEM)

1. In emission microscopy θ (emission angle) is large. Electron lenses can accept only small θ because of large chromatic and spherical aberrations
2. Solution of problem: accelerate electrons to high energy before lens \rightarrow Immersion objective lens or cathode lens

$$\begin{aligned} n \sin \theta &= \text{const} \\ n &\sim \sqrt{E} \\ \theta &\rightarrow \alpha \\ \sin \alpha / \sin \alpha_0 &= \sqrt{E_0/E} \end{aligned}$$

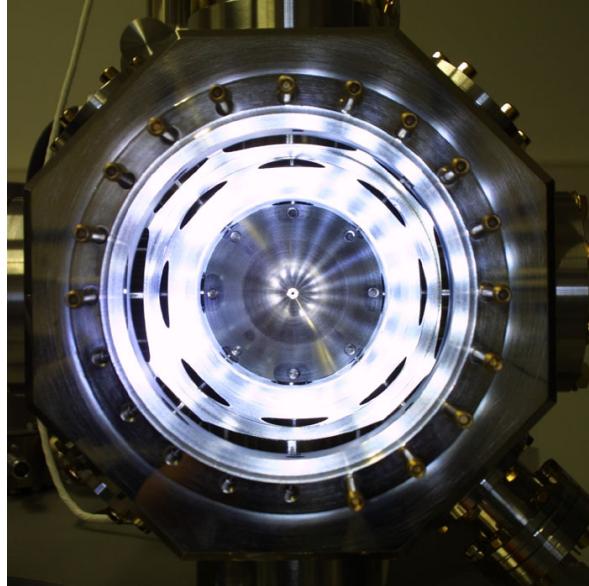
Example for $E = 20000$ eV:

E_0	2 eV	200 eV
α for $\alpha_0 = 45^\circ$	0.4°	4.5°



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Magnetic Lenses



Objective Lens of new S-PEEM in Essen

- **As Lenses for Optics**

- Focal Length

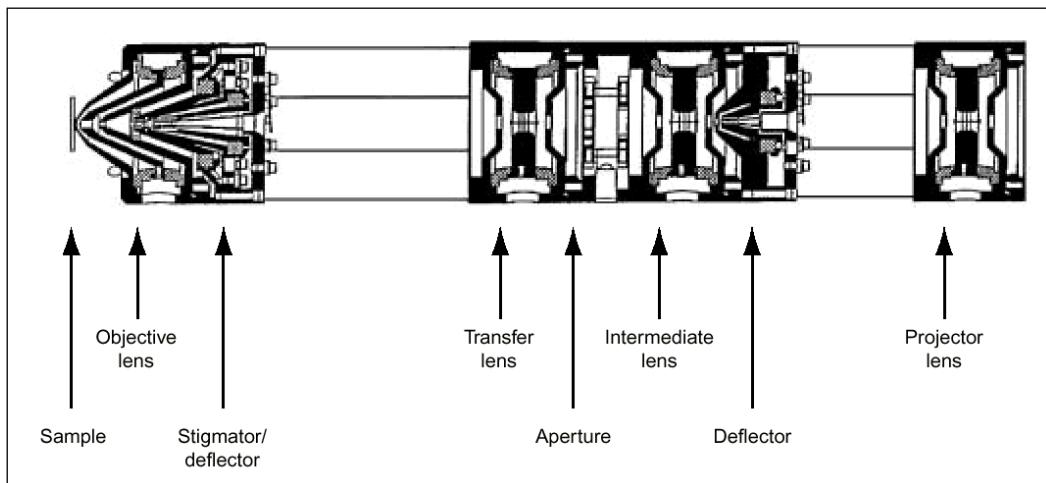
$$\frac{1}{f} = \frac{e}{8m_e U} \int_{-\infty}^{\infty} B_z^2 dz$$

- **A Few Differences**

- Image Rotation
- Only Convex Lenses

*Courtesy of F. Meyer-zu-Heringsdorf,
Univ. Essen-Duisburg
www.leem-user.de*

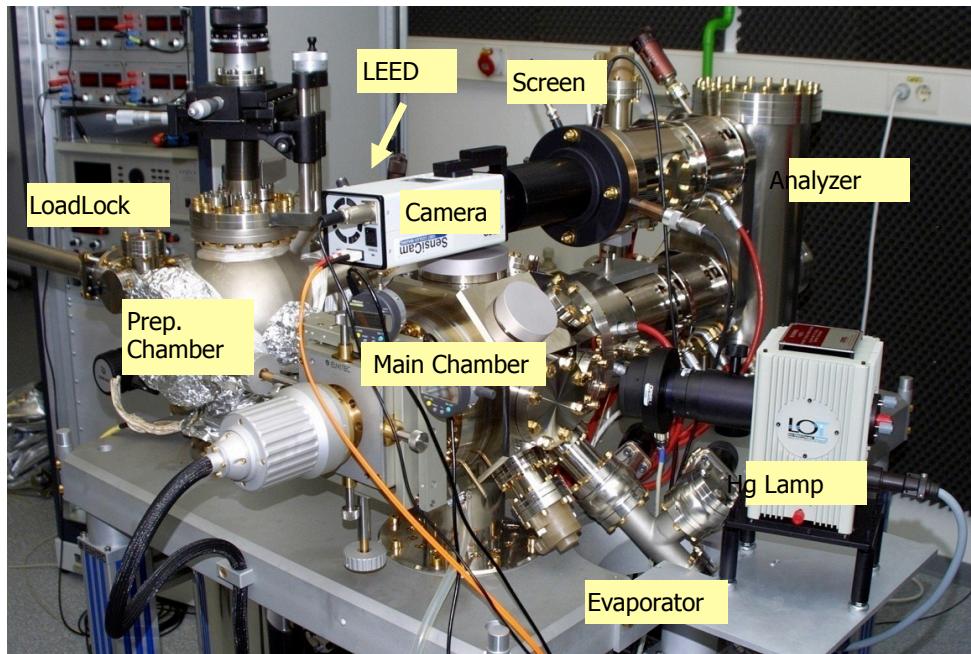
PEEM2, ALS Berkeley, USA (J. Stöhr, S. Anders et al)



Only electrostatic lenses !

from the endstation datasheet BL 7.3.1., ALS, Berkeley, USA, J. Stöhr et al

Spectroscopic PEEM (ELMITEC PEEM-III)

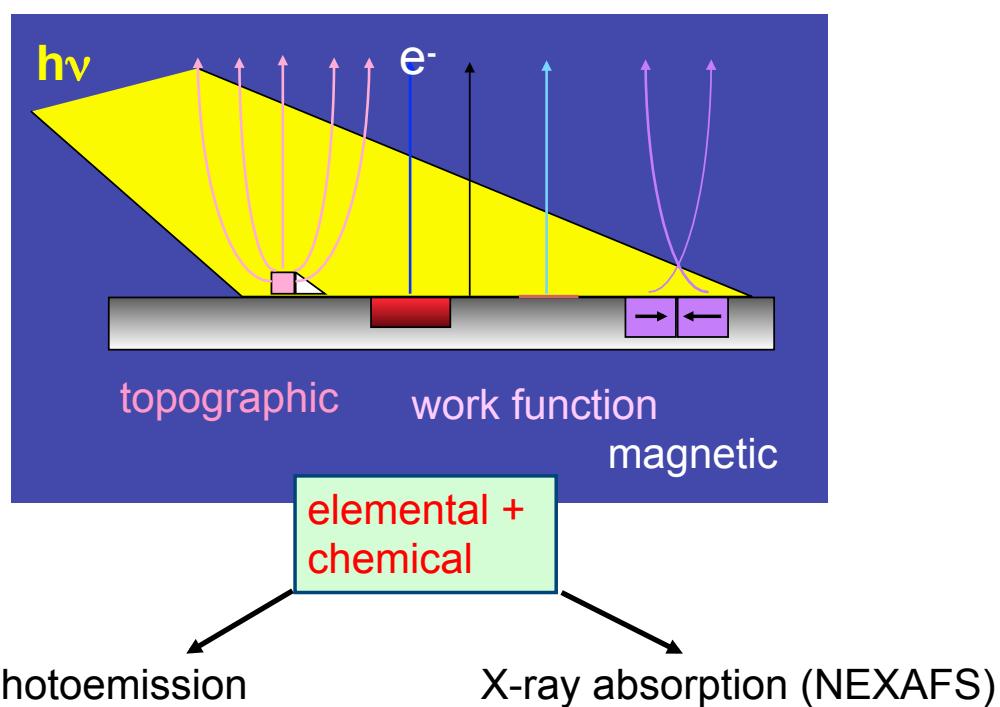


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- Energy Analyzer
- Preparation Chamber
- LN2 Sample Cooling
- LEED, Auger

<http://www.physik.uni-essen.de/SFB616>

PEEM contrast mechanisms



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Sample requirements for XPEEM

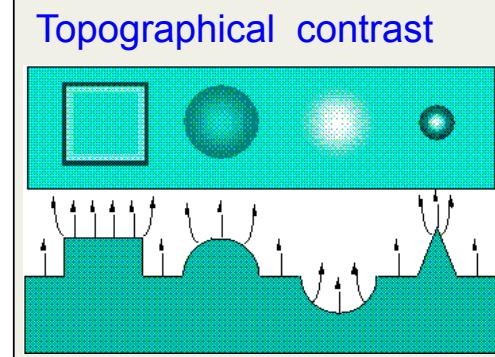
conducting samples (avoids charging !)

conducting substrates

surface roughness crucial

sample size > 8 mm diameter

sample in ultrahigh-vacuum

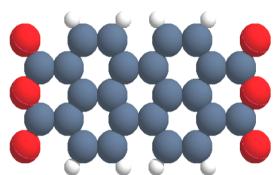


CRUCIAL: extreme high photon flux density → sample degradation

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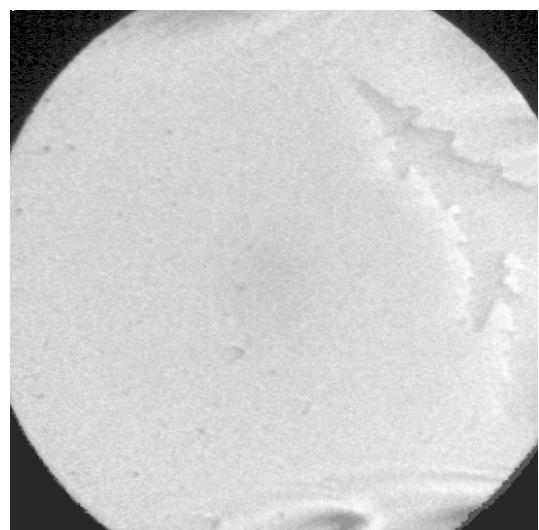
ORGANIC EPITAXY : layer-by-layer growth (1)

PTCDA/Ag(111)

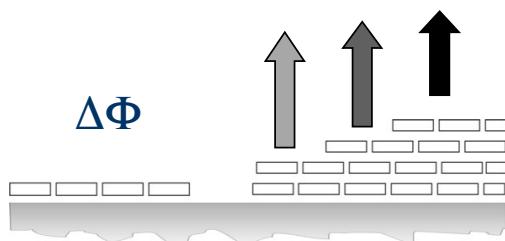


Deposition at **420 K (0.2 ML/min.)**

UV-PEEM: field of view 27 µm



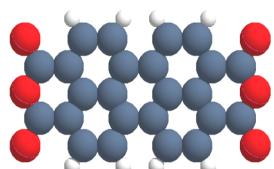
PEEM-contrast



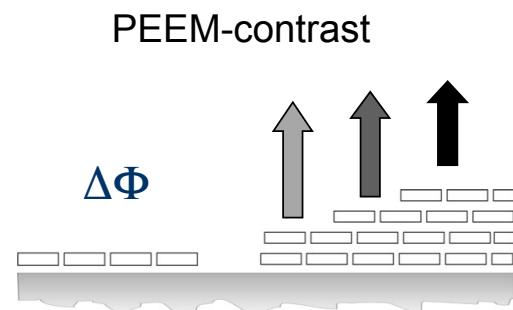
Exp.: Th.Schmidt, H. Marchetto, R.F.

ORGANIC EPITAXY : layer-by-layer growth (2)

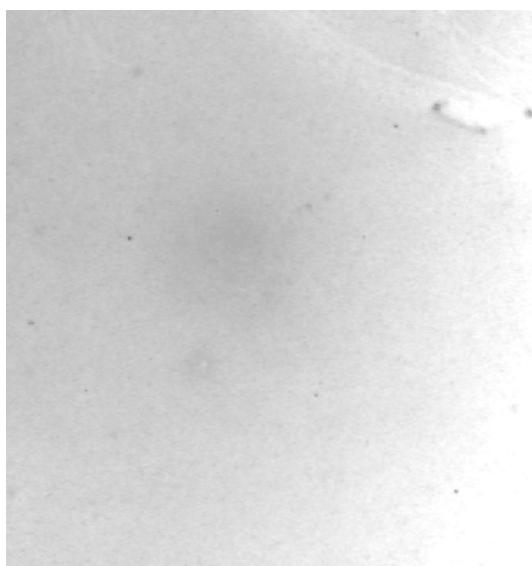
PTCDA/Ag(111)



Deposition at **room temperature**
(0.2 ML/min.)

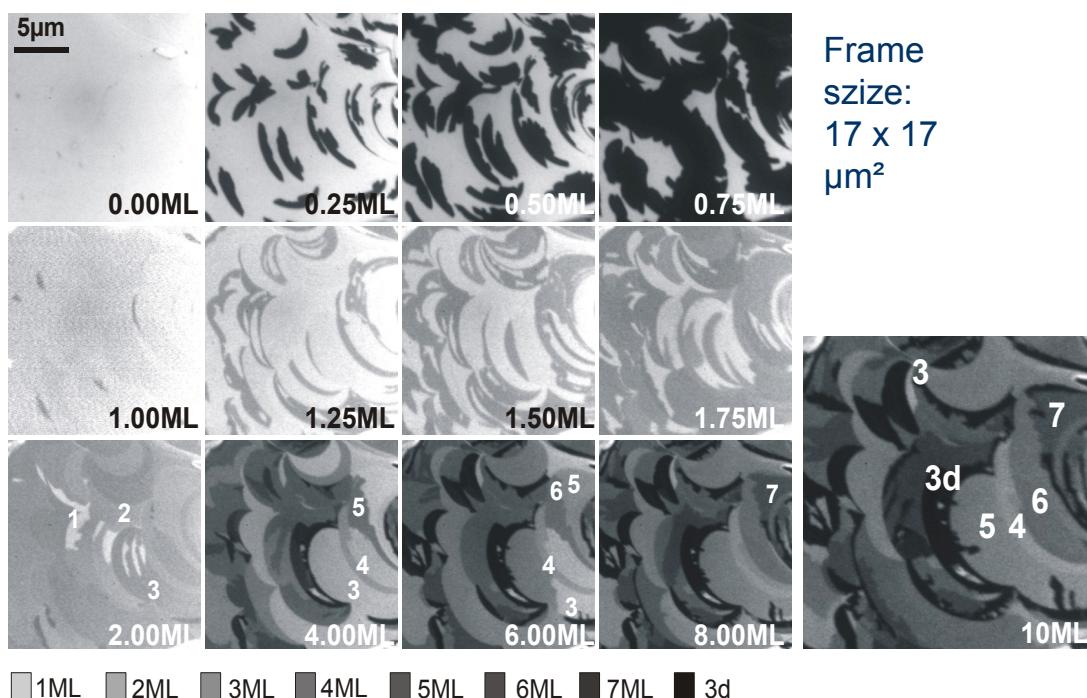


UV-PEEM: field of view 27 μm

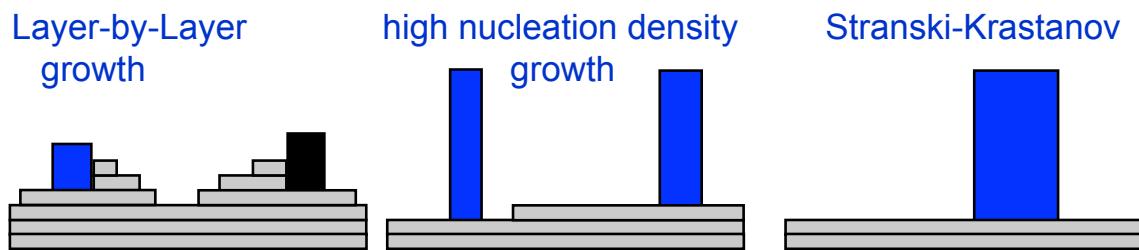
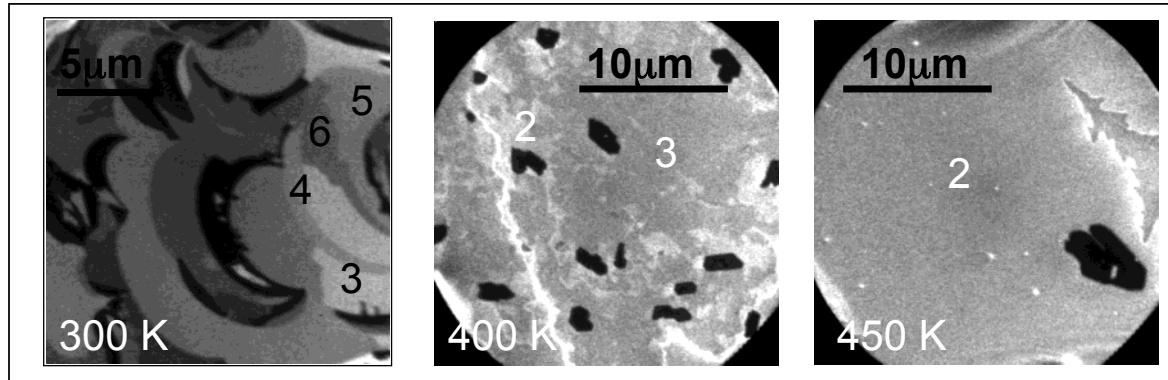


Exp.: Th.Schmidt, H. Marchetto, R.F.

PTCDA - epitaxial growth - PEEM

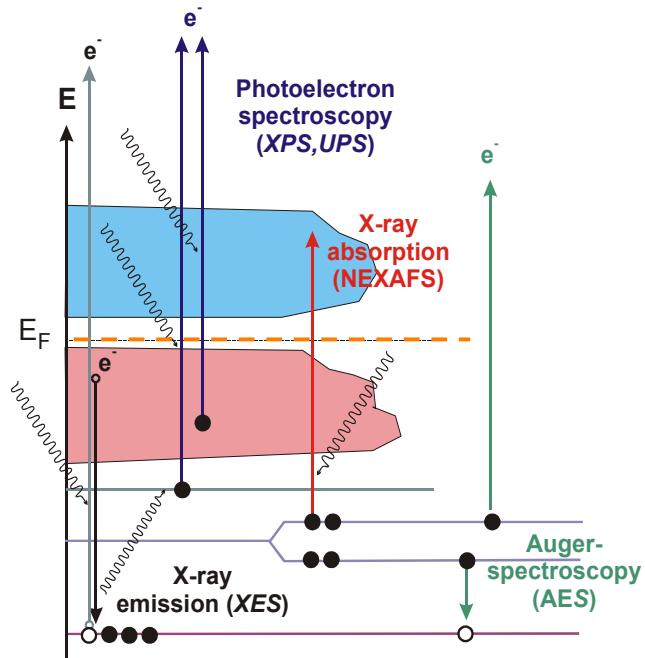


Temperature dependence of growth



Appendix: Core-level electron spectroscopies

Example: p-type semiconductor DOS



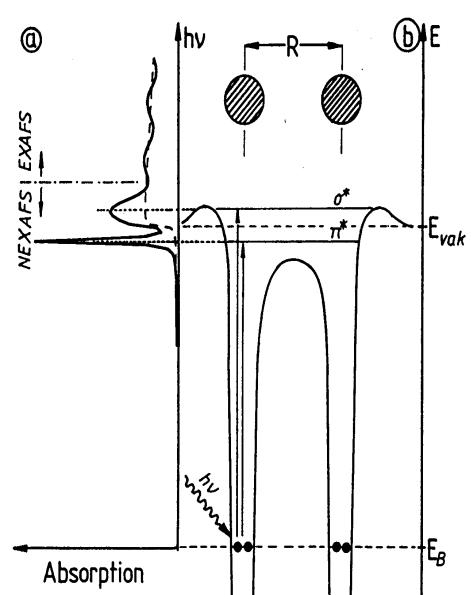
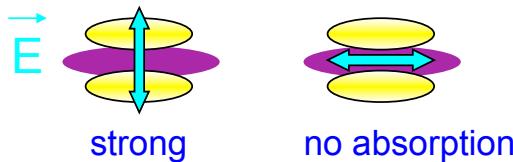
Energy dispersive **electron detection** (i.e. inelastic mean free path is limited)
→ surface sensitive
(PES, NEXAFS, AES)

Energy dispersive
photon detection (i.e., higher
penetration depth)
→ bulk sensitive
(XES)

NEXAFS = Near-Edge X-Ray Absorption Fine Structure

core excitation into unoccupied molecular orbitals (π^* -resonances) [LUMO] and into delocalized states (σ^* -resonances)
 local chemical probe
 → *chemical bonds*
 polarization dependence (*dichroism*)
 → molecular orientation

$$W_{if} \propto | \langle f | \mathbf{p} \cdot \mathbf{e} | i \rangle |^2 \cdot \rho(f)$$



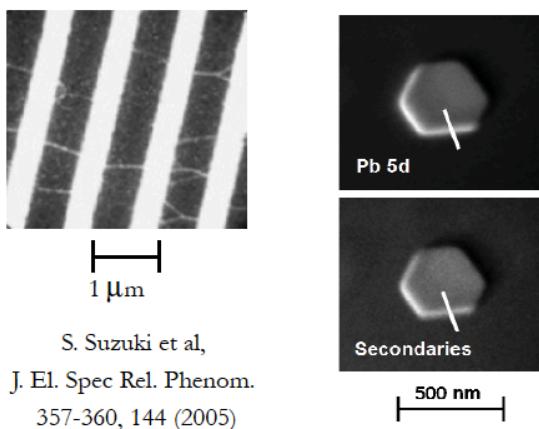
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many distinct resonances → model calculations

Properties accessible in XPEEM

ELEMENTAL COMPOSITION & CHEMICAL STATE

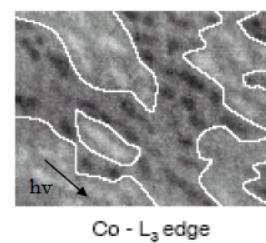
C1s image of SWCN Pb on W110



S. Suzuki et al,
J. El. Spec Rel. Phenom.
357-360, 144 (2005)

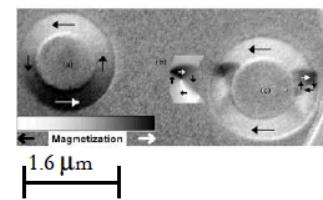
MAGNETIC STATE using XMCD

Co nanodots on Si-Ge



A. Mulders et al,
Phys. Rev. B 71,
214422 (2005).

patterned structures

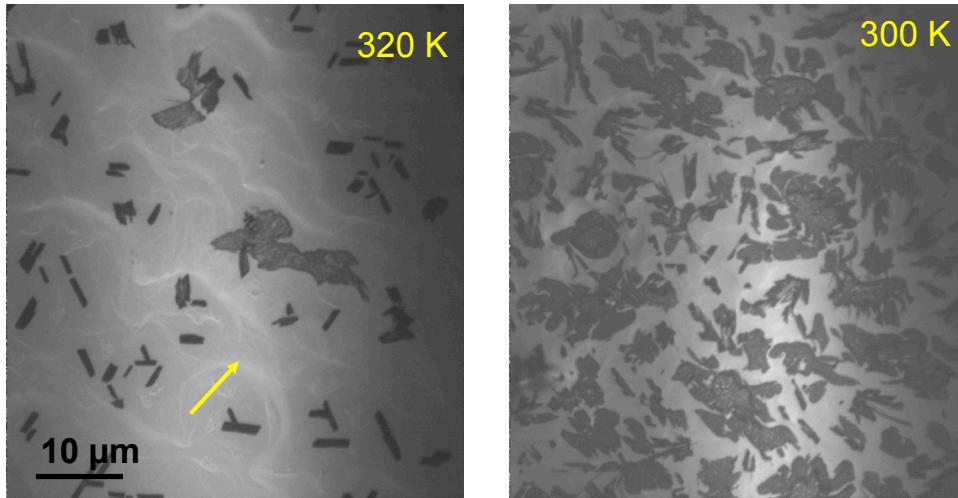


M. Klaeui et al,
Phys. Rev. B 68,
134426 (2003).

Taken from A. Locatelli (ELETTRA)

Film morphology - XPEEM

NTCDA/Ag(111):



Stranski-Krastanov growth mode (layer-plus-island)

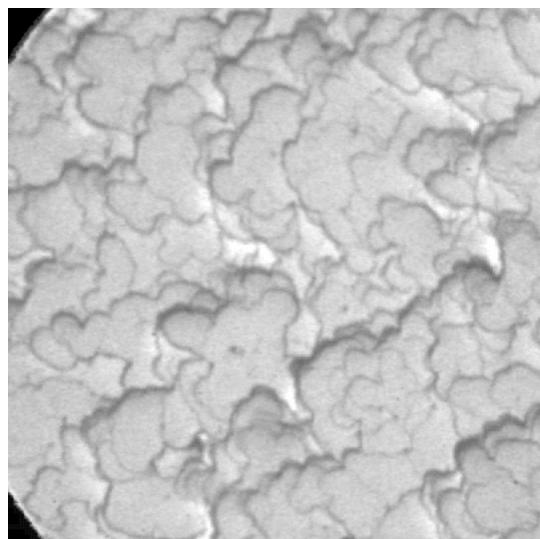
320 K: NTCDA crystallites (upright standing molecules)

300 K: dendritic growth, different island morphology

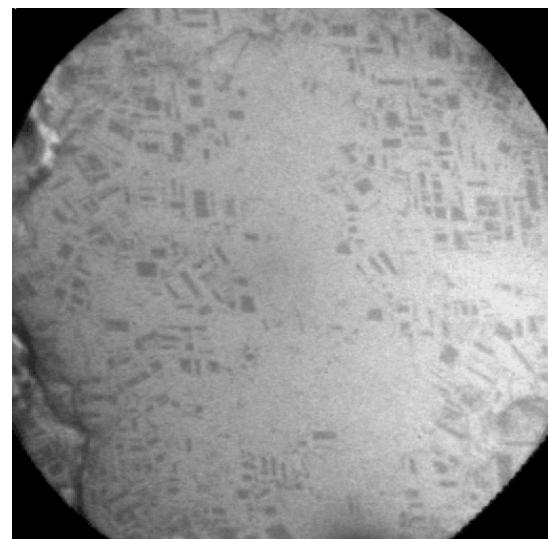
³⁷²

NTCDA/Ag(111) - in-situ growth (UV-PEEM)

Adsorption



Desorption (thermal annealing)



Photoelectron emission contrast partly
due to molecular reorientation:

modification of the surface dipole

³⁷³

Experiments: Thomas Schmidt

Biominerals in parrotfish teeth

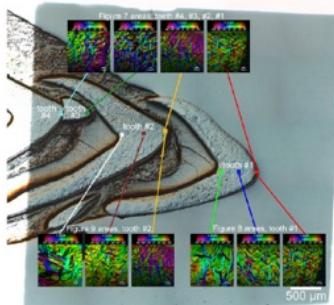
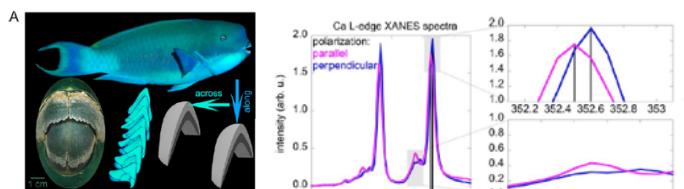


Figure 6. VLM micrograph of embedded and polished parrotfish teeth. The colored squares the 60 $\mu\text{m} \times 60 \mu\text{m}$ areas where the PIC maps of Figures 7–9 were acquired.

Nanocrystal orientation from polarization dependent imaging contrast (PIC, linear abs. dichroism)

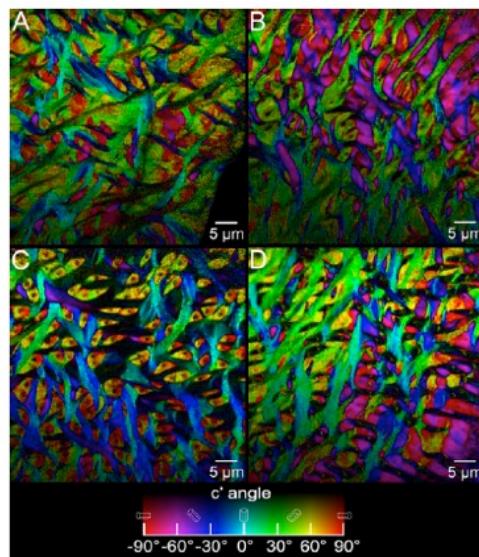
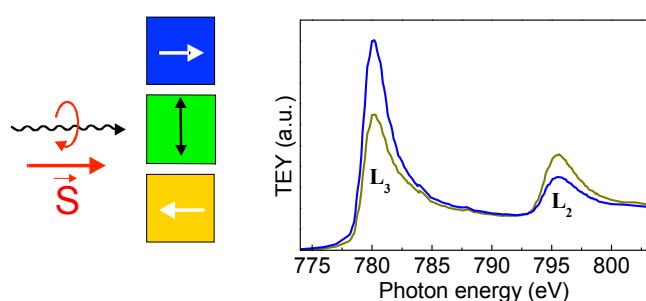


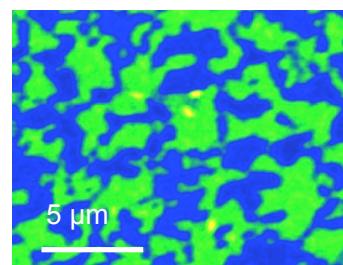
Figure 7. PIC maps showing different fluorapatite crystal sizes and orientations at the biting tips of teeth #1–4 as labeled in Figure 6. Two μm -wide fibers, made of 100 nm wide FAP crystals, are interwoven. The crystal orientations, shown schematically by prisms in the color bar, are as seen in projection perpendicular to the X-ray beam, which comes in from the right, at 30° from the surface.

M.A. Marcus et al., ACS Nano 2017, 11, 11856

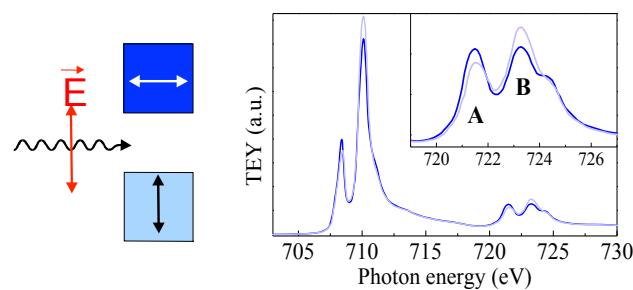
XMCD (X-ray Magnetic Circular Dichroism)



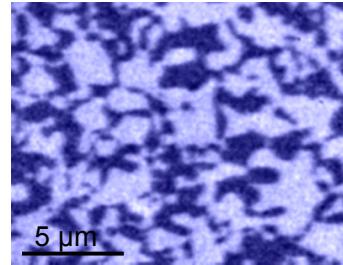
Co



XMLD (X-ray Magnetic Linear Dichroism)



LaFeO₃

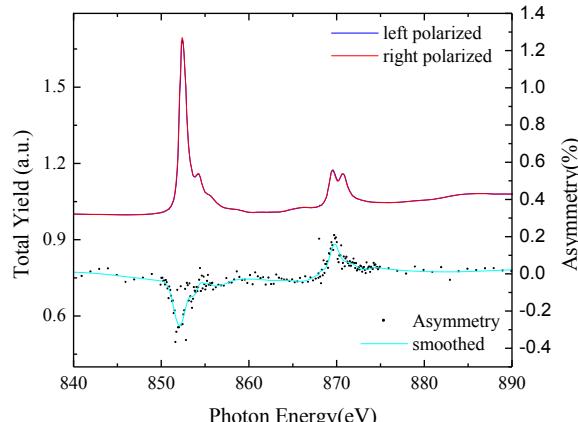


Exchange biased Co/NiO multilayer



XMCD
Left and right
circularly
polarized
light
 30°

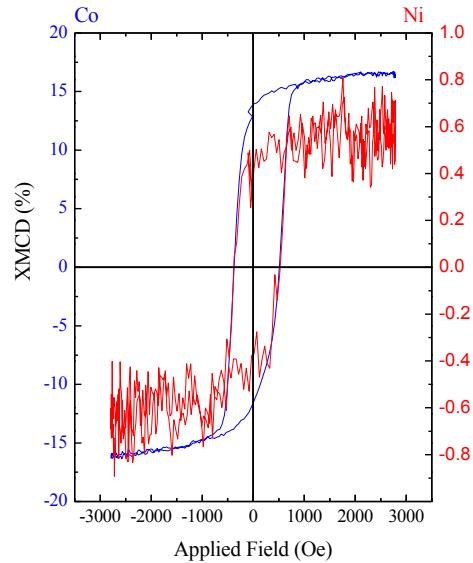
XMCD spectra – switching polarization



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EPU beamline 4 at ALS

Hysteresis of **Co** and **NiO**



H. Ohldag et al., Phys. Rev. Lett. 91(1), 017203/1-4 (2003).

(X)PEEM: Good points, bad points

- **Surface inspection**
Chemical composition info available
- **Large areas accessible**
- **In-situ observations**
- **Informative 3d images**
- **Aberration correction already available ! (later)**

- **Only used on conducting samples**
- **Sample in vacuum**
- **Edge effects cause image distortions**
- **Possibility of beam damage due to high photon intensities**